

FOR 10559534 by Cynthia Hamilton

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NEWS 1 OCT 02 Web Page for STN Seminar Schedule - N. America
NEWS 2 OCT 02 CA/Cplus enhanced with pre-1907 records from *Chemisches Zentralblatt*
NEWS 3 OCT 19 BEILSTEIN updated with new compounds
NEWS 4 NOV 15 Derwent Indian patent publication number format enhanced
NEWS 5 NOV 19 WPIX enhanced with XML display format
NEWS 6 NOV 30 ICSD reloaded with enhancements
NEWS 7 DEC 04 LINPACDOCDB now available on STN
NEWS 8 DEC 14 BEILSTEIN pricing structure to change
NEWS 9 DEC 17 USPATOLD added to additional database clusters
NEWS 10 DEC 17 IMSDRUGCONF removed from database clusters and STN
NEWS 11 DEC 17 DGENE now includes more than 10 million sequences
NEWS 12 DEC 17 TOXCENTER enhanced with 2008 MeSH vocabulary in MEDLINE segment
NEWS 13 DEC 17 MEDLINE and LMEDLINE updated with 2008 MeSH vocabulary
NEWS 14 DEC 17 CA/Cplus enhanced with new custom IPC display formats
NEWS 15 DEC 17 STN Viewer enhanced with full-text patent content from USPATOLD
NEWS 16 JAN 02 STN pricing information for 2008 now available
NEWS 17 JAN 02 CAS patent coverage enhanced to include exemplified prophetic substances
NEWS 18 JAN 28 USPATFULL, USPAT2, and USPATOLD enhanced with new custom IPC display formats
NEWS 19 JAN 28 MARPAT searching enhanced
NEWS 20 JAN 28 USGENE now provides USPTO sequence data within 3 days of publication
NEWS 21 JAN 28 TOXCENTER enhanced with reloaded MEDLINE segment
NEWS 22 JAN 28 MEDLINE and LMEDLINE reloaded with enhancements
NEWS 23 FEB 08 STN Express, Version 8.3, now available
NEWS 24 FEB 20 PCI now available as a replacement to DPCI
NEWS 25 FEB 25 IFIRE reload with enhancements
NEWS 26 FEB 25 IMPRODUCT reload with enhancements
NEWS 27 FEB 29 WPINDEX/WPIIDS/WPIX enhanced with ECLA and current U.S. National Patent Classification

NEWS EXPRESS FEBRUARY 08 CURRENT WINDOWS VERSION IS V8.3,
AND CURRENT DISCOVER FILE IS DATED 20 FEBRUARY 2008

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FILE 'HOME' ENTERED AT 20:31:49 ON 14 MAR 2008

=> file reg
COST IN U.S. DOLLARS
SINCE FILE
ENTRY
SESSION
TOTAL
0.21 0.21

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Property values tagged with IC are from the ZIC/VINITI data file provided by InfoChem.

STRUCTURE FILE UPDATES: 13 MAR 2008 HIGHEST RN 1007849-81-1
DICTIONARY FILE UPDATES: 13 MAR 2008 HIGHEST RN 1007849-81-1

New CAS Information Use Policies, enter HELP USAGETERMS for details.

TSCA INFORMATION NOW CURRENT THROUGH January 9, 2008.

Please note that search-term pricing does apply when conducting SmartSELECT searches.

REGISTRY includes numerically searchable data for experimental and predicted properties as well as tags indicating availability of experimental property data in the original document. For information on property searching in REGISTRY, refer to:

<http://www.cas.org/support/stngen/stndoc/properties.html>

```
=> s dimethylhexane and dimethacrylate
      292 DIMETHYLHEXANE
      4926 DIMETHACRYLATE
L1          0 DIMETHYLHEXANE AND DIMETHACRYLATE

=> s dimethylhexane and diacrylate
      292 DIMETHYLHEXANE
      4106 DIACRYLATE
```

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L2 2 DIMETHYLHEXANE AND DIACRYLATE

=> d 1-2

L2 ANSWER 1 OF 2 REGISTRY COPYRIGHT 2008 ACS on STN

RN 220767-20-4 REGISTRY

ED Entered STN: 28 Mar 1999

CN 2-Propenoic acid, 1,1,4,4-tetramethyl-1,4-butanediyl ester, polymer with 1-(1,1-dimethylethoxy)-4-ethenylbenzene and 4-ethenylphenol (9CI) (CA INDEX NAME)

OTHER CA INDEX NAMES:

CN Benzene, 1-(1,1-dimethylethoxy)-4-ethenyl-, polymer with 4-ethenylphenol and 1,1,4,4-tetramethyl-1,4-butanediyl di-2-propenoate (9CI)

CN Phenol, 4-ethenyl-, polymer with 1-(1,1-dimethylethoxy)-4-ethenylbenzene and 1,1,4,4-tetramethyl-1,4-butanediyl di-2-propenoate (9CI)

OTHER NAMES:

CN 4-tert-Butoxystyrene-2,5-dimethyl-2,5-hexanediol diacrylate-4-hydroxystyrene copolymer

CN p-tert-Butoxystyrene-2,5-dimethyl-2,5-hexanediol diacrylate-p-hydroxystyrene copolymer

CN p-tert-Butoxystyrene-2,5-dimethylhexane-2,5-diacrylate-p-hydroxystyrene copolymer

CN p-tert-Butoxystyrene-p-hydroxystyrene-2,5-dimethyl-2,5-hexanediol diacrylate copolymer

MF (C14 H22 O4 . C12 H16 O . C8 H8 O)x

CI PMS

PCT Polyacrylic, Polystyrene

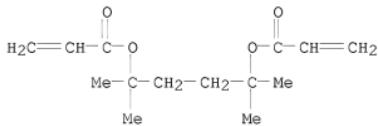
SR CA

LC STN Files: CA, CAPLUS, CHEMLIST, USPAT2, USPATFULL

CM 1

CRN 188837-15-2

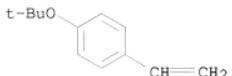
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CM 2

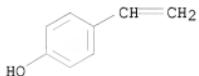
CRN 95418-58-9

CMF C12 H16 O



CM 3

CRN 2628-17-3
CMF C8 H8 O



PROPERTY DATA AVAILABLE IN THE 'PROP' FORMAT

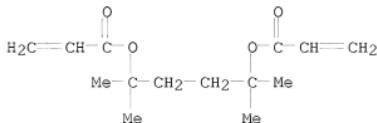
5 REFERENCES IN FILE CA (1907 TO DATE)
5 REFERENCES IN FILE CAPLUS (1907 TO DATE)

L2 ANSWER 2 OF 2 REGISTRY COPYRIGHT 2008 ACS on STN
RN 220767-18-0 REGISTRY
ED Entered STN: 28 Mar 1999
CN 2-Propenoic acid, 1,1,4,4-tetramethyl-1,4-butanediyl ester, polymer with 1,1-dimethylethyl 2-propenoate, ethenylbenzene and 4-ethenylphenol (9CI) (CA INDEX NAME)
OTHER CA INDEX NAMES:
CN 2-Propenoic acid, 1,1-dimethylethyl ester, polymer with ethenylbenzene, 4-ethenylphenol and 1,1,4,4-tetramethyl-1,4-butanediyl di-2-propenoate (9CI)
CN Benzene, ethenyl-, polymer with 1,1-dimethylethyl 2-propenoate, 4-ethenylphenol and 1,1,4,4-tetramethyl-1,4-butanediyl di-2-propenoate (9CI)
CN Phenol, 4-ethenyl-, polymer with 1,1-dimethylethyl 2-propenoate, ethenylbenzene and 1,1,4,4-tetramethyl-1,4-butanediyl di-2-propenoate (9CI)
OTHER NAMES:
CN tert-Butyl acrylate-2,5-dimethyl-2,5-hexanediol diacrylate-p-hydroxystyrene-styrene copolymer
CN tert-Butyl acrylate-2,5-dimethylhexane-2,5-diacrylate-p-hydroxystyrene-styrene copolymer
MF (C14 H22 O4 . C8 H8 O . C8 H8 . C7 H12 O2)x
CI PMS
PCT Polyacrylic, Polystyrene
SR CA
LC STN Files: CA, CAPLUS, USPATFULL

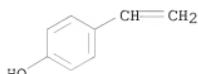
CM 1

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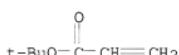
CRN 188837-15-2
CMF C14 H22 O4



CM 2
CRN 2628-17-3
CMF C8 H8 O



CM 3
CRN 1663-39-4
CMF C7 H12 O2



CM 4
CRN 100-42-5
CMF C8 H8

H₂C=CH-Ph

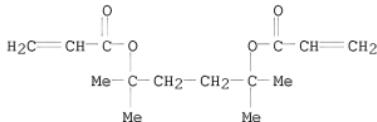
3 REFERENCES IN FILE CA (1907 TO DATE)
3 REFERENCES IN FILE CAPLUS (1907 TO DATE)

=> s 188837-15-2
L3 1 188837-15-2
(188837-15-2/RN)

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=> d

L3 ANSWER 1 OF 1 REGISTRY COPYRIGHT 2008 ACS on STN
RN 188837-15-2 REGISTRY
ED Entered STN: 07 May 1997
CN 2-Propenoic acid, 1,1'-(1,1,4,4-tetramethyl-1,4-butanediyl) ester (CA
INDEX NAME)
OTHER CA INDEX NAMES:
CN 2-Propenoic acid, 1,1,4,4-tetramethyl-1,4-butanediyl ester (9CI)
MF C14 H22 O4
CI COM
SR CA
LC STN Files: CA, CAPLUS, TOXCENTER, USPAT2, USPATFULL



PROPERTY DATA AVAILABLE IN THE 'PROP' FORMAT

14 REFERENCES IN FILE CA (1907 TO DATE)
2 REFERENCES TO NON-SPECIFIC DERIVATIVES IN FILE CA
14 REFERENCES IN FILE CAPLUS (1907 TO DATE)

=> s dimethyl and hexanediol and diacrylate
5483618 DIMETHYL
27667 HEXANEDIOL
4106 DIACRYLATE
L4 122 DIMETHYL AND HEXANEDIOL AND DIACRYLATE

=> s dimethyl and hexanediol and diacrylate
5483618 DIMETHYL
27667 HEXANEDIOL
4106 DIACRYLATE
L5 122 DIMETHYL AND HEXANEDIOL AND DIACRYLATE

=> d 122

L5 ANSWER 122 OF 122 REGISTRY COPYRIGHT 2008 ACS on STN
RN 62569-30-6 REGISTRY
ED Entered STN: 16 Nov 1984
CN 2-Propenoic acid, 2-methyl-, 2-(dimethylamino)ethyl ester, polymer
with (chloromethyl)oxirane, 1,6-hexanediyl di-2-propenoate,
1,3-isobenzofurandione, 4,4'-(1-methylethylidene)bis[phenol],
oxiranymethyl 2-methyl-2-propenoate and 2-propenoic acid (9CI) (CA
INDEX NAME)

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OTHER CA INDEX NAMES:

CN 1,3-Isobenzofurandione, polymer with (chloromethyl)oxirane, 2-(dimethylamino)ethyl 2-methyl-2-propenoate, 1,6-hexanediyl di-2-propenoate, 4,4'-(1-methylethylidene)bis[phenol], oxiranylmethyl 2-methyl-2-propenoate and 2-propenoic acid (9CI)
CN 2-Propenoic acid, 1,6-hexanediyl ester, polymer with (chloromethyl)oxirane, 2-(dimethylamino)ethyl 2-methyl-2-propenoate, 1,3-isobenzofurandione, 4,4'-(1-methylethylidene)bis[phenol], oxiranylmethyl 2-methyl-2-propenoate and 2-propenoic acid (9CI)
CN 2-Propenoic acid, 2-methyl-, oxiranylmethyl ester, polymer with (chloromethyl)oxirane, 2-(dimethylamino)ethyl 2-methyl-2-propenoate, 1,6-hexanediyl di-2-propenoate, 1,3-isobenzofurandione, 4,4'-(1-methylethylidene)bis[phenol] and 2-propenoic acid (9CI)
CN 2-Propenoic acid, polymer with (chloromethyl)oxirane, 2-(dimethylamino)ethyl 2-methyl-2-propenoate, 1,6-hexanediyl di-2-propenoate, 1,3-isobenzofurandione, 4,4'-(1-methylethylidene)bis[phenol] and oxiranylmethyl 2-methyl-2-propenoate (9CI)
CN Oxirane, (chloromethyl)-, polymer with 2-(dimethylamino)ethyl 2-methyl-2-propenoate, 1,6-hexanediyl di-2-propenoate, 1,3-isobenzofurandione, 4,4'-(1-methylethylidene)bis[phenol], oxiranylmethyl 2-methyl-2-propenoate and 2-propenoic acid (9CI)
CN Phenol, 4,4'-(1-methylethylidene)bis-, polymer with (chloromethyl)oxirane, 2-(dimethylamino)ethyl 2-methyl-2-propenoate, 1,6-hexanediyl di-2-propenoate, 1,3-isobenzofurandione, oxiranylmethyl 2-methyl-2-propenoate and 2-propenoic acid (9CI)

OTHER NAMES:

CN Acrylic acid-2-dimethylaminoethyl methacrylate-Epikote 828-glycidyl methacrylate-1,6-hexanediol diacrylate-phthalic anhydride copolymer
MF (C15 H16 O2 . C12 H18 O4 . C8 H15 N O2 . C8 H4 O3 . C7 H10 O3 . C3 H5 Cl O . C3 H4 O2)x
CI PMS

PCT Epoxy resin, Polyacrylic, Polyester, Polyester formed
LC STN Files: CA, CAPLUS

CM 1

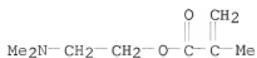
CRN 13048-33-4
CMF C12 H18 O4



CM 2

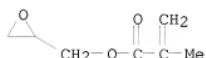
CRN 2867-47-2
CMF C8 H15 N O2

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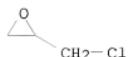
CM 3

CRN 106-91-2
CMF C7 H10 O3



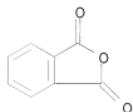
CM 4

CRN 106-89-8
CMF C3 H5 Cl O



CM 5

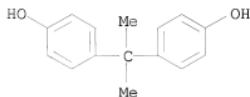
CRN 85-44-9
CMF C8 H4 O3



CM 6

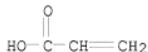
CRN 80-05-7
CMF C15 H16 O2

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CM 7

CRN 79-10-7
CMF C3 H4 O2

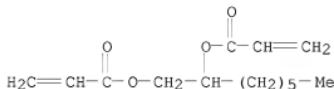


1 REFERENCES IN FILE CA (1907 TO DATE)
1 REFERENCES TO NON-SPECIFIC DERIVATIVES IN FILE CA
1 REFERENCES IN FILE CAPLUS (1907 TO DATE)

=> s C14H22O4/mf and diacrylate
2897 C14H22O4/MF
4106 DIACRYLATE
L6 2 C14H22O4/MF AND DIACRYLATE

=> d 1-2

L6 ANSWER 1 OF 2 REGISTRY COPYRIGHT 2008 ACS on STN
RN 160893-26-5 REGISTRY
ED Entered STN: 16 Feb 1995
CN 2-Propenoic acid, 1-hexyl-1,2-ethanediyl ester (9CI) (CA INDEX NAME)
OTHER NAMES:
CN 1,2-Octamethylene diacrylate
MF C14 H22 O4
SR CA
LC STN Files: CA, CAPLUS

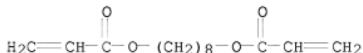


PROPERTY DATA AVAILABLE IN THE 'PROP' FORMAT

1 REFERENCES IN FILE CA (1907 TO DATE)
1 REFERENCES IN FILE CAPLUS (1907 TO DATE)

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L6 ANSWER 2 OF 2 REGISTRY COPYRIGHT 2008 ACS on STN
RN 10526-04-2 REGISTRY
ED Entered STN: 16 Nov 1984
CN 2-Propenoic acid, 1,1'-(1,8-octanediyl) ester (CA INDEX NAME)
OTHER CA INDEX NAMES:
CN 2-Propenoic acid, 1,8-octanediyl ester (9CI)
CN Acrylic acid, octamethylene ester (7CI, 8CI)
OTHER NAMES:
CN 1,8-Octamethylene diacrylate
CN 1,8-Octanediol diacrylate
CN Octamethylene diacrylate
MF C14 H22 O4
CI COM
LC STN Files: BEILSTEIN*, CA, CAOLD, CAPLUS, CASREACT, IFICDB, IFIUDB,
TOXCENTER, USPAT2, USPATFULL
(*File contains numerically searchable property data)



PROPERTY DATA AVAILABLE IN THE 'PROP' FORMAT

21 REFERENCES IN FILE CA (1907 TO DATE)
3 REFERENCES TO NON-SPECIFIC DERIVATIVES IN FILE CA
21 REFERENCES IN FILE CAPLUS (1907 TO DATE)
1 REFERENCES IN FILE CAOLD (PRIOR TO 1967)

=> file ca		SINCE FILE	TOTAL
COST IN U.S. DOLLARS		ENTRY	SESSION
FULL ESTIMATED COST		76.56	76.77

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FILE COVERS 1907 - 13 Mar 2008 VOL 148 ISS 12

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FILE LAST UPDATED: 13 Mar 2008 (20080313/ED)

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This file contains CAS Registry Numbers for easy and accurate substance identification.

=> d his

(FILE 'HOME' ENTERED AT 20:31:49 ON 14 MAR 2008)

FILE 'REGISTRY' ENTERED AT 20:32:00 ON 14 MAR 2008

L1	0 S DIMETHYLHEXANE AND DIMETHACRYLATE
L2	2 S DIMETHYLHEXANE AND DIACRYLATE
L3	1 S 188837-15-2
L4	122 S DIMETHYL AND HEXANEDIOL AND DIACRYLATE
L5	122 S DIMETHYL AND HEXANEDIOL AND DIACRYLATE
L6	2 S C14H22O4/MF AND DIACRYLATE

FILE 'CA' ENTERED AT 20:35:52 ON 14 MAR 2008

=> d 13

YOU HAVE REQUESTED DATA FROM FILE 'REGISTRY' - CONTINUE? (Y)/N:n

=> s 13

L7 14 L3

=> d all 1-14

L7 ANSWER 1 OF 14 CA COPYRIGHT 2008 ACS on STN
AN 148:202003 CA
ED Entered STN: 21 Feb 2008
TI Design of Reversible Cross-Linkers for Step and Flash Imprint Lithography
Imprint Resists
AU Palmieri, Frank; Adams, Jacob; Long, Brian; Heath, William; Tsiaertas,
Pavlos; Willson, C. Grant
CS Department of Chemical Engineering, The University of Texas at Austin,
Austin, TX, 78712, USA
SO ACS Nano (2007), 1(4), 307-312
CODEN: ANCAC3; ISSN: 1936-0851
PB American Chemical Society
DT Journal
LA English
CC 74-5 (Radiation Chemistry, Photochemistry, and Photographic and Other
Reproductive Processes)
AB Step and flash imprint lithog. is a low-cost, nanoimprint lithog. process
that generates nanopatterned polymeric films via the photopolymer. of
low-viscosity solns. containing crosslinking monomers in a transparent
template (mold). The highly crosslinked imprint materials are completely
insol. in all inert solvents, which poses a problem for reworking wafers
with faulty imprints and cleaning templates contaminated with cured
imprint resist. Degradable crosslinkers provide a means of stripping
crosslinked polymer networks. The controlled degradation of polymers
containing

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acetal- and tertiary ester-based crosslinkers is demonstrated here. The viscosity and dose to cure are presented for several prepolymer formulations, along with imprint resolution and tensile modulus results for the cured polymers. Optimum conditions for de-crosslinking and stripping of the crosslinked polymers are presented, including demonstrations of their utility.

ST degrdn polymer acetal ester based crosslinker nanoimprint lithog resist; resist formulation crosslinker step flash imprint lithog

IT Photoresists
(degradable crosslinking monomers for step-and-flash imprint lithog. resist formulations)

IT Crosslinking agents
Young's modulus
(degradable crosslinking monomers for step-and-flash imprint lithog. resist formulations that are strippable and reworkable)

IT Lithography
(nano-, imprint, step-and-flash; degradable crosslinking monomers for step-and-flash imprint lithog. resist formulations that are strippable and reworkable)

IT Crosslinking
(photochem.; degradable crosslinking monomers for step-and-flash imprint lithog. resist formulations that are strippable and reworkable)

IT Polymerization
(photopolymn.; degradable crosslinking monomers for step-and-flash imprint lithog. resist formulations that are strippable and reworkable)

IT 188837-15-2, reactions 1003196-88-0
RL: PEP (Physical, engineering or chemical process); PRP (Properties); RCT
(Reactant); PROC (Process); RACT (Reactant or reagent)
(monomer, resist formulation; degradable crosslinking monomers for step-and-flash imprint lithog. resist formulations that are strippable and reworkable)

IT 7473-98-5, 2-Hydroxy-2-methyl-1-phenyl-1-propanone
RL: CAT (Catalyst use); USES (Uses)
(resist formulation; degradable crosslinking monomers for step-and-flash imprint lithog. resist formulations that are strippable and reworkable)

IT 101-43-9, Cyclohexyl methacrylate 106-63-8, Isobutyl acrylate
RL: PEP (Physical, engineering or chemical process); PRP (Properties); RCT
(Reactant); PROC (Process); RACT (Reactant or reagent)
(resist formulation; degradable crosslinking monomers for step-and-flash imprint lithog. resist formulations that are strippable and reworkable)

IT 76-05-1, Trifluoroacetic acid, uses 108-10-1, 4-Methyl-2-pentanone
RL: NUU (Other use, unclassified); USES (Uses)
(stripping solution; degradable crosslinking monomers for step-and-flash imprint lithog. resist formulations that are strippable and reworkable)

RE.CNT 18 THERE ARE 18 CITED REFERENCES AVAILABLE FOR THIS RECORD
RE

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(1) Bailey, T; J Vac Sci Technol, B: Microelectron Nanometer Struct 2000, V18, P3572 CA
(2) Chen, X; Macromolecules 2003, V36, P1802 CA
(3) de Clercq, R; Macromolecules 1992, V25, P1109 CA
(4) Dickey, M; AIChE J 2005, V51, P2547 CA
(5) Heath, W; Macromolecules, submitted 2007
(6) Johnson, S; Proc SPIE-Int Soc Opt Eng, Emerging Lithographic Technologies VII 2003, V5037(Pt 1), P197
(7) Kim, E; J Vac Sci Technol, B: Microelectron Nanometer Struct-Process, Meas, Phenom 2004, V22, P131 CA
(8) Kim, E; J Vac Sci Technol, B: Microelectron Nanometer Struct-Process, Meas, Phenom 2004, V22, P131 CA
(9) Le, N; Microelectron Eng 2005, V78-79, P464 CA
(10) Mancini, D; Proc SPIE-Int Soc Opt Eng, Emerging Lithographic Technologies VIII 2004, V5374(Pt 1), P371
(11) McMackin, I; Proc SPIE-Int Soc Opt Eng, Emerging Lithographic Technologies VIII 2003, V5037(Pt 1), P178
(12) McMackin, I; Proc SPIE-Int Soc Opt Eng, Emerging Lithographic Technologies VIII 2004, V5374(Pt 1), P222
(13) Ogino, K; Chem Mater 1998, V10, P3833 CA
(14) Resnick, D; Proc SPIE-Int Soc Opt Eng, Emerging Lithographic Technologies VII 2003, V5037(Pt 1), P12
(15) Resnick, D; Proc SPIE-Int Soc Opt Eng, Nanostructure Science, Metrology, and Technology 2002, V4608, P176 CA
(16) Shirai, M; Chem Mater 2003, V15, P4075 CA
(17) Sijbesma, R; Science (Washington, D C) 1997, V278, P1601 CA
(18) Stockmayer, W; J Chem Phys 1943, V11, P45 CA

L7 ANSWER 2 OF 14 CA COPYRIGHT 2008 ACS on STN

AN 147:542837 CA

ED Entered STN: 13 Dec 2007

TI Radically curable adhesive compositions with good repair property, circuit

connection materials, connected members, and semiconductor devices

IN Katogi, Shigeki

PA Hitachi Chemical Co., Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 22pp.

CODEN: JKXXAF

DT Patent

LA Japanese

CC 38-3 (Plastics Fabrication and Uses)
Section cross-reference(s): 76

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	JP 2007297579	A	20071115	JP 2006-178590	20060628
PRAI	JP 2006-104090	A	20060405		

CLASS

PATENT NO.	CLASS	PATENT FAMILY CLASSIFICATION CODES
JP 2007297579	IPCI	C09J0004-00 [I,A]; C09J0009-02 [I,A]; C09J0009-00

[I,C*]; H01B0001-22 [I,A]; H01L0021-60 [I,A];
H01L0021-02 [I,C*]; H05K0003-36 [I,A]; H05K0003-32
[I,A]

IFCR C09J0004-00 [I,C]; C09J0004-00 [I,A]; C09J0009-00
[I,C]; C09J0009-02 [I,A]; H01B0001-22 [I,C];
H01B0001-22 [I,A]; H01L0021-02 [I,C]; H01L0021-60
[I,A]; H05K0003-32 [I,C]; H05K0003-32 [I,A];
H05K0003-36 [I,C]; H05K0003-36 [I,A];
H05K0003-36 [I,C]; H05K0003-36 [I,A];
4J040/DC071; 4J040/DC072; 4J040/DF031; 4J040/DF032;
4J040/EE001; 4J040/EE002; 4J040/EE051; 4J040/EE052;
4J040/EE061; 4J040/EE062; 4J040/EF001; 4J040/EF002;
4J040/EG001; 4J040/EG002; 4J040/EH031; 4J040/EH032;
4J040/EL001; 4J040/EL002; 4J040/FA131; 4J040/FA132;
4J040/FA281; 4J040/FA282; 4J040/HA026; 4J040/HA066;
4J040/HB14; 4J040/HB41; 4J040/HC14; 4J040/HD23;
4J040/KA12; 4J040/KA32; 4J040/LA09; 4J040/NA19;
4J040/NA20; 5E319/AA03; 5E319/AB05; 5E319/AC11;
5E319/BB12; 5E319/BB16; 5E319/CC03; 5E319/GG15;
5E344/AA02; 5E344/AA22; 5E344/BB02; 5E344/CC13;
5E344/CC21; 5E344/CD04; 5E344/CD05; 5E344/EE21;
5F044/LL09; 5F044/NN12; 5F044/NN19; 5G301/DA03;
5G301/DA05; 5G301/DA06; 5G301/DA10; 5G301/DA18;
5G301/DA42; 5G301/DD03; 5G301/DE01

AB The adhesive compns. contain (a) thermoplastic resins, (b) radically polymerizable compds. having ≥ 2 (meth)acryloyloxy groups a part of or all of which are forming direct bonds or secondary carbon or tertiary carbon, and (c) radical polymerization initiators. Optionally, the adhesive compns. further contain (a) vinyl compds. bearing phosphoric acid groups 0.1-20 parts per 100 parts of the thermoplastic resins. Optionally, the adhesive compns. further contain elec. conductive particles. Circuit members having circuit electrodes are connected elec. with the circuit electrodes by using the adhesive compns. The semiconductor device comprises (i) a substrate having a circuit electrode on the mounting surface and a semiconductor element mounted on the substrate and elec. connected with the circuit electrode via (iii) the connector comprising cured articles of the adhesive compns. Thus, 10.00 g 2,5-dimethyl-2,5-hexanediol was reacted with 14.85 g acryloyl chloride in THF in the presence of Et₃N and 4-dimethylaminopyridine to give 7.78 g of an oil with 2 acryloyloxy groups directly bonded to tertiary carbon, 5 parts of which was mixed with a MEK solution of PKHC (phenoxy resin) 87.5 (resin 35 parts), poly(butylene adipate) diol-polyoxytetramethylene glycol-1,4-butylene glycol-diphenylmethane diisocyanate copolymer (Mw 1.5 + 105) 15, Hitaloid 4861 (polyfunctional urethane acrylate) 45, and Light Ester P 2M [2-(meth)acryloyloxyethyl phosphate] 5, and Perhexyl O (tert-hexylperoxy 2-ethylhexanoate) 3 parts. The obtained solution was mixed with elec. conductive metal-coated polystyrene particles (Au/Ni/polystyrene particle, average particle diameter 4 μ m) to give elec. conductive adhesive composition with content of conductive particles 1.5 volume%. The adhesive composition was applied on a fluoropolymer film and dried at 70° for 10 min to give

a 15- μ m thick film-type circuit connection material, which was disposed between a flexible printed circuit board and a thin-film ITO-coated glass substrate and hot-pressed at 160° to give a connector showing connection resistance 1.6 Ω , adhesion strength (90° peel) 680 N/m, and easy removal from the ITO substrate.

ST elec conductive adhesive film acryloyloxy compd; phenoxy resin adhesive film elec packaging; semiconductor device adhesive film circuit connection; polyurethane adhesive film elec packaging

IT Adhesive films
(elec. conductive; radically curable adhesive compns. containing thermoplastic reins for circuit bonding films)

IT Printed circuit boards
(flexible, laminate with; radically curable adhesive compns. containing thermoplastic reins for circuit bonding films)

IT Polyurethanes, uses
RL: IMF (Industrial manufacture); POF (Polymer in formulation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (polyester-polyoxyalkylene-, block; radically curable adhesive compns. containing thermoplastic reins for circuit bonding films)

IT Electronic packaging materials
(radically curable adhesive compns. containing thermoplastic reins for circuit bonding films)

IT Phenoxy resins
RL: POF (Polymer in formulation); TEM (Technical or engineered material use); USES (Uses)
(radically curable adhesive compns. containing thermoplastic reins for circuit bonding films)

IT Glass substrates
(with ITO layer, laminate with; radically curable adhesive compns. containing thermoplastic reins for circuit bonding films)

IT 7440-02-0, Nickel, uses 7440-57-5, Gold, uses 9003-53-6, Polystyrene
RL: TEM (Technical or engineered material use); USES (Uses)
(Au/Ni/polystyrene elec. conductive particle; radically curable adhesive compns. containing thermoplastic reins for circuit bonding films)

IT 110680-75-6P
RL: IMF (Industrial manufacture); POF (Polymer in formulation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)
(assumed monomers; radically curable adhesive compns. containing thermoplastic reins for circuit bonding films)

IT 956699-55-1P 956699-56-2P 956699-57-3P
RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)
(crosslinked; radically curable adhesive compns. containing thermoplastic reins for circuit bonding films)

IT 50926-11-9, ITO
RL: TEM (Technical or engineered material use); USES (Uses)
(layer on glass substrate, laminate with; radically curable adhesive compns. containing thermoplastic reins for circuit bonding films)

IT 137791-98-1, Perhexyl O
RL: CAT (Catalyst use); USES (Uses)
(radically curable adhesive compns. containing thermoplastic reins for circuit bonding films)

FOR 10559534 by Cynthia Hamilton

IT 188837-15-2P, uses
RL: IMF (Industrial manufacture); RCT (Reactant); TEM (Technical or
engineered material use); PREP (Preparation); RACT (Reactant or reagent);
USES (Uses)
 (radically curable adhesive compns. containing thermoplastic reins for
 circuit bonding films)
IT 25068-38-6, PKHC
RL: POF (Polymer in formulation); TEM (Technical or engineered material
use); USES (Uses)
 (radically curable adhesive compns. containing thermoplastic reins for
 circuit bonding films)
IT 52496-08-9, APG 400 52628-03-2, Light Ester P 2M 61722-28-9, Light
Acrylate BP 4PA 956595-72-5, Hitaloid 4861
RL: RCT (Reactant); TEM (Technical or engineered material use); RACT
(Reactant or reagent); USES (Uses)
 (radically curable adhesive compns. containing thermoplastic reins for
 circuit bonding films)

L7 ANSWER 3 OF 14 CA COPYRIGHT 2008 ACS on STN

AN 145:302806 CA

ED Entered STN: 28 Sep 2006

TI Chemical amplification-type positive-working resist composition
containing

 star copolymer and method of forming resist pattern

IN Takayama, Toshikazu

PA Tokyo Ohka Kogyo Co., Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 26pp.

CODEN: JKXXAF

DT Patent

LA Japanese

CC 74-5 (Radiation Chemistry, Photochemistry, and Photographic and Other
Reproductive Processes)

Section cross-reference(s): 35, 38

FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
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PI JP 2006227532	A	20060831	JP 2005-44597	20050221
PRAI JP 2005-44597		20050221		

CLASS

PATENT NO.	CLASS	PATENT FAMILY CLASSIFICATION CODES
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JP 2006227532	IPCI	G03F0007-039 [I,A]; C08F0212-08 [I,A]; C08F0212-00 [I,C*]; C08F0222-10 [I,A]; C08F0222-00 [I,C*]; H01L0021-027 [I,A]; H01L0021-02 [I,C*] IPCR G03F0007-039 [I,C]; G03F0007-039 [I,A]; C08F0212-00 [I,C]; C08F0212-08 [I,A]; C08F0222-00 [I,C]; C08F0222-10 [I,A]; H01L0021-02 [I,C]; H01L0021-027 [I,A] FTERM 2H025/AA01; 2H025/AA02; 2H025/AB16; 2H025/AC04; 2H025/AC08; 2H025/AD03; 2H025/BE00; 2H025/BE10; 2H025/BG00; 2H025/BJ10; 2H025/CB14; 2H025/CB17; 2H025/CB41; 2H025/FA12; 4J100/AB07P; 4J100/AL08Q; 4J100/AL62R; 4J100/BA03P; 4J100/BC04Q; 4J100/BC09Q; 4J100/CA05; 4J100/JA38

AB Disclosed is a chemical amplification-type pos.-working resist
composition

comprising (a) a resin increasing its alkali solubility upon interaction with acid and (b) a photoacid, wherein the resin is a star copolymer with the arm made up of an alkenyl phenol having an ethylenic bond and an acrylate derivative and the core made up of a diacrylate derivative

ST chem amplification pos working resist compn photoresist star copolymer; acrylate alkenylphenol diacrylate

IT Photoresists
(Chemical amplification-type pos.-working resist composition containing star copolymer)

IT 188837-15-2DP, star copolymer
RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)
(core; Chemical amplification-type pos.-working resist composition containing star copolymer)

IT 871677-01-9P, 4-(2-Ethoxyethoxy)styrene-styrene copolymer 900806-20-4P, 1-Ethylcyclohexyl methacrylate homopolymer 907590-83-4P, 4-(2-Ethoxyethoxy)styrene-1-ethylcyclohexyl methacrylate copolymer 907590-85-6P, 1-Adamantyl methacrylate-4-(2-Ethoxyethoxy)styrene-styrene copolymer 907590-86-7P 907590-87-8P
RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)
(star; Chemical amplification-type pos.-working resist composition containing star copolymer)

L7 ANSWER 4 OF 14 CA COPYRIGHT 2008 ACS on STN

AN 145:272436 CA

ED Entered STN: 21 Sep 2006

TI Phenol-based star polymers with narrow molecular weight distribution

IN Nakamura, Mitsuhiro; Kubo, Hideo; Shimotori, Takeshi

PA Nippon Soda Co., Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 18pp.

CODEN: JKXXAF

DT Patent

LA Japanese

CC 37-3 (Plastics Manufacture and Processing)

FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI JP 2006225605	A	20060831	JP 2005-44378	20050221
PRAI JP 2005-44378		20050221		

CLASS

PATENT NO.	CLASS	PATENT FAMILY CLASSIFICATION CODES
JP 2006225605	IPCI	C08F0265-06 [I,A]; C08F0265-00 [I,C*]
	IPCR	C08F0265-00 [I,C]; C08F0265-06 [I,A]
	FTERM	4J026/AA46; 4J026/BA05; 4J026/BA06; 4J026/BA27; 4J026/DA02; 4J026/DA03; 4J026/DA18; 4J026/DB02; 4J026/DB18; 4J026/FA07; 4J026/GA01

AB The polymers, useful for ESCAP-type resists, consist of cores containing polymer chains consisting of $H_2C:CR_4CO_2CR_5R_6R_7CR_9R_10OCOC(R_8):CH_2$ -derived unit ($R_4, R_8 = H, Me$; $R_5, R_6, R_9, R_{10} = alkyl$; $R_7 = alkylene$) and arms,

which are polymer chains containing HO-p-C6H4C(R1):CH2-derived unit (R = H, Me) and H2C:CR2CO2R3-derived unit (R2 = H, Me; R3 = alicyclic hydrocarbon bonded by tertiary carbon). Thus, adding 41.6 g 4-(2-ethoxyethoxy)styrene to a THF/hexane mixture containing BuLi at -40°, terminating the reaction by MeOH, cooling the reaction system to -60°, adding THF containing LiCl and 5.3 g 1-ethylcyclohexyl methacrylate, terminating the reaction by MeOH, further adding 6.9 g 2,5-dimethylhexyl 2,5-diacrylate to the mixture, precipitating polymer in MeOH, filtering, washing, and drying gave 53 g polymer with Mn 29,000 in star polymer parts and 7100 in arm polymer parts and Mw/Mn 1.17 and 1.15, resp. It was dissolved in THF, hydrolyzed with HCl, filtered, washed, and dried to give 38 g polymer with Mn 24,000 in star polymer parts and 5600 in arm polymer parts and Mw/Mn 1.18 and 1.14, resp.

ST hydroxystyrene ethylcyclohexyl methacrylate star polymer dimethylhexyl diacrylate core

IT Polymers, preparation
RL: IMF (Industrial manufacture); PRP (Properties); PREP (Preparation) (star-branched; phenol-based star polymers with narrow mol. weight distribution)

IT 131787-39-8 188837-15-2
RL: RCT (Reactant); RACT (Reactant or reagent) (core; phenol-based star polymers with narrow mol. weight distribution)

IT 907590-83-4DP, hydrolyzed 907590-84-5DP, hydrolyzed 907590-85-6DP, hydrolyzed 907590-86-7DP, hydrolyzed 907590-87-8DP, hydrolyzed
RL: IMF (Industrial manufacture); PRP (Properties); PREP (Preparation) (star, arms; phenol-based star polymers with narrow mol. weight distribution)

L7 ANSWER 5 OF 14 CA COPYRIGHT 2008 ACS on STN

AN 145:58810 CA

ED Entered STN: 13 Jul 2006

TI Solid surface with immobilized degradable cationic polymer for transfecting eukaryotic cells

IN Tanaka, Yasunobu; Castello, Chris P.; Yu, Lei

PA Nitto Denko Corp., Japan

SO PCT Int. Appl., 41 pp.

CODEN: PIXXD2

DT Patent

LA English

CC 9-2 (Biochemical Methods)

Section cross-reference(s): 3, 16

FAN.CNT 2

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	WO 2006066001	A2	20060622	WO 2005-US45429	20051214
	WO 2006066001	A3	20060908		

W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KM, KN, KP, KR,

KZ, LC, LK, LR, LS, LT, LU, LV, LY, MA, MD, MG, MK, MN, MW, MX,
 MZ, NA, NG, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE,
 SG, SK, SL, SM, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC,
 VN, YU, ZA, ZM, ZW
 RW: AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE,
 IS, IT, LT, LU, LV, MC, NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ,
 CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG, BW, GH,
 GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY,
 KG, KZ, MD, RU, TJ, TM
 AU 2005316501 A1 20060622 AU 2005-316501 20051214
 CA 2590820 A1 20060622 CA 2005-2590820 20051214
 EP 1836293 A2 20070926 EP 2005-654197 20051214
 R: AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE,
 IS, IT, LI, LT, LU, LV, MC, NL, PL, PT, RO, SE, SI, SK, TR
 CN 101124316 A 20080213 CN 2005-80043453 20070618
 KR 2007101859 A 20071017 KR 2007-715797 20070710
 PRAI US 2004-637344P P 20041217
 WO 2005-US45429 W 20051214

CLASS

PATENT NO.	CLASS	PATENT FAMILY CLASSIFICATION CODES
WO 2006066001	IPCI	C12M0001-16 [I,A]; C12M0003-00 [I,A]; C12M0001-16 [I,C]; C12M0003-00 [I,C]; C12M0001-16 [I,A]; C12M0003-00 [I,A]
	IPCR	C12M0001-16 [I,A]; C12M0001-16 [I,C]; C12M0003-00 [I,C]; C12M0003-00 [I,A]
	ECLA	C12N005/00S
AU 2005316501	IPCI	C12M0001-16 [I,C]; C12M0001-16 [I,A]; C12M0003-00 [I,C]; C12M0003-00 [I,A]
	IPCR	C12M0001-16 [I,C]; C12M0001-16 [I,A]; C12M0003-00 [I,C]; C12M0003-00 [I,A]
	NCL	435/455.000; 435/285.200; 435/468.000
	ECLA	C12N005/00S; M12N; M12N
CA 2590820	IPCI	C12M0001-16 [I,A]; C12M0003-00 [I,A]
	IPCR	C12M0001-16 [I,C]; C12M0001-16 [I,A]; C12M0003-00 [I,C]; C12M0003-00 [I,A]
	NCL	435/455.000; 435/285.200; 435/468.000
	ECLA	C12N005/00S; M12N; M12N
EP 1836293	IPCI	C12M0001-16 [I,A]; C12M0003-00 [I,A]
	IPCR	C12M0001-16 [I,C]; C12M0001-16 [I,A]; C12M0003-00 [I,C]; C12M0003-00 [I,A]
	ECLA	C12N005/00S; M12N; M12N
CN 101124316	IPCI	C12M0001-16 [I,A]; C12M0003-00 [I,A]
KR 2007101859	IPCI	C12M0003-00 [I,A]
AB	A cell transfection/culture device is disclosed which includes a solid support coated with a degradable polymer cation as a transfection reagent.	
	The transfection/culture device is conveniently stored at room temperature until	
	use. Cell transfection is accomplished easily by adding the nucleic acid of interest and the cells to be transfected to the transfection/culture device. Cell transfection is completed in less than one hour by using the transfection/culture device described herein.	
ST	cationic polymer immobilized cell transfection	

IT Animal cell line
(293; solid surface with immobilized degradable cationic polymer for transfecting eukaryotic cells)

IT Animal cell line
(705; solid surface with immobilized degradable cationic polymer for transfecting eukaryotic cells)

IT Animal cell line
(A549; solid surface with immobilized degradable cationic polymer for transfecting eukaryotic cells)

IT Animal cell line
(CHO; solid surface with immobilized degradable cationic polymer for transfecting eukaryotic cells)

IT Animal cell line
(COS-7; solid surface with immobilized degradable cationic polymer for transfecting eukaryotic cells)

IT Animal cell line
(CV-1; solid surface with immobilized degradable cationic polymer for transfecting eukaryotic cells)

IT Animal cell line
(DU-145; solid surface with immobilized degradable cationic polymer for transfecting eukaryotic cells)

for
IT Animal cell line
(HT-1080; solid surface with immobilized degradable cationic polymer for transfecting eukaryotic cells)

IT Animal cell line
(Hep G2; solid surface with immobilized degradable cationic polymer for transfecting eukaryotic cells)

IT Animal cell line
(MDCK; solid surface with immobilized degradable cationic polymer for transfecting eukaryotic cells)

IT Acid halides
RL: BUU (Biological use, unclassified); TEM (Technical or engineered material use); BIOL (Biological study); USES (Uses)
(acid chlorides, dimers and multimers of, as a linker; solid surface with immobilized degradable cationic polymer for transfecting eukaryotic cells)

IT Immobilization, molecular or cellular
(biochem.; solid surface with immobilized degradable cationic polymer for transfecting eukaryotic cells)

IT Polyelectrolytes
(cationic; solid surface with immobilized degradable cationic polymer for transfecting eukaryotic cells)

IT Polysaccharides, biological studies
Proteins
RL: BSU (Biological study, unclassified); TEM (Technical or engineered material use); BIOL (Biological study); USES (Uses)
(cationic; solid surface with immobilized degradable cationic polymer for transfecting eukaryotic cells)

IT Human
(cell lines; solid surface with immobilized degradable cationic polymer for transfecting eukaryotic cells)

IT Sulfonyl halides

RL: BUU (Biological use, unclassified); TEM (Technical or engineered material use); BIOL (Biological study); USES (Uses)
(chlorides, dimers and multimers of, as a linker; solid surface with immobilized degradable cationic polymer for transfecting eukaryotic cells)

IT Aldehydes, biological studies
Anhydrides
Carboxylic acids, biological studies
Epoxides
Halides
Isocyanates
Isothiocyanates
RL: BUU (Biological use, unclassified); TEM (Technical or engineered material use); BIOL (Biological study); USES (Uses)
(dimers and multimers of, as a linker; solid surface with immobilized degradable cationic polymer for transfecting eukaryotic cells)

IT Gene, animal
RL: BSU (Biological study, unclassified); BIOL (Biological study)
(egfp; solid surface with immobilized degradable cationic polymer for transfecting eukaryotic cells)

IT Proteins
RL: BSU (Biological study, unclassified); BIOL (Biological study)
(green fluorescent, enhanced; solid surface with immobilized degradable cationic polymer for transfecting eukaryotic cells)

IT Gene, animal
RL: BSU (Biological study, unclassified); BIOL (Biological study)
(luc; solid surface with immobilized degradable cationic polymer for transfecting eukaryotic cells)

IT Animal cell
(mammalian; solid surface with immobilized degradable cationic polymer for transfecting eukaryotic cells)

IT Animal tissue culture
Cell division
Cytotoxicity
HeLa cell
Linking agents
Microtiter plates
Plant cell
Plasmid vectors
Polymerization
Stem cell
Storage
(solid surface with immobilized degradable cationic polymer for transfecting eukaryotic cells)

IT Double stranded RNA
Nucleic acids
Ribozymes
RL: BSU (Biological study, unclassified); BIOL (Biological study)
(solid surface with immobilized degradable cationic polymer for transfecting eukaryotic cells)

IT Epoxy resins, biological studies
RL: BUU (Biological use, unclassified); BIOL (Biological study); USES (Uses)
(solid surface with immobilized degradable cationic polymer for

transfected eukaryotic cells)

IT Glass, biological studies
RL: BUU (Biological use, unclassified); BIOL (Biological study); USES (Uses)
(solid surface with immobilized degradable cationic polymer for
transfected eukaryotic cells)

IT Transformation, genetic
(transfection; solid surface with immobilized degradable cationic
polymer for transfected eukaryotic cells)

IT Polyesters, biological studies
RL: BUU (Biological use, unclassified); TEM (Technical or engineered
material use); BIOL (Biological study); USES (Uses)
(with at least three acrylate or acrylamide side groups; solid surface
with immobilized degradable cationic polymer for transfecting
eukaryotic cells)

IT 1070-70-8, 1,4-Butanediol diacrylate 4986-89-4, Pentaerythritol
tetraacrylate 13048-33-4, 1,6-Hexanediol diacrylate 15625-89-5,
Trimethylolpropane triacrylate 19485-03-1, 1,3-Butanediol diacrylate
26570-48-9, Polyethylene glycol diacrylate 60506-81-2,
DiPentaerythritol
pentaacrylate 94108-97-1, Di(trimethylolpropane) tetraacrylate
184223-36-7, 2,4-Pentanediol diacrylate 188837-15-2
188837-17-4
RL: BUU (Biological use, unclassified); TEM (Technical or engineered
material use); BIOL (Biological study); USES (Uses)
(as a linker; solid surface with immobilized degradable cationic
polymer for transfecting eukaryotic cells)

IT 26937-01-9, PAMAM
RL: BSU (Biological study, unclassified); TEM (Technical or engineered
material use); BIOL (Biological study); USES (Uses)
(dendritic; solid surface with immobilized degradable cationic polymer
for transfecting eukaryotic cells)

IT 79-06-1, Acrylamide, biological studies 79-10-7, Acrylic acid,
biological studies 541-59-3, Maleimide 6066-82-6D,
N-Hydroxysuccinimide, esters
RL: BUU (Biological use, unclassified); TEM (Technical or engineered
material use); BIOL (Biological study); USES (Uses)
(dimers and multimers of, as a linker; solid surface with immobilized
degradable cationic polymer for transfecting eukaryotic cells)

IT 144189-73-1, DOTAP 344612-27-7, LipofectAMINE 2000 762293-77-6,
JetPEI
RL: BSU (Biological study, unclassified); BIOL (Biological study)
(solid surface with immobilized degradable cationic polymer for
transfected eukaryotic cells)

IT 9002-98-6
RL: BSU (Biological study, unclassified); CPS (Chemical process); PEP
(Physical, engineering or chemical process); RCT (Reactant); TEM
(Technical or engineered material use); BIOL (Biological study); PROC
(Process); RACT (Reactant or reagent); USES (Uses)
(solid surface with immobilized degradable cationic polymer for
transfected eukaryotic cells)

IT 26338-45-4P
RL: BSU (Biological study, unclassified); SPN (Synthetic preparation);
TEM
(Technical or engineered material use); BIOL (Biological study); PREP

FOR 10559534 by Cynthia Hamilton

(Preparation); USES (Uses)
(solid surface with immobilized degradable cationic polymer for
transfected eukaryotic cells)
IT 56-18-8, N-(3-Aminopropyl)-1,3-propanediamine 71-44-3, Spermine
124-20-9, Spermidine 140-31-8, N-(2-Aminoethyl)piperazine 4097-89-6,
Tri(2-aminoethyl)amine 4741-99-5, N,N'-Bis(2-aminoethyl)-1,3-
propanediamine 7209-38-3, 1,4-Bis(3-aminopropyl)piperazine 9003-53-6,
Polystyrene 9012-76-4, Chitosan 13531-52-7, N-(2-Aminoethyl)-1,3-
propanediamine 25037-42-7, Polypropyleneimine 25104-18-1,
Poly(L-lysine) 25154-86-3, PDMAEMA 50985-30-3,
N,N'-Bis(2-aminopropyl)-
ethylenediamine
RL: BSU (Biological study, unclassified); TEM (Technical or engineered
material use); BIOL (Biological study); USES (Uses)
(solid surface with immobilized degradable cationic polymer for
transfected eukaryotic cells)
IT 26913-06-4P, Poly[imino(1,2-ethanediyl)]
RL: BUU (Biological use, unclassified); SPN (Synthetic preparation); TEM
(Technical or engineered material use); BIOL (Biological study); PREP
(Preparation); USES (Uses)
(solid surface with immobilized degradable cationic polymer for
transfected eukaryotic cells)
IT 25038-59-9, Mylar, biological studies
RL: BUU (Biological use, unclassified); TEM (Technical or engineered
material use); BIOL (Biological study); USES (Uses)
(solid surface with immobilized degradable cationic polymer for
transfected eukaryotic cells)
IT 7647-01-0, Hydrogen chloride, processes
RL: CPS (Chemical process); PEP (Physical, engineering or chemical
process); PROC (Process)
(solid surface with immobilized degradable cationic polymer for
transfected eukaryotic cells)

L7 ANSWER 6 OF 14 CA COPYRIGHT 2008 ACS on STN
AN 144:360317 CA
ED Entered STN: 27 Apr 2006
TI Positive photosensitive resin and novel dithiol compound
IN Mita, Takahito
PA Japan
SO U.S. Pat. Appl. Publ., 13 pp.
CODEN: USXXCO
DT Patent
LA English
INCL 430270100
CC 74-5 (Radiation Chemistry, Photochemistry, and Photographic and Other
Reproductive Processes)
Section cross-reference(s): 35, 38
FAN_CNT 1
PATENT NO. KIND DATE APPLICATION NO. DATE
----- ----- ----- -----
PI US 2006068324 A1 20060330 US 2005-235045 20050926
JP 2006091762 A 20060406 JP 2004-280353 20040927
PRAI JP 2004-280353 A 20040927
CLASS
PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES

US 2006068324 INCL 430270100
IPCI G03C0001-76 [I,A]
IPCR G03C0001-76 [I,A]; G03C0001-76 [I,C]
NCL 430/270.100
ECLA G03F007/039C; G03F007/039C1S
JP 2006091762 IPCI G03F0007-039 [I,A]; C07C0323-52 [I,A]; C07C0323-00
[I,C*]; H01L0021-027 [I,A]; H01L0021-02 [I,C*]
FTERM 2H025/AA01; 2H025/AA02; 2H025/AA04; 2H025/AB16;
2H025/AC04; 2H025/AC08; 2H025/AD03; 2H025/BE00;
2H025/BE10; 2H025/BG00; 2H025/BJ10; 2H025/CB14;
2H025/CB20; 2H025/CB41; 4H006/AA01; 4H006/AA03;
4H006/AB46; 4H006/TA04; 4H006/TB52

OS MARPAT 144:360317

AB A pos. photosensitive resin having, in the high-mol. main chain, a structure represented by the following general formula -S-R1-COO-CR4R5-R3-CR6R7-OOC-R2-S- (R1-3 = bivalent hydrocarbon; R4-7 = monovalent hydrocarbon with Cl-4) : and a dithiol compound represented by the following general formula HS-R1-COO-CR4R5-R3-CR6R7-OOC-R2-SH. The pos. photosensitive resin can alleviate the problems of conventional technique and, when used for formation of a fine pattern in semiconductor production, can show a higher resist sensitivity than conventional products and can bring about effects such as reduction in impurities after development.

The dithiol compound is novel and extremely suitable for use in production of the pos. photosensitive resin.

ST pos photosensitive resin dithiol compd resist

IT Photoinitiation materials
Resists
(Pos. photosensitive resin and novel dithiol compound for resists)

IT 881211-92-3P 881211-93-4P 881211-94-5P
RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)
(Pos. photosensitive resin and novel dithiol compound for resists)

IT 864679-84-5P 881211-91-2P
RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation);

RACT
(Reactant or reagent)
(preparation of novel dithiol compound for resists)

IT 507-09-5, Thioacetic acid, reactions 188837-15-2
RL: RCT (Reactant); RACT (Reactant or reagent)
(preparation of novel dithiol compound for resists)

L7 ANSWER 7 OF 14 CA COPYRIGHT 2008 ACS on STN
AN 143:388238 CA
ED Entered STN: 10 Nov 2005
TI Efficient release method of adherends without damaging nor residual adhesive
IN Iwanaga, Shinichiro; Ota, Masaru
PA JSR Ltd., Japan
SO Jpn. Kokai Tokyo Koho, 9 pp.
CODEN: JKXXAF
DT Patent

LA Japanese
 IC ICM C09J005-00
 ICS H01L021-304
 CC 38-3 (Plastics Fabrication and Uses)
 FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI JP 2005290146	A	20051020	JP 2004-105578	20040331
PRAI JP 2004-105578		20040331		

CLASS

PATENT NO.	CLASS	PATENT FAMILY CLASSIFICATION CODES
JP 2005290146	ICM C09J005-00 ICS H01L021-304 IPCI C09J0005-00 [ICM, 7]; H01L0021-304 [ICS, 7]; H01L0021-02 [ICS, 7, C*] IPCR C09J0005-00 [I,A]; C09J0005-00 [I,C*]; H01L0021-02 [N,C*]; H01L0021-304 [N,A] FTERM 4J040/NA20; 4J040/PA30; 4J040/PA32; 4J040/PA42	

AB The release method, useful for semiconductor device fabrication, includes irradiating laminates comprising substrates, cured adhesives, and adherends with light so as to decompose the adhesives, heating/expanding the adhesives, dissolving in solvents and/or alkaline aqueous solns., and separating the adherends from the substrates. Thus, a laminate comprising a Si wafer, a cured adhesive layer containing tert-Bu methacrylate homopolymer, 2,5-dimethyl-2,5-hexanediol diacrylate, photopolymer. initiator, and 4,7-di-n-butoxynaphthyltetrahydrothiophenium trifluoromethanesulfonate, and quartz glass was irradiated with light from the glass surface, heated at 120° for 5 min, and dipped in acetone for 5 min to give a recovered Si wafer and quartz glass.

ST adhesive removal semiconductor device fabrication; photoacid generator polyacrylate adhesive removal expansion

IT Expansion
 Semiconductor device fabrication
 (efficient release method of adherends without damaging nor residual adhesive)

IT 25189-00-8F, tert-Butyl methacrylate homopolymer
 RL: CPS (Chemical process); IMF (Industrial manufacture); PEP (Physical, engineering or chemical process); PYP (Physical process); PREP (Preparation); PROC (Process)
 (adhesive, crosslinked; efficient release method of adherends without damaging nor residual adhesive)

IT 188837-15-2
 RL: RCT (Reactant); RACT (Reactant or reagent)
 (crosslinker for adhesives; efficient release method of adherends without damaging nor residual adhesive)

IT 67-64-1, Acetone, uses
 RL: NUU (Other use, unclassified); USES (Uses)
 (for dissolving adhesives; efficient release method of adherends without damaging nor residual adhesive)

IT 366464-82-6
 RL: MOA (Modifier or additive use); USES (Uses)
 (photoacid generator, adhesives containing; efficient release method of

FOR 10559534 by Cynthia Hamilton

adherends without damaging nor residual adhesive)
IT 60676-86-0, Quartz glass
RL: MSC (Miscellaneous)
(substrate; efficient release method of adherends without damaging nor residual adhesive)
IT 7440-21-3, Silicon, miscellaneous
RL: MSC (Miscellaneous)
(wafer, adherend; efficient release method of adherends without damaging nor residual adhesive)

L7 ANSWER 8 OF 14 CA COPYRIGHT 2008 ACS on STN

AN 143:315452 CA

ED Entered STN: 13 Oct 2005

TI Resist polymer, resist composition, process for pattern formation, and

starting compounds for production of the resist polymer

IN Momose, Hikaru; Ootake, Atsushi; Nakamura, Tadashi; Ueda, Akifumi

PA Mitsubishi Rayon Co., Ltd., Japan

SO PCT Int. Appl., 178 pp.

CODEN: PIXXD2

DT Patent

LA Japanese

IC ICM C08F020-10

ICS G03F007-039; H01L021-027

CC 74-5 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)

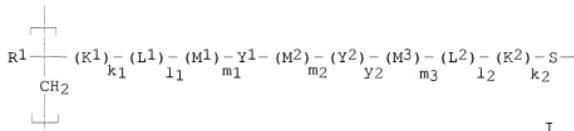
Section cross-reference(s): 35, 38

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	WO 2005085301	A1	20050915	WO 2005-JP4402	20050308
	W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SM, SY, TJ, TM, TN, TR, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM,				
ZW	RW: BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LT, LU, MC, NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG				
	CN 1930194	A	20070314	CN 2005-80007225	20050308
	US 2007190449	A1	20070816	US 2006-592057	20060908
	KR 785585	B1	20071213	KR 2006-720723	20061002
	JP 2007327062	A	20071220	JP 2007-160177	20070618
PRAI	JP 2004-63616	A	20040308		
	JP 2004-73183	A	20040315		
	JP 2004-189889	A	20040628		
	JP 2004-220036	A	20040728		
	JP 2004-253002	A	20040831		
	JP 2004-376738	A	20041227		
	JP 2005-4315	A	20050111		
	JP 2006-519412	A3	20050308		
	WO 2005-JP4402	W	20050308		

CLASS	PATENT NO.	CLASS	PATENT FAMILY CLASSIFICATION CODES
WO 2005085301		ICM	C08F020-10
		ICS	G03F007-039; H01L021-027
		IPCI	C08F0020-10 [ICM,7]; C08F0020-00 [ICM,7,C*]; G03F0007-039 [ICS,7]; H01L0021-027 [ICS,7];
H01L0021-02			[ICS,7,C*]
		IPCR	C08F0020-00 [I,C*]; C08F0020-10 [I,A]; G03F0007-039 [I,C*]; G03F0007-039 [I,A]; H01L0021-027 [I,C*]; H01L0021-027 [I,A]
CN 1930194		IPCI	C08F0020-10 [I,A]; C08F0020-00 [I,C*]; G03F0007-039 [I,A]; H01L0021-027 [I,A]; H01L0021-02 [I,C*]
US 2007190449		IPCR	C08F0020-00 [I,C]; C08F0020-10 [I,A]
US 2007190449		IPCI	G03C0001-00 [I,A]
KR 785585		NCL	430/270.100
JP 2007327062		IPCI	G03F0007-039 [I,A]; C08F0020-00 [I,A]; C08F0020-10 [I,A]; H01L0021-027 [I,A]; H01L0021-02 [I,C*]
		IPCI	C08F0220-10 [I,A]; C08F0220-00 [I,C*]; G03F0007-039 [I,A]; H01L0021-027 [I,A]; H01L0021-02 [I,C*]
		FTERM	2H025/AA02; 2H025/AA04; 2H025/AB16; 2H025/AC04; 2H025/AC06; 2H025/AC08; 2H025/AD03; 2H025/BE07; 2H025/BF02; 2H025/BG00; 2H025/FA12; 2H025/FA17; 4J100/AL08P; 4J100/AL08Q; 4J100/BA03P; 4J100/BA03Q; 4J100/BA04P; 4J100/BA40P; 4J100/BA40Q; 4J100/BB13Q; 4J100/BB18P; 4J100/BC04P; 4J100/BC04Q; 4J100/BC07P; 4J100/BC08P; 4J100/BC09P; 4J100/BC09Q; 4J100/BC12P; 4J100/BC12Q; 4J100/BC53P; 4J100/BC53Q; 4J100/CA04; 4J100/CA05; 4J100/DA01; 4J100/DA05; 4J100/JA38

GI



I

AB A resist polymer which exhibits reduced line edge roughness and little causes defects in DUV excimer laser lithog. or the like and which contains as the structural unit an acid-dissociable unit represented by the general formula J{{(K1)k1}n{(L1)l1}n{(M1)m1}n{Y}n{(M2)m2}n{(L2)l2}n{(K2)k2}n[E]n}n or I (n = 2-24; J = n-valent hydrocarbon group optionally bearing a substituent and/or a heteroatom; E = residue of polymerization inhibitor, chain

FOR 10559534 by Cynthia Hamilton

transfer agent, or polymerization initiator; K1 and K2 = at least one kind of groups selected from among alkylene, cycloalkylene, oxyalkylene, arylene, etc.; L1 and L2 = at least one kind of groups selected from among -C(O)O-, -C(O)-, and -OC(O)-; M1, M2, and M3 = at least one kind of groups selected from among alkylene, cycloalkylene, oxyalkylene, and arylene; Y, Y1, and Y2 = acid-dissociable linkage; k1, k2, l1, l2, m1, m2, and m3 = 0 or 1; and R1 is H or Me).

ST resist DUV photoresist polymer compn telomer chain transfer agent

IT Photolithography

Photoresists

(UV; preparation of telomer for synthesis of DUV photoresist polymer)

IT Chain transfer agents

Resists

(preparation of telomer for synthesis of DUV photoresist polymer)

IT 864679-84-5P

RL: CAT (Catalyst use); IMF (Industrial manufacture); PREP (Preparation); USES (Uses)

(chain transfer agent; preparation of chain transfer agent)

IT 864679-76-5P

RL: CAT (Catalyst use); IMF (Industrial manufacture); PREP (Preparation); USES (Uses)

(preparation of chain transfer agent)

IT 10553-78-3 188837-15-2 864679-86-7

RL: RCT (Reactant); RACT (Reactant or reagent)

(preparation of chain transfer agent)

IT 864679-77-6P 864679-79-8P 864679-81-2P 864683-70-5P 864683-73-8P

864683-74-9P 864683-76-1P 864683-78-3P

RL: IMF (Industrial manufacture); PRP (Properties); RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent)

(preparation of telomer for synthesis of DUV photoresist polymer)

RE.CNT 16 THERE ARE 16 CITED REFERENCES AVAILABLE FOR THIS RECORD

RE

(1) E I Du Pont de Nemours & Co; EP 1479700 A1 2004 CA

(2) E I Du Pont de Nemours & Co; US 2004242798 A1 2004 CA

(3) E I Du Pont de Nemours & Co; US 2004248039 A1 2004 CA

(4) E I Du Pont de Nemours & Co; JP 2004352989 A 2004 CA

(5) Fuji Photo Film Co Ltd; JP 2002303984 A 2002 CA

(6) Fuji Photo Film Co Ltd; JP 200272481 A 2002

(7) Fuji Photo Film Co Ltd; JP 2003344994 A 2003 CA

(8) Nedolya, N; Zhurnal Organicheskoi Khimii 1987, V23(6), P1181 CA

(9) Rohm And Haas Electronic Materials L L C; EP 1403708 A2 2004 CA

(10) Rohm And Haas Electronic Materials L L C; CN 1514302 A 2004 CA

(11) Rohm And Haas Electronic Materials L L C; KR 2004030341 A 2004 CA

(12) Rohm And Haas Electronic Materials L L C; US 2004063030 A1 2004 CA

(13) Rohm And Haas Electronic Materials L L C; JP 2004151691 A 2004 CA

(14) Sumitomo Chemical Co Ltd; JP 2003280200 A 2003 CA

(15) Toray Industries Inc; JP 2003342306 A 2003 CA

(16) Tsuaitowanfarengonyejishuuyjenjiouyuen; JP 03-134669 A 1991 CA

L7 ANSWER 9 OF 14 CA COPYRIGHT 2008 ACS on STN

AN 142:82296 CA

ED Entered STN: 20 Jan 2005

FOR 10559534 by Cynthia Hamilton

TI Photocurable resin composition
IN Yashiro, Takao; Tataro, Ryoji; Tanabe, Takayoshi
PA DSM IP Assets B.V., Neth.
SO PCT Int. Appl., 40 pp.
CODEN: PIXXD2
DT Patent
LA English
IC ICM G03F007-038
ICS G03F007-00
CC 74-5 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)
Section cross-reference(s): 38

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	WO 2004111733	A1	20041223	WO 2004-NL423	20040614
	W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW				
	RW: BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG				
	JP 2005053936	A	20050303	JP 2003-175755	20030620
	EP 1634124	A1	20060315	EP 2004-748655	20040614
	R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, FI, RO, CY, TR, BG, CZ, EE, EU, PL, SK				
	CN 1806208	A	20060719	CN 2004-80016383	20040614
	US 2007004816	A1	20070104	US 2005-559534	20051202
PRAI	JP 2003-169206	A	20030613		
	JP 2003-175755	A	20030620		
	WO 2004-NL423	W	20040614		

CLASS

	PATENT NO.	CLASS	PATENT FAMILY CLASSIFICATION CODES
WO 2004111733	ICM	G03F007-038	
	ICS	G03F007-00	
	IPCI	G03F007-038 [ICM, 7]; G03F0007-00 [ICS, 7]	
	IPCR	C08L0021-00 [I,C*]; C08L0021-00 [I,A]; C08G0059-00 [I,C*]; C08G0059-40 [I,A]; C08L0063-00 [I,C*]; C08L0063-00 [I,A]; G03F0007-00 [I,C*]; G03F0007-027 [N,C*]; G03F0007-027 [N,A]; G03F0007-038 [I,C*]; G03F0007-038 [I,A]; ECLA G03F007/00S, G03F007/038	
JP 2005053936	IPCI	C08G0059-40 [ICM, 7]; C08G0059-00 [ICM, 7,C*]; C08L0021-00 [ICS, 7]; C08L0063-00 [ICS, 7]	
	IPCR	G03F0007-00 [I,A]; G03F0007-00 [I,C*]; G03F0007-027 [N,A]; G03F0007-027 [N,C*]; G03F0007-038 [I,A]; G03F0007-038 [I,C*]	
	FTERM	4J002/A002; 4J002/CD081; 4J002/CD101; 4J002/EB106; 4J002/EV296; 4J002/FD146; 4J036/AA02; 4J036/AB01;	

4J036/AD08; 4J036/AD09; 4J036/AG04; 4J036/AG06;
4J036/AJ05; 4J036/AJ08; 4J036/AK19; 4J036/DC15;
4J036/DC16; 4J036/EA03; 4J036/FB03

EP 1634124 IPCI G03F0007-038 [ICM, 7]; G03F0007-00 [ICS, 7]
IPCR C08L0021-00 [I,C*]; C08L0021-00 [I,A]; C08G0059-00
[I,C*]; C08G0059-40 [I,A]; C08L0063-00 [I,C*];
C08L0063-00 [I,A]; G03F0007-00 [I,C*]; G03F0007-00
[I,A]; G03F0007-027 [N,C*]; G03F0007-027 [N,A];
G03F0007-038 [I,C*]; G03F0007-038 [I,A]

ECLA G03F0007/00S; G03F0007/038

CN 1806208 IPCI G03F0007-038 [I,A]; G03F0007-00 [I,A]
US 2007004816 IPCI C08F0002-46 [I,A]
NCL 522/069.000
ECLA G03F0007/00S; G03F0007/038

AB The present invention relates to a liquid photocurable resin composition exhibiting high photocurability and producing cured products with superior mech. strength and fracture toughness, and particularly to a photocurable resin composition useful as a resin composition for three-dimensional photofabrication of a cured product for which impact resistance is particularly required. The photocurable composition comprises (A) a component comprising a carboxyl group that may dissociate in the presence of an acid,
(B) a cationically polymerizable compound, and (C) a cationic photoinitiator.

ST photocurable resin compn 3D photofabrication impact resistance mech strength

IT Synthetic rubber, uses
RL: TEM (Technical or engineered material use); USES (Uses)
(Resinous Bond RKB; photocurable resin composition)

IT Coating materials
(photocurable; photocurable resin composition)

IT 109856-58-8, UVI 1697
RL: TEM (Technical or engineered material use); USES (Uses)
(UVI 1697; photocurable resin composition)

IT 1663-39-4DP, polymers with epoxy resins and acrylic resins and synthetic rubber 1675-54-3DP, polymers with epoxy resins and acrylic resins and synthetic rubber 2386-87-0DP, polymers with epoxy resins and acrylic resins and synthetic rubber 3130-19-6DP, polymers with epoxy resins and acrylic resins and synthetic rubber 25723-16-4DP, polymers with epoxy resins and acrylic resins and synthetic rubber 29570-58-9DP, polymers with epoxy resins and acrylic resins and synthetic rubber 88465-93-4DP, polymers with epoxy resins and acrylic resins and synthetic rubber 168274-87-1DP, polymers with epoxy resins and acrylic resins and synthetic rubber 188837-15-2DP, polymers with epoxy resins and acrylic resins and synthetic rubber

RL: PNU (Preparation, unclassified); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)
(photocurable resin composition)

IT 79-10-7, Acrylic acid, uses 947-19-3, Irgacure 184 1663-39-4,
tert-Butyl acrylate 25068-38-6 25085-98-7, UVR 6110 25791-96-2,
Sannix GP 400 29570-58-9, Dipentaerythritol hexaacrylate 68924-34-5,
UVR 6199 88465-93-4 121915-68-2, Viscoat 2100 168274-87-1,

FOR 10559534 by Cynthia Hamilton

tert-Butyl acrylate-p-isopropenyl phenol copolymer 188837-15-2
RL: TEM (Technical or engineered material use); USES (Uses)
(photocurable resin composition)

RE.CNT 6 THERE ARE 6 CITED REFERENCES AVAILABLE FOR THIS RECORD
RE

- (1) Asahi Denka Kogyo Kabushiki Kaisha; EP 0831127 A 1998 CA
- (2) Canon K K; DE 3621477 A 1987 CA
- (3) Imperial Chemical Industries Plc; EP 0535828 A 1993 CA
- (4) Jsr Corporation; EP 0910143 A 1999 CA
- (5) Tokyo Ohka Kogyo Co Ltd; EP 1182506 A 2002 CA
- (6) Wako Pure Chemical Industries Ltd; EP 0887706 A 1998 CA

L7 ANSWER 10 OF 14 CA COPYRIGHT 2008 ACS on STN

AN 140:321916 CA

ED Entered STN: 06 May 2004

TI (Meth)acrylic ester and its polymer
IN Chamoto, Shigeru; Morikawa, Toshiyuki
PA Yasuhara Chemical Co., Ltd., Japan
SO Jpn. Kokai Tokkyo Koho, 12 pp.
CODEN: JKXXAF

DT Patent

LA Japanese

IC ICM C07C069-54

ICS C08F020-20; G03F007-033

CC 35-2 (Chemistry of Synthetic High Polymers)
Section cross-reference(s): 74

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	JP 2004123611	A	20040422	JP 2002-290583	20021003
PRAI	JP 2002-290583			20021003	

CLASS

	PATENT NO.	CLASS	PATENT FAMILY CLASSIFICATION CODES
JP	2004123611	ICM	C07C069-54
		ICS	C08F020-20; G03F007-033
		IPCI	C07C0069-54 [ICM, 7]; C07C0069-00 [ICM, 7,C*]; C08F0020-20 [ICS, 7]; C08F0020-00 [ICS, 7,C*]; G03F0007-033 [ICS, 7]
		IPCR	C07C0069-00 [I,C*]; C07C0069-54 [I,A]; C08F0020-00 [I,C*]; C08F0020-20 [I,A]; G03F0007-033 [I,A]; G03F0007-033 [I,C*]
		FTERM	2H025/AA02; 2H025/AA09; 2H025/AA14; 2H025/AB14; 2H025/AB16; 2H025/AB17; 2H025/AC04; 2H025/AC05; 2H025/AC06; 2H025/AD01; 2H025/AD03; 2H025/BE00; 2H025/BE10; 2H025/BG00; 2H025/CB14; 2H025/FA17; 4H006/AA01; 4H006/AA03; 4H006/AB46; 4H006/AB92; 4H006/KC14; 4H006/KE00; 4J100/AB07Q; 4J100/AJ02Q; 4J100/AK32Q; 4J100/AL00Q; 4J100/AL62P; 4J100/AL65P; 4J100/AL67P; 4J100/BA03P; 4J100/BA03Q; 4J100/BA04Q; 4J100/CA01; 4J100/CA04; 4J100/DA01; 4J100/JA01; 4J100/JA03; 4J100/JA07; 4J100/JA15; 4J100/JA32; 4J100/JA35; 4J100/JA37; 4J100/JA38; 4J100/JA44; 4J100/JA46; 4J100/JA52

AB A (meth)acrylic ester is represented by

CH2:CHX1CO2CR1R3Y(Z)CR2R4OCOCHX2:C

H₂, where R1-4 = C₁-20 alkyl, X₁, X₂ = H, Me, Y = C₀-20 linear or branched alkylene, Z = H, C₁-20 linear or branched alkyl, or a (meth)acrylic ester group of a tertiary mono- or polyalc. The monomers are useful in photoresists with high resolution, good dry etching properties, transparency and adhesion. 2,5-Dimethyl-2,5-hexanediol diacrylate was prepared from 2,5-dimethyl-2,5-hexanediol and acryloyl chloride.

ST acrylate ester manuf photoresist
 IT 188837-15-2P
 RL: IMF (Industrial manufacture); PREP (Preparation)
 ((meth)acrylic ester and its polymer)
 IT 110-03-2, 2,5-Dimethyl-2,5-hexanediol 814-68-6, Acrylic acidchloride
 RL: RCT (Reactant); RACT (Reactant or reagent)
 ((meth)acrylic ester and its polymer)

L7 ANSWER 11 OF 14 CA COPYRIGHT 2008 ACS on STN
 AN 136:310611 CA
 ED Entered STN: 09 May 2002
 TI Crosslinkable compositions, crosslinked resins, and their decrosslinked resins
 IN Watanabe, Yasuhige; Okada, Hiroshi
 PA NOF Corporation, Japan
 SO Jpn. Kokai Tokkyo Koho, 8 pp.
 CODEN: JKXXAF
 DT Patent
 LA Japanese
 IC ICM C08F220-20
 ICS C08F002-44; C08F008-10; C08F265-04
 CC 37-3 (Plastics Manufacture and Processing)

FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
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PI JP 2002121228	A	20020423	JP 2000-311213	20001011
PRAI JP 2000-311213		20001011		

CLASS

PATENT NO.	CLASS	PATENT FAMILY CLASSIFICATION CODES
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JP 2002121228	ICM C08F220-20 ICS C08F002-44; C08F008-10; C08F265-04 IPCI C08F0220-20 [ICM, 7]; C08F0220-00 [ICM, 7,C*]; C08F0002-44 [ICS, 7]; C08F0008-10 [ICS, 7]; C08F0008-00 [ICS, 7,C*]; C08F0265-04 [ICS, 7]; C08F0265-00 [ICS, 7,C*]	
	IPCR C08F0002-44 [I,C*]; C08F0002-44 [I,A]; C08F0008-00 [I,C*]; C08F0008-10 [I,A]; C08F0220-00 [I,C*]; C08F0220-20 [I,A]; C08F0265-00 [I,C*]; C08F0265-04 [I,A]	

AB The compns. for giving recyclable resins, comprise vinyl monomers, di(meth)acrylates shown as RIC(:CH₂)CO₂CMe₂R₂OC(:CH₂)R₁ (R₁ = H, Me; R₂ = C₁-11 linear or branched alkylene), and radical generators. The crosslinked resins are obtained by polymerizing and crosslinking the compns., and the decrosslinked resins are obtained by heat-cutting the tertiary alc.-derived ester bonds in the crosslinked resins. Thus, a mixture containing

FOR 10559534 by Cynthia Hamilton

styrene, 2,5-dimethyl-2,5-hexanediol dimethacrylate, and
tert-butyldiisobutyrate was polymerized to give a crosslinked polymer
showing
solubility in benzene \leq 10% initially and 100% after heating.
ST diacrylate crosslinked vinyl resin decrosslinking recycling;
dimethacrylate crosslinked vinyl resin decrosslinking recycling; styrene
dimethylhexanediol dimethacrylate crosslinked polymer heating
decrosslinking
IT Recycling of plastics and rubbers
(crosslinkable vinyl monomer-di(meth)acrylate compns. for resins and
their decrosslinked resins for recycling)
IT Polymer degradation
(thermal; crosslinkable vinyl monomer-di(meth)acrylate compns. for
resins and their decrosslinked resins for recycling)
IT 382137-73-7P 410070-19-8P, 2,5-Dimethyl-2,5-hexanediol
dimethacrylate-styrene copolymer 410070-20-1P 410070-21-2P
410070-22-3P
RL: CPS (Chemical process); IMF (Industrial manufacture); PEP (Physical,
engineering or chemical process); PREP (Preparation); PROC (Process)
(crosslinkable vinyl monomer-di(meth)acrylate compns. for resins and
their decrosslinked resins for recycling)
IT 131787-39-8P 188837-15-2P 382137-71-5P
RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation);
RACT
(Reactant or reagent)
(crosslinkable vinyl monomer-di(meth)acrylate compns. for resins and
their decrosslinked resins for recycling)

L7 ANSWER 12 OF 14 CA COPYRIGHT 2008 ACS on STN
AN 130:66874 CA
ED Entered STN: 30 Jan 1999
TI Synthesis and Characterization of Thermally Degradable Polymer Networks
AU Ogino, Kenji; Chen, Jir-Shyr; Ober, Christopher K.
CS Department of Materials Science and Engineering, Cornell University,
Ithaca, NY, 14853-1501, USA
SO Chemistry of Materials (1998), 10(12), 3833-3838
CODEN: CMATEX; ISSN: 0897-4756
PB American Chemical Society
DT Journal
LA English
CC 35-8 (Chemistry of Synthetic High Polymers)
AB Diacrylate and dimethacrylate monomers containing thermally cleavable
tertiary
ester linkages with different spacer lengths were synthesized and
studied.
The photopolymn. rates of diacrylates were much higher than those of
dimethacrylates. Glass transition temps. of the prepared networks
increased
from 55 to 180° with decreasing spacer length for diacrylate
networks and from 135 to 185° for dimethacrylate networks.
Isothermal thermogravimetric analyses revealed that the networks were
thermally stable up to 150° and decomposed smoothly between 180 and
200° to form partially dehydrated linear poly(acrylic acid) or
poly(methacrylic acid). It was observed that decomposition reactions
were

dominated by an autoacceleration mechanism. The rate of weight loss was dependent on temperature and spacer length. Although the resulting decomposition

products were insol. in DMF, methanol, and water due to the presence of intermol. acid anhydride groups, they were completely soluble in dilute sodium

hydroxide and ammonium hydroxide solution accompanied by hydrolysis or ammonolysis of the anhydride units.

ST thermally degradable polyacrylate network; thermal degrdn tertiary diol diacrylate polymer; methacrylate tertiary diol polymer degrdn; kinetics thermal degrdn polyacrylate network; mechanism thermal degrdn polyacrylate network

IT Polymerization
(photopolylmn.; in preparation of thermally degradable tertiary diol di(meth)acrylate networks)

IT Glass transition temperature
(preparation and thermal degradation mechanism and kinetics of tertiary diol
di(meth)acrylate networks)

IT Polymer degradation
(thermal, mechanism; preparation and thermal degradation mechanism and kinetics
of tertiary diol di(meth)acrylate networks)

IT Polymer degradation kinetics
(thermal; preparation and thermal degradation mechanism and kinetics
of tertiary
diol di(meth)acrylate networks)

IT 19781-07-8P, 2,7-Dimethyl-2,7-octanediol
RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
(Reactant or reagent)
(monomer intermediate; preparation and thermal degradation mechanism
and

kinetics of tertiary diol di(meth)acrylate networks)
IT 75-16-1, Methylmagnesium bromide 110-03-2, 2,5-Dimethyl-2,5-hexanediol
627-93-0, Dimethyl adipate 814-68-6, Acryloyl chloride 920-46-7,
Methacryloyl chloride 22092-57-5, 2,9-Dimethyl-2,9-decanediol
22092-59-7, 2,11-Dimethyl-2,11-dodecanediol
RL: RCT (Reactant); RACT (Reactant or reagent)
(monomer starting material; preparation and thermal degradation
mechanism and

kinetics of tertiary diol di(meth)acrylate networks)
IT 131787-39-8P 188837-15-2P 217973-62-1P 217973-86-9P
217973-93-8P 217974-02-2P 217974-11-3P 217974-18-0P
RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
(Reactant or reagent)
(monomer; preparation and thermal degradation mechanism and kinetics
of tertiary
diol di(meth)acrylate networks)

IT 217974-25-9P 217974-33-9P 217974-38-4P 217974-43-1P 217974-47-5P
217974-54-4P 217974-61-3P 217974-67-9P
RL: PEP (Physical, engineering or chemical process); PRP (Properties);

SPN
(Synthetic preparation); PREP (Preparation); PROC (Process)
(preparation and thermal degradation mechanism and kinetics of
tertiary diol

FOR 10559534 by Cynthia Hamilton

di(meth)acrylate networks)
RE.CNT 13 THERE ARE 13 CITED REFERENCES AVAILABLE FOR THIS RECORD
RE

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(4) Grant, D; Polymer 1960, V1, P445 CA
(5) Jager, W; Macromolecules 1997, V30, P780 CA
(6) Kloosterboer, J; Adv Polym Sci 1988, V84, P1 CA
(7) Kloosterboer, J; Polymer 1990, V31, P95 CA
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(9) Sastri, V; J Appl Polym Sci 1990, V39, P1439 CA
(10) Schaeffgen, J; J Polym Sci 1962, V58, P1049 CA
(11) Tesoro, G; J Appl Polym Sci 1990, V39, P1425 CA
(12) Torres-Filho, A; Adv Polym Sci 1994, V51, P931 CA
(13) Yang, S; Chem Mater 1998, V10(6), P1475 CA

L7 ANSWER 13 OF 14 CA COPYRIGHT 2008 ACS on STN

AN 127:313184 CA

ED Entered STN: 25 Nov 1997

TI Low voltage drive liquid crystal device and its manufacture

IN Ogawa, Hiroshi; Kuriyama, Takeshi

PA Dainippon Ink and Chemicals, Inc., Japan

SO Jpn. Kokai Tokkyo Koho, 11 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

IC ICM G02F001-1333

ICS C09K019-54

CC 74-13 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	JP 09251157	A	19970922	JP 1996-59092	19960315
	JP 3669456	B2	20050706		
PRAI	JP 1996-59092		19960315		

CLASS

	PATENT NO.	CLASS	PATENT FAMILY CLASSIFICATION CODES
	JP 09251157	ICM	G02F001-1333
		ICS	C09K019-54
		IPCI	G02F0001-1333 [ICM,6]; C09K0019-54 [ICS,6]
		IPCR	G02F0001-13 [I,C*]; G02F0001-1333 [I,A]; C09K0019-54 [I,C*]; C09K0019-54 [I,A]; G02F0001-1334 [I,A]

AB In the title device comprising a light-adjusting layer, comprised of liquid

crystals and transparent polymers, interposed between a pair of electrode-bearing transparent substrates, the light-adjusting layer is formed by irradiating alkylene di(meth)acrylates, which have alkyl sidechains at α -positions and C3-18 alkylene principal chains. The device is drivable with less than 10 V and can be used as a projection apparatus

ST liq crystal display projection app

IT Projection apparatus

FOR 10559534 by Cynthia Hamilton

(electrooptical, liquid-crystal; low voltage drive liquid crystal device
and its manufacture)
IT Liquid crystal displays
(low voltage drive liquid crystal device and its manufacture)
IT 14201-97-9, Octadecane-1,12-diol diacrylate 62722-22-9, Aronix M 120
93804-11-6, Light ester IMA 197504-20-4
RL: DEV (Device component use); USES (Uses)
(light adjusting layer of low voltage drive liquid crystal device)
IT 188837-15-2P 197504-13-5P 197504-14-6P 197504-15-7P
197504-16-8P
RL: DEV (Device component use); SPN (Synthetic preparation); PREP
(Preparation); USES (Uses)
(light adjusting layer of low voltage drive liquid crystal device)
IT 79-10-7, 2-Propenoic acid, reactions 110-03-2 144-19-4
RL: RCT (Reactant); RACT (Reactant or reagent)
(light adjusting layer of low voltage drive liquid crystal device)

L7 ANSWER 14 OF 14 CA COPYRIGHT 2008 ACS on STN

AN 126:264464 CA

ED Entered STN: 13 May 1997

TI Preparation of tertiary-alkyl acrylates from acrylic chloride in
hydrocarbon solvents

IN Nakamura, Tetsuya; Amaya, Naoyuki

PA Nippon Oils & Fats Co Ltd, Japan

SO Jpn. Kokai Tokkyo Koho, 7 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

IC ICM C07C067-14

ICS B01J031-02; C07C069-54; C07B061-00

CC 35-2 (Chemistry of Synthetic High Polymers)

FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI JP 09052864	A	19970225	JP 1995-203569	19950809

PRAI JP 1995-203569

CLASS

PATENT NO.	CLASS	PATENT FAMILY CLASSIFICATION CODES
JP 09052864	ICM	C07C067-14
	ICS	B01J031-02; C07C069-54; C07B061-00
	IPCI	C07C0067-14 [ICM,6]; B01J0031-02 [ICS,6]; C07C0069-54 [ICS,6]; C07B0061-00 [ICS,6]
	IPCR	B01J0031-02 [I,C*]; B01J0031-02 [I,A]; C07B0061-00 [I,C*]; C07B0061-00 [I,A]; C07C0067-00 [I,C*]; C07C0067-14 [I,A]; C07C0069-00 [I,C*]; C07C0069-54 [I,A]

OS MARPAT 126:264464

AB CH2:CHCO2CR1R2R3 (R1-3 = C1-12 substituents; total number of C atoms of

R1-3 = 5-21) are prepared in high yield by treatment of acrylic chloride (I)

with

R2R2R3COH in hydrocarbon solvents. A mixture of Et3N and toluene was
added

FOR 10559534 by Cynthia Hamilton

to a mixture of toluene, I, and 2,5-dimethyl-2,5-hexanediol at 50° over 1 h and the reaction mixture was further stirred at 50° for 2 h to give 85% 2,5-di methyl-2,5-hexanediol diacrylate, vs. 7% for a control using acetone instead of toluene.

ST tert alkyl acrylate prep; acrylic chloride esterification tertiary alc; hydrocarbon solvent tertiary alc acryloylation

IT Hydrocarbons, uses

RL: NUU (Other use, unclassified); USES (Uses)
(preparation of tert-alkyl acrylates from acrylic chloride and tertiary
alcs. in hydrocarbon solvents)

IT 106-63-8P 13141-04-3P 188837-15-2P 188837-16-3P
188837-17-4P

RL: IMF (Industrial manufacture); SPN (Synthetic preparation); PREP
(Preparation)
(preparation of tert-alkyl acrylates from acrylic chloride and tertiary
alcs. in hydrocarbon solvents)

IT 71-43-2, Benzene, uses 96-14-0, 3-Methylpentane 108-88-3, Toluene,
uses 110-54-3, Hexane, uses 110-82-7, Cyclohexane, uses 111-65-9,
Octane, uses 142-82-5, Heptane, uses 592-43-8, 2-Hexene 1330-20-7,
Xylene, uses

RL: NUU (Other use, unclassified); USES (Uses)
(preparation of tert-alkyl acrylates from acrylic chloride and tertiary
alcs. in hydrocarbon solvents)

IT 75-65-0, 2-Methyl-2-propanol, reactions 76-09-5, 2,3-Dimethyl-2,3-
butanediol 107-41-5, 2-Methyl-2,4-pentanediol 110-03-2,
2,5-Dimethyl-2,5-hexanediol 625-23-0, 2-Methyl-2-hexanol 814-68-6,
2-Propenoyl chloride

RL: RCT (Reactant); RACT (Reactant or reagent)
(preparation of tert-alkyl acrylates from acrylic chloride and tertiary
alcs. in hydrocarbon solvents)

=> d his

(FILE 'HOME' ENTERED AT 20:31:49 ON 14 MAR 2008)

FILE 'REGISTRY' ENTERED AT 20:32:00 ON 14 MAR 2008

L1 0 S DIMETHYLHEXANE AND DIMETHACRYLATE
L2 2 S DIMETHYLHEXANE AND DIACRYLATE
L3 1 S 188837-15-2
L4 122 S DIMETHYL AND HEXANEDIOL AND DIACRYLATE
L5 122 S DIMETHYL AND HEXANEDIOL AND DIACRYLATE
L6 2 S C14H22O4/MF AND DIACRYLATE

FILE 'CA' ENTERED AT 20:35:52 ON 14 MAR 2008

L7 14 S L3

=> s CO and acrylate and photoacid
879673 CO
194728 ACRYLATE
4104 PHOTOACID

FOR 10559534 by Cynthia Hamilton

L8 32 CO AND ACRYLATE AND PHOTOACID

=> d 32

L8 ANSWER 32 OF 32 CA COPYRIGHT 2008 ACS on STN
AN 130:160483 CA
TI Dissolution characteristics of chemically amplified 193 nm resists
AU Itani, Toshiro; Yoshino, Hiroshi; Hashimoto, Shuichi; Yamana, Mitsuharu;
Miyasaka, Mami; Tanabe, Hiroyoshi
CS NEC Corporation, Sagamihara, Kanagawa, 229-1198, Japan
SO Journal of Vacuum Science & Technology, B: Microelectronics and Nanometer
Structures (1998), 16(6), 3726-3729
CODEN: JVTBDD; ISSN: 0734-211X
PB American Institute of Physics
DT Journal
LA English
RE.CNT 14 THERE ARE 14 CITED REFERENCES AVAILABLE FOR THIS RECORD
ALL CITATIONS AVAILABLE IN THE RE FORMAT

=> d all 1-32

L8 ANSWER 1 OF 32 CA COPYRIGHT 2008 ACS on STN
AN 147:436726 CA
ED Entered STN: 01 Nov 2007
TI Exposure dose effects on the reaction-diffusion process in model extreme
ultraviolet photoresists
AU Lavery, Kristopher A.; Vogt, Bryan D.; Prabhu, Vivek M.; Lin, Eric K.;
Wu, Wen-Li; Satija, Sushil K.; Choi, Kwang-Woo
CS Polymers Division, National Institute of Standards and Technology,
Gaithersburg, MD, 20899-8541, USA
SO Journal of Vacuum Science & Technology, B: Microelectronics and Nanometer
Structures--Processing, Measurement, and Phenomena (2006), 24(6),
3044-3047
CODEN: JVSTBM; ISSN: 1071-1023
PB American Institute of Physics
DT Journal
LA English
CC 74-5 (Radiation Chemistry, Photochemistry, and Photographic and Other
Reproductive Processes)
AB The effect of exposure dose on the latent image deprotection profile in a
model extreme UV (EUV) photoresist polymer, poly(hydroxystyrene-co
-d9-tert-Bu acrylate), is measured with neutron reflectometry.
As the photoacid concentration is increased by exposure dose, the
spatial extent of propagation increases but eventually becomes
self-limited by the products of the reaction. A long-range deprotection
path occurs with diffusion length between 10 and 100 Å, while an
addnl. subnanometer short-range deprotection length scale proceeds
monotonically with dose. These measurements show that the
photoacid diffusion length into unexposed regions of a photoresist
is limited even in the absence of base quencher additives. These
fundamental data can be used to highlight materials effects on
photoresist
processing and to improve quant. models for EUV photoresists needed at
the

FOR 10559534 by Cynthia Hamilton

sub-32-nm half pitch lithog.
ST exposure dose diffusion model extreme UV photoresist
IT Photoresists
(UV, EUV; exposure dose effects on reaction-diffusion process in model extreme UV photoresists)
IT Diffusion
(exposure dose effects on reaction-diffusion process in model extreme UV photoresists)
IT Photolithography
(sub-32 nm half-pitch; exposure dose effects on reaction-diffusion process in model extreme UV photoresists)
IT 867296-29-5 928163-01-3
RL: POF (Polymer in formulation); TEM (Technical or engineered material use); USES (Uses)
(exposure dose effects on reaction-diffusion process in model extreme UV photoresists)
IT 144317-44-2, Triphenylsulfonium perfluorobutanesulfonate
RL: RGT (Reagent); TEM (Technical or engineered material use); RACT (Reactant or reagent); USES (Uses)
(exposure dose effects on reaction-diffusion process in model extreme UV photoresists)
IT 108-65-6
RL: TEM (Technical or engineered material use); USES (Uses)
(exposure dose effects on reaction-diffusion process in model extreme UV photoresists)

RE.CNT 29 THERE ARE 29 CITED REFERENCES AVAILABLE FOR THIS RECORD

RE

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FOR 10559534 by Cynthia Hamilton

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(29) Zuniga, M; Proc SPIE 1995, V2438, P113 CA

L8 ANSWER 2 OF 32 CA COPYRIGHT 2008 ACS on STN
AN 146:390493 CA
ED Entered STN: 26 Apr 2007
TI Copolymers with large maximum dissolution rate and high contrast for semiconductor lithography, their manufacture, and compositions containing them
IN Yamagishi, Takanori; Hane, Yukiko
PA Maruzen Oil Co., Ltd., Japan
SO Jpn. Kokai Tokkyo Koho, 41pp.
CODEN: JKXXAF
DT Patent
LA Japanese
CC 74-5 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)
Section cross-reference(s): 76
FAN.CNT 1

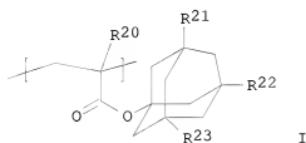
PATENT NO. KIND DATE APPLICATION NO. DATE

PI JP 2007077261 A 20070329 JP 2005-266620 20050914
PRAI JP 2005-266620 CLASS

PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES

JP 2007077261 IPCI C08F0220-28 [I,A]; C08F0220-00 [I,C*]; G03F0007-039 [I,A]; G03F0007-004 [I,A]; H01L0021-027 [I,A]; H01L0021-02 [I,C*]
IPCR C08F0220-00 [I,C]; C08F0220-28 [I,A]; G03F0007-004 [I,C]; G03F0007-004 [I,A]; G03F0007-039 [I,C]; G03F0007-039 [I,A]; H01L0021-02 [I,C]; H01L0021-027 [I,A]
FTERM 2H025/AA04; 2H025/AB16; 2H025/AC04; 2H025/AC08; 2H025/AD03; 2H025/BE00; 2H025/BE10; 2H025/BG00; 2H025/BJ01; 2H025/CB14; 2H025/CB41; 2H025/FA17; 4J100/AL03P; 4J100/AL08P; 4J100/AL08Q; 4J100/BA02P; 4J100/BA03Q; 4J100/BA15P; 4J100/BC03P; 4J100/BC04P; 4J100/BC07P; 4J100/BC08P; 4J100/BC09P; 4J100/BC09Q; 4J100/BC12P; 4J100/BC53Q; 4J100/CA04; 4J100/CA05; 4J100/CA06; 4J100/JA46

GI



AB The copolymers contain (A) repeating units having structures generating polar groups by decomposition with acids and (B) repeating units having polar groups, wherein A include $\text{CH}_2\text{CR}_{10}[\text{CO}(\text{OAlCO})_n\text{OCR}_{11}\text{R}_{12}\text{R}_{13}]$ ($\text{R}_{10} = \text{H, C1-4 hydrocarbyl; Al = C7-12 bridged alicyclic hydrocarbylene; n = 0, 1; R}_{11} = \text{H, C1-4 hydrocarbyl; when R}_{11} \text{ is H, R}_{12} \text{ is H or C1-4 hydrocarbyl, and R}_{13} \text{ is C1-12 saturated hydrocarbon-substituted oxy group; when R}_{11} \text{ is C1-4 hydrocarbyl, R}_{11} \text{ is C1-4 hydrocarbyl, and R}_{13} \text{ is C1-12 saturated hydrocarbyl; R}_{12} \text{ and R}_{13} \text{ may be bonded to form rings) and I (R}_{20} = \text{H, C1-4 hydrocarbyl; R}_{21-23} = \text{H, OH, OCHR24OR25; R}_{24} = \text{H, C1-4 hydrocarbyl; R}_{25} = \text{C1-15 saturated hydrocarbyl; R}_{24} \text{ and R}_{25} \text{ may be bonded to form rings; at least one of R}_{21-23} \text{ is OCHR24OR25). The manufacturing method includes reacting enol ethers with copolymers as above, but containing I (R}_{20} = \text{same as above; R}_{21-23} = \text{H, OH; at least one of R}_{21-23} \text{ is OH) instead of I defined as above in the presence of acids. The compns. contain the copolymers, photoacid generators, and solvents.}$

ST copolymer chem amplified pos photoresist high resoln; butyrolactonyl methyldamantyl hydroxyadamantyl acrylate polymer ethoxyethene protection; large max dissoln rate photoresist semiconductor lithog

IT Positive photoresists (chemical amplified; manufacture of copolymers for chemical amplified pos. photoresists with large maximum dissoln. rate and high contrast)

IT Ethers, reactions RL: RCT (Reactant); RACT (Reactant or reagent) (enol, protective groups; manufacture of copolymers for chemical amplified pos. photoresists with large maximum dissoln. rate and high contrast)

IT Photolithography Semiconductor materials (manufacture of copolymers for chemical amplified pos. photoresists with large maximum dissoln. rate and high contrast)

IT 75-59-2, Tetramethylammonium hydroxide RL: MSC (Miscellaneous) (developer; manufacture of copolymers for chemical amplified pos. photoresists with large maximum dissoln. rate and high contrast)

IT 468758-27-2P RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent) (manufacture of copolymers for chemical amplified pos. photoresists with large maximum dissoln. rate and high contrast)

IT 109-92-2DP, Ethyl vinyl ether, reaction products with hydroxyadamantyl-containing acrylate polymers 2182-55-0DP, Cyclohexyl vinyl ether, reaction products with hydroxyadamantyl-containing acrylate polymers

468758-27-2DP, reaction products with enol ethers
RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)
(manufacture of copolymers for chemical amplified pos. photoresists with large maximum dissoln. rate and high contrast)
IT 284474-28-8
RL: CAT (Catalyst use); TEM (Technical or engineered material use); USES (Uses)
(photoacid generator; manufacture of copolymers for chemical amplified pos. photoresists with large maximum dissoln. rate and high contrast)
IT 97-64-3, Ethyl lactate 84540-57-8, Propylene glycol methyl ether acetate
RL: TEM (Technical or engineered material use); USES (Uses)
(solvent; manufacture of copolymers for chemical amplified pos. photoresists with large maximum dissoln. rate and high contrast)

L8 ANSWER 3 OF 32 CA COPYRIGHT 2008 ACS on STN

AN 146:229775 CA

ED Entered STN: 08 Mar 2007

TI Acetal group-containing multifunctional (meth)acrylic acid esters, polymers, resist compositions, and manufacture of substrates having patterns

IN Nakamura, Masashi

PA Mitsubishi Rayon Co., Ltd., Japan

SO Jpn. Kokai Tokyo Koho, 53pp.

CODEN: JKXXAF

DT Patent

LA Japanese

CC 35-2 (Chemistry of Synthetic High Polymers)
Section cross-reference(s): 74

FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI JP 2007045924	A	20070222	JP 2005-231384	20050809
PRAI JP 2005-231384		20050809		

CLASS

PATENT NO.	CLASS	PATENT FAMILY CLASSIFICATION CODES
JP 2007045924	IPC1	C08F0020-28 [I,A]; C08F0020-00 [I,C*]; G03F0007-039 [I,A]; H01L0021-027 [I,A]; H01L0021-02 [I,C*]; C07C0069-003 [I,A]; C07C0069-54 [I,A]; C07C0069-00 [I,C*]
	IPCR	C08F0020-00 [I,C]; C08F0020-28 [I,A]; C07C0069-00 [I,C]; C07C0069-003 [I,A]; C07C0069-54 [I,A]; G03F0007-039 [I,C]; G03F0007-039 [I,A]; H01L0021-02 [I,C]; H01L0021-027 [I,A]
	FTERM	2H025/AA03; 2H025/AA04; 2H025/AA10; 2H025/AA11; 2H025/AB16; 2H025/AC04; 2H025/AC08; 2H025/AD03; 2H025/BE00; 2H025/BE10; 2H025/BG00; 2H025/CB14; 2H025/CB21; 2H025/CB41; 2H025/CB45; 2H025/FA17; 4H006/AA01; 4H006/AA03; 4H006/AB46; 4H006/KC14; 4J100/AL08P; 4J100/AL08Q; 4J100/AL08R; 4J100/AL65S;

FOR 10559534 by Cynthia Hamilton

4J100/BA02S; 4J100/BA03R; 4J100/BC09Q; 4J100/BC09R;
4J100/BC53P; 4J100/CA06; 4J100/FA19; 4J100/JA37

OS MARPAT 146:229775

AB H₂C:CR1CO₂CHR3OA(O)m(CO)n(C₂):CH₂ [R1-3 = H, Me; A = (un)substituted alkylene, (un)substituted cycloalkylene, (un)substituted oxyalkylene, (un)substituted polyoxyalkylene, (un)substituted arylene; m, n = 0-1; there are no acetal structures in OA(O)m] and their polymers are prepared Compns. comprising the resist compns. containing the polymers

and

photoacid generators are applied on substrates, irradiated with light, and developed to form patterns. Thus, 50.3 g 2-(vinyloxyethoxy)ethyl acrylate was treated with 29.2 g acrylic acid at 60°, mixed with PhMe, washed with saturated aqueous NaHCO₃ solution

twice, concentrated, and distilled to give 52.8 g

H₂C:CHCO₂CH₂CH₂CH₂CH₂COCH₃:

CH₂ showing good thermal stability.

ST thermal stability acetal acrylate polymer resist pattern

IT Photolithography

Photoresists

(acetal group-containing multifunctional (meth)acrylates with good thermal

stability for resists)

IT 924635-26-7P, 3,6-Dioxaoctane-1,7-diol diacrylate- α -methacryloyloxy- γ -butyrolactone-3-hydroxy-1-adamantyl methacrylate-2-methyl-2-adamantyl methacrylate copolymer 924635-27-8P, Cyanonorbornane methacrylate-3,6-dioxaoctane-1,7-diol dimethacrylate- α -methacryloyloxy- γ -butyrolactone-2-methyl-2-adamantyl methacrylate copolymer

RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)
(acetal group-containing multifunctional (meth)acrylates with good

thermal

stability for resists)

IT 905280-45-7P, 3,6-Dioxaoctane-1,7-diol diacrylate 924635-25-6P,
3,6-Dioxaoctane-1,7-diol dimethacrylate

RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation);

RACT

(Reactant or reagent)

(acetal group-containing multifunctional (meth)acrylates with good thermal

stability for resists)

IT 79-10-7, Acrylic acid, reactions 79-41-4, Methacrylic acid, reactions 76392-22-8, 2-(2-Vinyloxyethoxy)ethyl methacrylate 86273-46-3

RL: RCT (Reactant); RACT (Reactant or reagent)

(acetal group-containing multifunctional (meth)acrylates with good thermal

stability for resists)

L8 ANSWER 4 OF 32 CA COPYRIGHT 2008 ACS on STN

AN 146:186147 CA

ED Entered STN: 22 Feb 2007

TI Ink compositions with good colorant dispersibility, printing method using them, and their printed articles

IN Tsujibata, Shigetomo

FOR 10559534 by Cynthia Hamilton

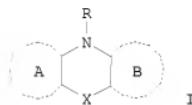
PA Fujifilm Holdings Corp., Japan
SO Jpn. Kokai Tokkyo Koho, 39pp.
CODEN: JKXXAF
DT Patent
LA Japanese
CC 42-12 (Coatings, Inks, and Related Products)
FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 2007023073	A	20070201	JP 2005-203088	20050712

CLASS

PATENT NO.	CLASS	PATENT FAMILY CLASSIFICATION CODES
JP 2007023073	IPCI	C09D0011-00 [I,A]; B41J0002-01 [I,A]; B41M0005-00 [I,A]
	IPCR	C09D0011-00 [I,C]; C09D0011-00 [I,A]; B41J0002-01 [I,C]; B41M0005-00 [I,C]; B41M0005-00 [I,A]
	FTERM	2C056/EA04; 2C056/FC02; 2C056/FD20; 2C056/HA44; 2H186/AB11; 2H186/BA08; 2H186/DA12; 2H186/FA18; 2H186/FA20; 2H186/FB04; 2H186/FB13; 2H186/FB29; 2H186/FB30; 2H186/FB31; 2H186/FB36; 2H186/FB38; 2H186/FB40; 2H186/FB41; 2H186/FB44; 2H186/FB45; 2H186/FB46; 2H186/FB54; 4J039/AD10; 4J039/AD21; 4J039/AE05; 4J039/AE07; 4J039/BC67; 4J039/BC73; 4J039/BC74; 4J039/BC79; 4J039/BE01; 4J039/BE22; 4J039/BE27; 4J039/EA04; 4J039/EA06; 4J039/EA42; 4J039/EA44; 4J039/GA24

OS MARPAT 146:186147
GI



AB The compns. contain colorants and I (R = alkyl, acyl, carbamoyl, alkoxy carbonyl, aryl, sulfonyl, sulfamoyl; X = direct bonding, alkylene, O, S, NR1, CO; R1 = H, alkyl; at least one of A and B is an aromatic ring). Thus, a composition comprising quinacridone pigment (PR 122) 5.0, heterocyclic compound prepared by reacting 9(10H)-acridone with tetraethylene glycol Bu glycidyl ether 1.5, hexanediol diacrylate 60.0, caprolactone-modified dipentaerythritol hexaacrylate (DPCA 60) 27.5, and acylphosphine oxide (Lucirin TPO-L) 5.0 parts showed volume-average particle

FOR 10559534 by Cynthia Hamilton

diameter (D50) <100 nm, good curability in irradiation of UV at 100 mJ/cm², and
no precipitate nor viscosity increase after storing at 25° or 70° for 1 mo.

ST jet printing ink compn colorant dispersibility; acridone tetraethylene glycol butyl glycidyl ether dispersant; hexanediol caprolactone modification dipentaerythritol acrylate quinacridone ink

IT Dispersing agents
Ink-jet printing
(heterocyclic dispersant-containing jet-printing ink compns. with good colorant dispersibility)

IT Epoxy resins, uses
RL: TEM (Technical or engineered material use); USES (Uses)
(heterocyclic dispersant-containing jet-printing ink compns. with good colorant dispersibility)

IT Inks
(jet-printing; heterocyclic dispersant-containing jet-printing ink compns.
with good colorant dispersibility)

IT Polymerization catalysts
(radical photochem. or photoacid generators; heterocyclic dispersant-containing jet-printing ink compns. with good colorant dispersibility)

IT 25610-58-6DP, Butyl glycidyl ether homopolymer, acridon-10-yl-terminated 921435-58-7P 921435-59-8P
RL: IMF (Industrial manufacture); MOA (Modifier or additive use); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)
(dispersant; heterocyclic dispersant-containing jet-printing ink compns.
with good colorant dispersibility)

IT 980-26-7, PR 122
RL: RCT (Reactant); TEM (Technical or engineered material use); RACT (Reactant or reagent); USES (Uses)
(for dispersants or pigment; heterocyclic dispersant-containing jet-printing ink compns. with good colorant dispersibility)

IT 578-95-0, 9(10H)-Acridone 2426-08-6, Butyl glycidyl ether 126021-43-0
RL: RCT (Reactant); RACT (Reactant or reagent)
(heterocyclic dispersant-containing jet-printing ink compns. with good colorant dispersibility)

IT 473925-57-4, Celloxide 3000-OXT 221 copolymer 786655-03-6
RL: TEM (Technical or engineered material use); USES (Uses)
(heterocyclic dispersant-containing jet-printing ink compns. with good colorant dispersibility)

IT 18393-55-0D, Triphenylsulfonium, salts 84434-11-7, Lucirin TPO-L 273400-00-3, Cyra cure UVI 6992
RL: CAT (Catalyst use); USES (Uses)
(polymerization initiator; heterocyclic dispersant-containing jet-printing ink
compns. with good colorant dispersibility)

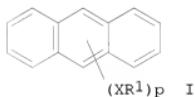
L8 ANSWER 5 OF 32 CA COPYRIGHT 2008 ACS on STN
AN 145:19039 CA
ED Entered STN: 22 Jun 2006
TI Radiation-sensitive resists, resist films and transfer films both made from same, and manufacture of electroplated electrically conductive metal

FOR 10559534 by Cynthia Hamilton

structures by using patterned resists as templates
IN Yokoyama, Kenichi; Nishikawa, Koji; Iwanaga, Shinichiro
PA Jsr Ltd., Japan
SO Jpn. Kokai Tokkyo Koho, 44 pp.
CODEN: JKXXAF
DT Patent
LA Japanese
CC 76-2 (Electric Phenomena)
Section cross-reference(s): 38, 74

FAN.CNT 1				
	PATENT NO.	KIND	DATE	APPLICATION NO.
PI	JP 2006145853	A	20060608	JP 2004-336055
PRAI	JP 2004-336055			20041119
CLASS	PATENT NO.	CLASS	PATENT FAMILY	CLASSIFICATION CODES
JP 2006145853	IPCI	G03F0007-004 [I,A]; G03F0007-039 [I,A]; G03F0007-20 [I,A]; H05K0003-18 [I,A]		
	FTERM	2H025/AA01; 2H025/AA02; 2H025/AA17; 2H025/AB11; 2H025/AB16; 2H025/AC01; 2H025/AD01; 2H025/AD03; 2H025/BE00; 2H025/BE07; 2H025/BE10; 2H025/BG00; 2H025/CA41; 2H025/CB10; 2H025/CB13; 2H025/CB14; 2H025/CB16; 2H025/CB17; 2H025/CB43; 2H025/CB45; 2H025/CC13; 2H025/CC20; 2H025/FA17; 2H025/FA35; 2H025/FA39; 2H025/FA43; 2H025/FA48; 2H097/FA02; 2H097/LA09; 5E343/AA22; 5E343/BB24; 5E343/BB38; 5E343/BB71; 5E343/CC62; 5E343/DD43; 5E343/DD56; 5E343/DD76; 5E343/EE36; 5E343/ER12; 5E343/ER18; 5E343/ER26; 5E343/GG08		

OS MARPAT 145:19039
GI



AB The resists contain (A) 0.1-20 weight parts of anthracene derivs. I [p = 1-10; R₁ = H, Cl-8 (substituted) alkyl, C₃-20 (substituted) alicyclic group, C₂-4 alkenyl, etc.; ≥2 of R₁ may form ring (containing hetero atoms); X = direct bond, O, S, CO, N(R'), etc.; R' = H, Cl-8 (substituted) alkyl, C₃-20 (substituted) alicyclic group, etc.; ≥2 of R' may form ring], (B) 0.1-20 weight parts of photoacid generators, and (C) 100 weight parts of polymers, and show sensitivity for 300-450 nm radiation. Also claimed are pos.-working above resists containing polymers bearing acid-labile groups as C. Also claimed are neg.-working

above resists containing alkali-soluble polymers as C, and crosslinking agents capable of reaction with the alkali-soluble polymers under the presence of acids. In manufacture of elec. conductive metal structures (e.g., bumps and wirings of circuits), electroplating of the metal is carried out on patterned resists used as templates. The resists, sensitive for both i-line and g-line, provide patterns with good profile.

ST UV resist anthracene sensitizer; pos UV resist anthracene sensitizer; neg UV resist anthracene sensitizer; elec circuit conductor metal electroplating UV photoresist

IT Electrodeposition
(UV resists containing anthracene sensitizers, transfer films, and electroplating of conductor metals on patterned resists)

IT Negative photoresists

Photoresists
Positive photoresists
(UV; UV resists containing anthracene sensitizers, transfer films, and electroplating of conductor metals on patterned resists)

IT Bump contacts
Interconnections, electric
(electroplating of; UV resists containing anthracene sensitizers, transfer films, and electroplating of conductor metals on patterned resists)

IT Transfers
(resist films; UV resists containing anthracene sensitizers, transfer films, and electroplating of conductor metals on patterned resists)

IT 68818-86-0, 9,10-Diethoxyanthracene 76275-14-4, 9,10-Dibutoxyanthracene
RL: MOA (Modifier or additive use); TEM (Technical or engineered material use); USES (Uses)
(UV resists containing anthracene sensitizers, transfer films, and electroplating of conductor metals on patterned resists)

IT 17464-88-9, Cymel 1174
RL: TEM (Technical or engineered material use); USES (Uses)
(crosslinking agent, neg. resist component; UV resists containing anthracene sensitizers, transfer films, and electroplating of conductor metals on patterned resists)

IT 24979-70-2, Maruka Lyncur S 2P 27029-76-1, m-Cresol-p-cresol-formaldehyde copolymer
RL: TEM (Technical or engineered material use); USES (Uses)
(neg. resist component; UV resists containing anthracene sensitizers, transfer films, and electroplating of conductor metals on patterned resists)

IT 41580-58-9, N-(Trifluoromethylsulfonyloxy)phthalimide 66003-78-9, Triphenylsulfonium trifluoromethanesulfonate 133710-62-0
RL: CAT (Catalyst use); TEM (Technical or engineered material use); USES (Uses)
(photoacid generator, resist component; UV resists containing anthracene sensitizers, transfer films, and electroplating of conductor metals on patterned resists)

IT 887704-12-3P, 2-Benzyl-2-propyl methacrylate-2-hydroxyethyl acrylate-p-hydroxyphenyl methacrylamide-isobornyl acrylate

FOR 10559534 by Cynthia Hamilton

- α -methyl-4-hydroxystyrene copolymer 887704-13-4P 887704-14-5P
887704-15-6P, Butyl acrylate-1,6-dimethacrylate
hexane-2-hydroxyethyl acrylate-2-methoxyethyl acrylate
- α -methyl-4-hydroxystyrene copolymer
RL: IMF (Industrial manufacture); TEM (Technical or engineered material
use); PREP (Preparation); USES (Uses)
(pos. resist component; UV resists containing anthracene sensitizers,
transfer films, and electroplating of conductor metals on patterned
resists)

L8 ANSWER 6 OF 32 CA COPYRIGHT 2008 ACS on STN
AN 142:178881 CA
ED Entered STN: 24 Feb 2005
TI Jet-printing ink compositions with good ink curability and bleeding
resistance, their use in image formation, and epoxy compounds
IN Nishizeki, Masato; Miura, Norio; Okubo, Kimihiko
PA Konica Minolta Medical & Graphic, Inc., Japan
SO Jpn. Kokai Tokkyo Koho, 54 pp.
CODEN: JKXXAF
DT Patent
LA Japanese
IC ICM C09D011-00
ICS B41J002-01; B41M005-00
CC 42-12 (Coatings, Inks, and Related Products)
Section cross-reference(s): 37, 74
FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	JP 2005029632	A	20050203	JP 2003-194224	20030709
PRAI	JP 2003-194224				

CLASS

PATENT NO.	CLASS	PATENT FAMILY CLASSIFICATION CODES
JP 2005029632	ICM ICS IPCI IPCR FTERM	C09D011-00 B41J002-01; B41M005-00 C09D0011-00 [ICM,7]; B41J0002-01 [ICS,7]; B41M0005-00 [ICS,7] B41J0002-01 [I,A]; B41J0002-01 [I,C*]; B41M0005-00 [I,A]; B41M0005-00 [I,C*]; C09D0011-00 [I,A]; C09D0011-00 [I,C*] 2C056/EA05; 2C056/FC02; 2C056/HA44; 2H086/BA05; 2H086/BA59; 2H086/BA60; 4J039/AB05; 4J039/BC02; 4J039/BC03; 4J039/BC17; 4J039/BC33; 4J039/BC41; 4J039/BC50; 4J039/BC52; 4J039/BC54; 4J039/BC65; 4J039/BC72; 4J039/BC73; 4J039/BC74; 4J039/BC76; 4J039/BC77; 4J039/BC79; 4J039/BE26; 4J039/CA06; 4J039/EA05; 4J039/EA10; 4J039/EA15; 4J039/EA16; 4J039/EA17; 4J039/EA18; 4J039/EA19; 4J039/EA20; 4J039/EA42; 4J039/EA47; 4J039/GA24

OS MARPAT 142:178881
GI

FOR 10559534 by Cynthia Hamilton

* STRUCTURE DIAGRAM TOO LARGE FOR DISPLAY - AVAILABLE VIA OFFLINE PRINT *

AB The compns. without showing skin sensitization contain epoxy compds. ALAr,

A(CO2)pL[(O2C)qA]r, or ACH200(OC)pL[(CO)qOCH2A]r [A = 0-2 group-substituted 4-methyl-3,4-epoxy-cyclohexane residue; r = 1-3; L = (r + 1)-valent C1-15 linking group optionally containing O or S in main chain, direct bond; p, q = 0, 1]. Alternatively, the compns. contain epoxy compds. III, IV, V, and/or VI (R103-R106 = substituent; m3, m4, m6 = 0-2; m5 = 1, 2; p3, p4 = 0, 1; L3, L4 = divalent C1-8 linking group optionally containing O or S in main chain, direct bond). The image formation is carried

out by imagewise-jetting the compns. from printing head nozzles to printing media and irradiating UV to the jet inks. The above epoxy compds. are also claimed. Thus, a magenta ink containing III (m3 = 0,

L3 = direct bond, p3 = 1), OXT 221 (di[1-ethyl(3-oxetanyl)]methyl ether), CyraCure UVI 6990 (tri-Ph sulfonium salt photoacid generator), a pigment, and an additive was jet-printed on a film and UV-irradiated to give an image showing no spread between 2 dots.

ST epoxy compd jet printing ink UV curing; ink jet printing epoxy compd oxetane photoacid generator

IT Ink-jet printing (jet-printing ink compns. containing epoxy compds. and their use in image formation by UV irradiation)

IT Epoxy resins, uses

RL: IMF (Industrial manufacture); RCT (Reactant); TEM (Technical or engineered material use); PREP (Preparation); RACT (Reactant or reagent); USES (Uses)

(jet-printing ink compns. containing epoxy compds. and their use in image formation by UV irradiation)

IT Inks

(jet-printing; jet-printing ink compns. containing epoxy compds. and their use in image formation by UV irradiation)

IT Sulfonium compounds

RL: CAT (Catalyst use); USES (Uses)

(photoacid generator, ink containing; jet-printing ink compns. containing epoxy compds. and their use in image formation by UV irradiation)

IT Inks

(photocurable; jet-printing ink compns. containing epoxy compds. and their use in image formation by UV irradiation)

IT 744910-08-5P 744910-12-1P 744910-32-5P 835611-95-5P

RL: IMF (Industrial manufacture); RCT (Reactant); TEM (Technical or engineered material use); PREP (Preparation); RACT (Reactant or reagent); USES (Uses)

(curable ink containing; jet-printing ink compns. containing epoxy compds. and their use in image formation by UV irradiation)

IT 18934-00-4, OXT 221 744910-23-4 835611-87-5 835611-91-1

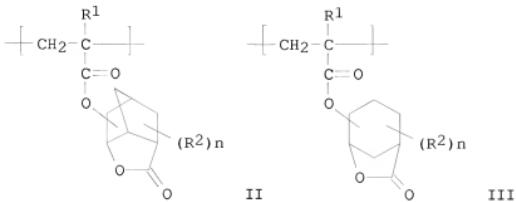
FOR 10559534 by Cynthia Hamilton

835611-97-7 835611-99-9 835612-01-6
RL: RCT (Reactant); TEM (Technical or engineered material use); RACT (Reactant or reagent); USES (Uses)
(curable ink containing; jet-printing ink compns. containing epoxy compds. and
their use in image formation by UV irradiation)
IT 835612-03-8P 835612-05-0P 835612-07-2P 835612-09-4P 835612-10-7P
835612-11-8P 835612-12-9P 835612-13-0P 835612-14-1P 835612-15-2P
RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)
(cured ink; jet-printing ink compns. containing epoxy compds. and
their use
in image formation by UV irradiation)
IT 6493-79-4P, Methyl-(4-methyl-3-cyclohexenecarboxylate) 39155-38-9P
744910-18-7P 744910-27-8P 744910-51-8P 805236-02-6P 805236-03-7P
RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation);
RACT
(Reactant or reagent)
(epoxy compound from; jet-printing ink compns. containing epoxy compds. and
their use in image formation by UV irradiation)
IT 57-55-6, 1,2-Propanediol, reactions 78-79-5, Isoprene, reactions
96-33-3, Methyl acrylate 106-89-8, Epichlorohydrin, reactions
107-02-8, Acrolein, reactions 107-21-1, Ethylene glycol, reactions
108-30-5, Succinic anhydride, reactions 126-30-7, 2,2-Dimethyl-propane-
1,3-diol
RL: RCT (Reactant); RACT (Reactant or reagent)
(epoxy compound from; jet-printing ink compns. containing epoxy compds. and
their use in image formation by UV irradiation)
IT 18393-55-0D, Triphenyl sulfonium, salts 104558-95-4, Cyra cure UVI 6990
146062-15-9 205944-57-6, Adeka Optomer SP 152 835612-17-4
835612-19-6
RL: CAT (Catalyst use); USES (Uses)
(photoacid generator, ink containing; jet-printing ink compns.
containing epoxy compds. and their use in image formation by UV
irradiation)

L8 ANSWER 7 OF 32 CA COPYRIGHT 2008 ACS on STN
AN 141:44851 CA
ED Entered STN: 08 Jul 2004
TI Sulfonium salts and their chemically amplified positive resist
compositions giving fine sharp patterns
IN Yamada, Airi; Uetani, Yasunori; Kamabuchi, Akira
PA Sumitomo Chemical Co., Ltd., Japan
SO Jpn. Kokai Tokkyo Koho, 31 pp.
CODEN: JKXXAF
DT Patent
LA Japanese
IC ICM C08F020-38
ICS C07C381-12; G03F007-004; G03F007-039; H01L021-027
CC 74-5 (Radiation Chemistry, Photochemistry, and Photographic and Other
Reproductive Processes)
FAN.CNT 1
PATENT NO. KIND DATE APPLICATION NO. DATE

FOR 10559534 by Cynthia Hamilton

PI	JP 2004162040	A	20040610	JP 2003-354759	20031015
	US 2004138353	A1	20040715	US 2003-682038	20031010
	US 7160669	B2	20070109		
	US 2007123674	A1	20070531	US 2006-643662	20061222
PRAI	JP 2002-301517	A	20021016		
	US 2003-682038	A3	20031010		
CLASS					
	PATENT NO.	CLASS	PATENT FAMILY	CLASSIFICATION CODES	
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JP 2004162040	ICM	C08F020-38			
	ICS	C07C381-12; G03F007-004; G03F007-039; H01L021-027			
	IPCI	C08F0020-38 [ICM,7]; C08F0020-00 [ICM,7,C*]; C07C0381-12 [ICS,7]; C07C0381-00 [ICS,7,C*]; G03F0007-004 [ICS,7]; G03F0007-039 [ICS,7]; H01L0021-027 [ICS,7]; H01L0021-02 [ICS,7,C*]			
	IPCR	C07C0381-00 [I,C*]; C07C0381-12 [I,A]; C08F0020-00 [I,C*]; C08F0020-38 [I,A]; G03F0007-004 [I,A]; G03F0007-004 [I,C*]; G03F0007-039 [I,A]; G03F0007-039 [I,C*]; H01L0021-02 [I,C*]; H01L0021-027 [I,A]			
	FTERM	2H025/AA01; 2H025/AA02; 2H025/AC04; 2H025/AC08; 2H025/AD03; 2H025/BE07; 2H025/BG00; 2H025/CC20; 2H025/FA03; 2H025/FA12; 2H025/FA17; 4H006/AA01; 4H006/AA03; 4H006/AB92; 4J100/AB07Q; 4J100/AK32R; 4J100/AL08P; 4J100/AL08Q; 4J100/AR11R; 4J100/BA15P; 4J100/BA15Q; 4J100/BA56P; 4J100/BA58P; 4J100/BB18P; 4J100/BC09Q; 4J100/BC43P; 4J100/BC53Q; 4J100/CA01; 4J100/CA03; 4J100/CA04; 4J100/CA05; 4J100/JA37			
US 2004138353	IPCI	G03F0007-039 [I,A]			
	IPCR	C07C0381-00 [I,C*]; C07C0381-12 [I,A]; C08K0005-00 [I,C*]; C08K0005-41 [I,A]; G03F0007-004 [I,C*]; G03F0007-004 [I,A]; G03F0007-039 [I,C*]; G03F0007-039 [I,A]			
	NCL	524/155.000			
US 2007123674	IPCI	C08F0028-02 [I,A]; C08F0028-00 [I,C*]			
	IPCR	C08F0028-00 [I,C]; C08F0028-02 [I,A]; C07C0381-00 [I,C*]; C07C0381-12 [I,A]; C08K0005-00 [I,C*]; C08K0005-41 [I,A]; G03F0007-004 [I,C*]; G03F0007-004 [I,A]; G03F0007-039 [I,C*]; G03F0007-039 [I,A]			
	NCL	526/286.000; 526/923.000			
	ECLA	G03F007/004D; C07C381/12; G03F007/039C1; G03F007/039C1S			
OS	MARPAT	141:44851			
GI					



AB The sulfonium salts are represented by the general formula Q5S03-Q1Q2S-CHQ3C02(C06CQ70)nQ4 (I; Q1, Q2 = C1-6 alkyl, C3-10 cycloalkyl; Q3, Q6, Q7 = H, Me; Q4 = hydrocarbyl bearing CO, CO2, O, and/or unsatd. bond, C3-10 cycloalkyl bearing ≥ 1 of these bond or functional group; n1 \geq 0 integer; Q5S03- = organic sulfonate ion; Q5 = C1-8 perfluoroalkyl, C1-8 alkyl, aromatic or camphor which may bear C6-12 substituent). The chemical amplified pos. resist compns. contain (A) ≥ 1 acid generators selected from sulfonium salts I and polymers derived from I and (B) resins containing mer units which bear acid-labile groups, are insol. or slightly soluble in alkalis, and become soluble in alkalis.

alkalis with acids. Preferably, the mer units of the resins B comprise 2-alkyl-2-adamantyl (meth)acrylate and/or 1-(1-adamantyl)-1-alkylalkyl (meth)acrylate. The resins B may further contain ≥ 1 mer units derived from m- or p-hydroxystyrene, 3-hydroxy-1-adamantyl (meth)acrylate, 3,5-dihydroxy-1-adamantyl (meth)acrylate, (meth)acryloyloxy- γ -butyrolactone whose lactone ring may be substituted with alkyl, and alicyclic lactones II and III (R1 = H, Me, CF3; R2 = Me, CF3; n = 1-3). The resins B may further contain mer units derived from 2-norbornene and aliphatic unsatd. dicarboxylic acid anhydride. The resist compns. may further contain triphenylsulfonium salts Ph3S+ P6S03- and/or diphenyliodonium salts Ph2I+ P7S03- (these Ph have H, OH, C1-6 alkyl, or C1-6 alkoxy; P6S03-, P7S03- = organic sulfonate ion; and amines as quenchers.

ST organic sulfonate ion, and amines as quenchers; sulfonium salt photoacid generator pos photoresist; chem amplified resist pos sulfonium salt; line edge roughness improvement pos photoresist; deep UV resist pos sulfonium salt

IT Amines, uses

RL: MOA (Modifier or additive use); USES (Uses)
(quencher; sulfonium salts and their chemical
compns. giving fine sharp patterns)

IT Positive photoresists

(sulfonium salts and their chemical amplified pos. resist compns. giving

fine sharp patterns)

IT Sulfonic compounds
RL: CAT (Catalyst use); IMF (Industrial manufacture); PREP (Preparation)
USES (Uses)
4-¹⁸ and their chemical applications.

FOR 10559534 by Cynthia Hamilton

fine sharp patterns)

IT 195000-67-0, α -Methacryloyloxy- γ -butyrolactone-2-methyl-2-adamantyl methacrylate copolymer
RL: TEM (Technical or engineered material use); USES (Uses)
(IHM 55-10K; sulfonium salts and their chemical amplified pos. resist compns. giving fine sharp patterns)

IT 177034-80-9, 4-Methylphenyldiphenylsulfonium perfluorooctanesulfonate 284474-28-8 700877-76-5
RL: CAT (Catalyst use); USES (Uses)
(photoacid generator; sulfonium salts and their chemical amplified pos. resist compns. giving fine sharp patterns)

IT 700877-74-3P 700877-75-4P
RL: CAT (Catalyst use); IMF (Industrial manufacture); PREP (Preparation); USES (Uses)
(photoacid generator; sulfonium salts and their chemical amplified pos. resist compns. giving fine sharp patterns)

IT 24544-04-5, 2,6-Diisopropylaniline
RL: MOA (Modifier or additive use); USES (Uses)
(quencher; sulfonium salts and their chemical amplified pos. resist compns. giving fine sharp patterns)

IT 258879-89-9P, 2-Ethyl-2-adamantyl methacrylate-3-hydroxy-1-adamantyl methacrylate- α -methacryloyloxy- γ -butyrolactone copolymer
RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)
(sulfonium salts and their chemical amplified pos. resist compns. giving fine sharp patterns)

giving
fine sharp patterns)

L8 ANSWER 8 OF 32 CA COPYRIGHT 2008 ACS on STN

AN 140:383120 CA

ED Entered STN: 27 May 2004

TI Positive-working photoresist compositions for far-UV microlithography

IN Sato, Kenichiro; Kodama, Kunihiko

PA Fuji Photo Film Co., Ltd., Japan

SO Jpn. Kokai Tokyo Koho, 64 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

IC ICM G03F007-039

ICS C08F002-50; C08F220-28; H01L021-027

CC 74-5 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)

FAN.CNT 1

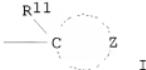
PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
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PI JP 2004138790	A	20040513	JP 2002-303088	20021017
PRAI JP 2002-303088		20021017		

CLASS

PATENT NO.	CLASS	PATENT FAMILY CLASSIFICATION CODES
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JP 2004138790	ICM ICS IPC1	G03F007-039 C08F002-50; C08F220-28; H01L021-027 G03F0007-039 [ICM,7]; C08F0002-50 [ICS,7]; C08F0002-46 [ICS,7,C*]; C08F0220-28 [ICS,7]; C08F0220-00 [ICS,7,C*]; H01L0021-027 [ICS,7]; H01L0021-02

[ICS, 7,C*]
 IPCR C08F0002-46 [I,C*]; C08F0002-50 [I,A]; C08F0220-00
 [I,C*]; C08F0220-28 [I,A]; G03F0007-039 [I,A];
 G03F0007-039 [I,C*]; H01L0021-02 [I,C*]; H01L0021-027
 [I,A]
 FTERM 2H025/AA01; 2H025/AA04; 2H025/AA09; 2H025/AA11;
 2H025/AA14; 2H025/AB16; 2H025/AC04; 2H025/AC08;
 2H025/AD03; 2H025/BE07; 2H025/BE10; 2H025/BG00;
 2H025/CB14; 2H025/CB41; 2H025/CB43; 2H025/CB45;
 4J011/QA03; 4J011/SB87; 4J011/UA01; 4J011/UA04;
 4J011/VA01; 4J011/WA01; 4J100/AL08P; 4J100/AL08Q;
 4J100/AL08R; 4J100/AL08S; 4J100/AL08T; 4J100/BA03P;
 4J100/BA03R; 4J100/BA03S; 4J100/BA12P; 4J100/BA15P;
 4J100/BA16P; 4J100/BC09Q; 4J100/BC09R; 4J100/BC12Q;
 4J100/BC53S; 4J100/BC53T; 4J100/CA04; 4J100/CA05;
 4J100/CA06; 4J100/FA19; 4J100/JA38

OS MARPAT 140:383120
GI



AB The compns., which show reduced development defects such as bridging, less
 PCD (post-coating delay) and PED (post-exposure delay), and wide process margin, and are useful for microphotofabrication with far-IR such as ArF excimer laser light, contain (A) resins which show increased alkali solubility upon action of acids and comprise (A1) $[\text{CH}_2\text{CR}_1(\text{CO}_2\text{LZ})]$ (R1 = H, alkyl; L = direct bond, alkylene, ether bond, ester bond, CO, their combination; Z = CO₂H, OH; COCH₂COR4; R4 = hydrocarbyl), (A2) $[\text{CH}_2\text{CHR}_2(\text{ACO}_2\text{ALG})]$ [R2 = H, Me; A = direct bond, linking group; ALG = I (R11 = Me, Et, Pr, CHMe₂, Bu, CH₂CHMe₂, CHMeEt; Z = atomic group forming alicyclyl); CR12R13R14 (R12-R14 = C1-4 alkyl, alicyclyl; ≥ 1 R12-R14 = alicyclyl), CR16OR15(R15, R16 = C1-4 alkyl, alicyclyl; R15 and/or R16 = alicyclyl), CR19R21CR17:CR18R20 (R17-R21 = H, C1-4 alkyl, alicyclyl; ≥ 1 R17-R21 = alicyclyl; R19 and/or R21 = C1-4 alkyl, alicyclyl), CR22R25CHR23COR24 (R22-R25 = H, C1-4 alkyl, alicyclyl; ≥ 1 of R22-R25 = alicyclyl; R23 and R24 may be bonded to form a ring)], and (A3) $[\text{CH}_2\text{CR}_3\{\text{A323}(\text{OH})\text{p}\}]$ [R3 = H, Me; A3 = direct bond, linking group; Z3 = (p + 1)-valent alicyclyl; p = 1-3], (B) R1bR2bR3bS⁺ X⁻ (R1b-R3b = alkyl; X⁻ = anion) which generate acids upon irradiation with actinic ray or radiation, and (C) solvents.
 ST far UV pos photoresist trialkylsulfonium salt photoacid generator; adamantyl acrylate copolymer trialkylsulfonium salt pos photoresist

IT Positive photoresists
(UV; far-UV pos.-working photoresist compns. containing
alkali-solubilizable resins and trialkylsulfonium salts for reduced
post-coating and post-exposure delays and wide process margin)
IT 66003-78-9 144317-44-2 258872-05-8 347193-29-7 383367-32-6
414911-52-7 677351-28-9 677351-30-3 683810-50-6
RL: CAT (Catalyst use); USES (Uses)
(far-UV pos.-working photoresist compns. containing
alkali-solubilizable
resins and trialkylsulfonium salts for reduced post-coating and
post-exposure delays and wide process margin)
IT 680223-02-3P 680223-07-8P 680223-09-0P 683809-91-8P 683810-47-1P
683810-48-2P
RL: IMF (Industrial manufacture); TEM (Technical or engineered material
use); PREP (Preparation); USES (Uses)
(far-UV pos.-working photoresist compns. containing
alkali-solubilizable
resins and trialkylsulfonium salts for reduced post-coating and
post-exposure delays and wide process margin)

L8 ANSWER 9 OF 32 CA COPYRIGHT 2008 ACS on STN
AN 140:261271 CA
ED Entered STN: 08 Apr 2004
TI High-performance 193-nm photoresist materials based on a new class of
polymers containing spaced ester functionalities
AU Khojasteh, Mahmoud; Chen, K. Rex; Kwong, Ranee W.; Lawson, Margaret C.;
Varanasi, Pushkara R.; Patel, Kaushal S.; Kobayashi, Eiichi
CS IBM Microelectronics, Hopewell Junction, 12533, USA
SO Proceedings of SPIE-The International Society for Optical Engineering
(2003), 5039(Pt. 1, Advances in Resist Technology and Processing XX),
187-194
CODEN: PSISDG; ISSN: 0277-786X
PB SPIE-The International Society for Optical Engineering
DT Journal
LA English
CC 74-5 (Radiation Chemistry, Photochemistry, and Photographic and Other
Reproductive Processes)
Section cross-reference(s): 76
AB ArF lithog. has been selected as the imaging method for the 90 nm
technol.
node. Manufacturing related issues will have to be addressed when
designing
advanced 193 nm resists that are production worthy. Post exposure bake
(PEB)
sensitivity, dissoln. properties and process window are some issues that
need continuous improvement. Initially the authors investigation focused
on a cyclic olefin (CO) platform which led to a better
understanding of the relationship between polymer structure and phys.
properties and how to improve cyclic olefin resist performance. Since
then the authors developed a new class of acrylate polymers with
pendant "spaced ester" functionality. The authors investigated the
potential use of "spaced ester" functionality on improving the lithog.
performance of CO and acrylate resist platforms. The
authors found that with "spaced ester" as pending group in CO
polymer structures, it can lower the Tg and improve the dissoln.

properties of the CO resists. Resists formulated with acrylate containing "spaced ester" group exhibit excellent PEB temperature sensitivity (1 nm/ $^{\circ}$ C), and are soluble in PGMEA. In addition, the authors demonstrated sub-100 nm resolution with excellent process window through formulation optimization for acrylate based resists. This paper will focus on the "spaced ester" based polymer design, material properties; resist characteristics, and the lithog. performance for logic dense line applications.

ST polymer design chem amplified photoresist vacuum UV lithog; cyclic olefin acrylate polymer spaced ester functionality photoresist

IT Onium compounds
RL: PRP (Properties); TEM (Technical or engineered material use); USES (Uses)
(iodonium, photoacid generator; properties and lithog. performance of 193-nm photoresists based on cyclic olefin or acrylate platforms containing spaced ester functionalities)

IT Dissolution
(kinetics; properties and lithog. performance of 193-nm photoresists based on cyclic olefin or acrylate platforms containing spaced ester functionalities)

IT Thermal decomposition
(properties and lithog. performance of 193-nm photoresists based on cyclic olefin or acrylate platforms containing spaced ester functionalities)

IT Photoresists
(vacuum-UV, chemical amplified; properties and lithog. performance of 193-nm photoresists based on cyclic olefin or acrylate platforms containing spaced ester functionalities)

IT 84540-57-8, Propylene glycol monomethyl ether acetate
RL: NUU (Other use, unclassified); USES (Uses)
(casting solvent; properties and lithog. performance of 193-nm photoresists based on cyclic olefin or acrylate platforms containing spaced ester functionalities)

IT 75-59-2, Tetramethylammonium hydroxide
RL: NUU (Other use, unclassified); USES (Uses)
(developer; properties and lithog. performance of 193-nm photoresists based on cyclic olefin or acrylate platforms containing spaced ester functionalities)

IT 195154-78-0 670249-83-9 670249-84-0 670249-85-1
RL: PRP (Properties); TEM (Technical or engineered material use); USES (Uses)
(properties and lithog. performance of 193-nm photoresists based on cyclic olefin or acrylate platforms containing spaced ester functionalities)

RE.CNT 3 THERE ARE 3 CITED REFERENCES AVAILABLE FOR THIS RECORD

RE
(1) Kumar, U; Proc SPIE 1997, V3049, P135 CA
(2) Lin, Q; Proc SPIE 2001, V4345, P78 CA
(3) Varanasi, P; Proc SPIE 1999, V3678, P51 CA

L8 ANSWER 10 OF 32 CA COPYRIGHT 2008 ACS on STN
AN 139:92748 CA
ED Entered STN: 31 Jul 2003

FOR 10559534 by Cynthia Hamilton

TI Novel sulfonyldiazomethane compounds, photoacid generators, photoresists therewith, and photolithography employing the same
IN Osawa, Yoichi; Kobayashi, Katsuhiro; Maeda, Kazuki
PA Shin-Etsu Chemical Industry Co., Ltd., Japan
SO Jpn. Kokai Tokkyo Koho, 51 pp.
CODEN: JKXXAF
DT Patent
LA Japanese
IC ICM C07C381-14
ICS G03F007-004; G03F007-039; H01L021-027
CC 74-5 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)
Section cross-reference(s): 23

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	JP 2003192665	A	20030709	JP 2002-275029	20020920
	JP 3991214	B2	20071017		
	US 2003180653	A1	20030925	US 2002-255770	20020927
	US 6689530	B2	20040210		
	TW 2259568	B	20050101	TW 2002-91122378	20020927
PRAI	JP 2001-300345	A	20010928		

CLASS

	PATENT NO.	CLASS	PATENT FAMILY CLASSIFICATION CODES
	JP 2003192665	ICM	C07C381-14
		ICS	G03F007-004; G03F007-039; H01L021-027
		IPCI	C07C0317-28 [I,A]; C07C0317-00 [I,C*]; G03F0007-004 [I,A]; G03F0007-039 [I,A]; H01L0021-027 [I,A]; H01L0021-02 [I,C*]
		IPCR	G03F0007-004 [I,C*]; G03F0007-004 [I,A]; C07C0381-00 [I,C*]; C07C0381-14 [I,A]; G03F0007-039 [I,C*]; G03F0007-039 [I,A]; H01L0021-02 [I,C*]; H01L0021-027 [I,A]
	US 2003180653	IPCR	C07C0317-00 [I,C*]; C07C0317-28 [I,A]; G03F0007-004 [I,C*]; G03F0007-004 [I,A]; G03F0007-038 [N,C*]; G03F0007-038 [N,A]; G03F0007-039 [N,C*]; G03F0007-039 [N,A]
		NCL	430/170.000; 430/270.100; 430/905.000; 534/558.000
	TW 225968	IPCR	C07C0317-00 [I,C*]; C07C0317-28 [I,A]; G03F0007-004 [I,C*]; G03F0007-004 [I,A]; G03F0007-038 [N,C*]; G03F0007-038 [N,A]; G03F0007-039 [N,C*]; G03F0007-039 [N,A]

OS MARPAT 139:92748

AB The compds. are $[(\text{Me}(\text{CH}_2)\text{mOn})\text{RkC}_6\text{H}_5-\text{n-kSO}_2]\text{pC:N}_2(\text{GR}_3)\text{q}$ [R = H, C1-4 alkyl(oxy); G = SO₂, CO; R₃ = C1-10 alkyl, C6-14 aryl; p = 1, 2; q = 0, 1; p + q = 2; n = 0, 1; m = 3-11; k = 0-4]. Photoresists containing acid-labile alkali-developable resins and radiation-sensitive acid generators including the compds. are also claimed. Further claimed is photolithog. wherein the photoresists are applied on substrates, annealed, exposed to ≤ 300 -nm actinic rays or electron beams, (annealed,) and developed. The photoresists produce patterns containing min. number of foreign

matter and having sharp profile.

ST amplified photoresist acid generator sulfonyldiazomethane photolithog; pattern profile development scum prevention photoresist PAG; butoxyphenylsulfonyldiazomethane photoacid generator amplified photoresist

IT Photoresists
(chemical amplified; novel sulfonyldiazomethane compds. for photoacid generators of chemical amplified photoresists)

IT Photolithography
(novel sulfonyldiazomethane compds. for photoacid generators of chemical amplified photoresists)

IT 532411-47-5P 552840-41-2P 552840-43-4P 552840-44-5P
RL: CAT (Catalyst use); CPS (Chemical process); IMF (Industrial manufacture); PEP (Physical, engineering or chemical process); TEM (Technical or engineered material use); PREP (Preparation); PROC (Process); USES (Uses)
(novel sulfonyldiazomethane compds. for photoacid generators of chemical amplified photoresists)

IT 552840-31-0P 552840-33-2P 552840-35-4P 552840-37-6P 552840-39-8P 552840-45-6P 552840-47-8P
RL: CAT (Catalyst use); IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)
(novel sulfonyldiazomethane compds. for photoacid generators of chemical amplified photoresists)

IT 24979-70-2D, Poly(p-hydroxystyrene), ethoxyethylated, butoxycarboxylated 130501-59-6, Poly(p-hydroxystyrene) acetate 159296-87-4, tert-Butyl acrylate-p-hydroxystyrene copolymer 326925-68-2, 1-Ethylcyclopentyl methacrylate-p-hydroxystyrene copolymer 345580-95-2, 1-Ethylcyclopentyl methacrylate-p-hydroxystyrene-styrene copolymer 552840-49-0 552840-50-3 552840-52-5D, tert-butoxycarboxylated 552840-54-7
RL: CPS (Chemical process); PEP (Physical, engineering or chemical process); TEM (Technical or engineered material use); PROC (Process);

USES
(Uses)
(novel sulfonyldiazomethane compds. for photoacid generators of chemical amplified photoresists)

IT 1129-79-9P 1135-49-5P 30752-19-3P 30752-20-6P 39969-57-8P 97412-69-6P 114833-51-2P 123883-51-2P 177217-25-3P 177217-27-5P 552840-56-9P 552840-58-1P 552840-59-2P 552840-61-6P 552840-63-8P 552840-65-0P 552840-68-3P 552840-69-4P 552840-70-7P 552840-74-1P 552840-77-4P 552840-81-0P
RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation);

RACT
(Reactant or reagent)
(novel sulfonyldiazomethane compds. for photoacid generators of chemical amplified photoresists)

IT 59-50-7, 4-Chloro-3-methylphenol 89-83-8 95-87-4, p-Xylenol 106-41-2, 4-Bromophenol 109-65-9 111-25-1 111-83-1 112-29-8, n-Decylbromide 143-15-7, n-Dodecyl bromide 2374-05-2, 2,6-Dimethyl-4-bromophenol 2416-94-6, 2,3,6-Trimethylphenol 51554-93-9, 1-Bromo-4-octylbenzene
RL: RCT (Reactant); RACT (Reactant or reagent)
(novel sulfonyldiazomethane compds. for photoacid generators of chemical amplified photoresists)

L8 ANSWER 11 OF 32 CA COPYRIGHT 2008 ACS on STN
AN 138:392956 CA
ED Entered STN: 12 Jun 2003
TI Highly transparent resist platforms for 157-nm microlithography: an update
AU Vohra, Vaishali Raghu; Douki, Katsuji; Kwark, Young-Je; Liu, Xiang-Qian; Ober, Christopher Kemper; Bae, Young C.; Conley, Will; Miller, Daniel; Zimmerman, Paul
CS Dep. Mater. Sci. Eng., Cornell Univ., Ithaca, NY, 14853, USA
SO Proceedings of SPIE-The International Society for Optical Engineering (2002), 4690(Pt. 1, Advances in Resist Technology and Processing XIX), 84-93
CODEN: PSISDG; ISSN: 0277-786X
PB SPIE-The International Society for Optical Engineering
DT Journal
LA English
CC 74-5 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)
AB Hexafluoroisopropyl alc.-functionalized acrylate monomers and their (co)polymers were prepared as photoresist platforms for 157 nm imaging. In order to balance transparency with other desirable traits such as etch resistance, the authors developed several copolymer systems. One is using 2-Me adamantly methacrylate as a comonomer, and the copolymer system showed better dissoln. contrast compared to the copolymer with tetrahydropyranyl methacrylate without sacrificing transparency. To further improve the absorption properties at 157 nm, monomers having a -trifluoromethyl group were prepared and polymerized in anionic mechanism. The product polymer was unexpectedly transparent at 157 nm ($\lambda = 1.6 \mu\text{m}^{-1}$) in spite that all the monomers contain carbonyl group. The second system is the copolymer with p-tert-butoxytetrafluorostyrene. P-Hydroxy-tetrafluorostyrene and p-tert-butoxy-tetrafluorostyrene were polymerized radically using AIBN in good yield, and the two resulting polymers showed distinctive solubility differences in aqueous base solution
Finally, this paper describes the synthesis of new monomers having fluorine (e.g CF₃-group) in the vicinity of the double bond to improve transparency at 157 nm. Due to the lower electron d. of the double bond, these monomers can be copolymerd. with electron-rich vinyl monomers using radical initiators.
ST highly transparent photoresist platform microlithog; vacuum UV photolithog
IT highly transparent resist platform
IT Fluoropolymers, properties
RL: PRP (Properties); SPN (Synthetic preparation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)
(design and evaluation of monomers and polymers for highly transparent chemical amplified photoresist platforms for 157-nm microlithog.)
IT Photoresists
(vacuum-UV, chemical amplified; design and evaluation of monomers and polymers for highly transparent chemical amplified photoresist platforms
for 157-nm microlithog.)
IT 403814-65-3P, p-Hydroxytetrafluorostyrene homopolymer

FOR 10559534 by Cynthia Hamilton

RL: PRP (Properties); SPN (Synthetic preparation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)
(design and evaluation of highly transparent resist platforms for 157-nm microlithog. based on tetrafluorostyrene polymers)

IT 403814-66-4P, p-tert-Butoxytetrafluorostyrene homopolymer
RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)
(design and evaluation of highly transparent resist platforms for 157-nm microlithog. based on tetrafluorostyrene polymers)

IT 479072-80-5DP, hydrolyzed, ethers with tetrahydropyran 524935-40-8P
RL: PRP (Properties); RCT (Reactant); SPN (Synthetic preparation); TEM (Technical or engineered material use); PREP (Preparation); RACT (Reactant or reagent); USES (Uses)
(design and evaluation of highly transparent resist platforms for 157-nm microlithog. based on trifluoromethylvinyl acetate)

IT 524935-40-8D, hydrolyzed
RL: PRP (Properties); TEM (Technical or engineered material use); USES (Uses)
(design and evaluation of highly transparent resist platforms for 157-nm microlithog. based on trifluoromethylvinyl acetate)

IT 479072-80-5
RL: RCT (Reactant); RACT (Reactant or reagent)
(design and evaluation of highly transparent resist platforms for 157-nm microlithog. based on trifluoromethylvinyl acetate)

IT 75-59-2, Tetramethylammonium hydroxide
RL: TEM (Technical or engineered material use); USES (Uses)
(developer; lithog. evaluation of highly transparent photoresist platforms for 157-nm microlithog.)

IT 479072-79-2P 479072-82-7P 479072-84-9P
RL: PRP (Properties); SPN (Synthetic preparation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)
(lithog. evaluation of hexafluoroisopropyl alc.-functionalized acrylate monomers and their polymers as chemical amplified photoresist for 157 nm exposures)

IT 1116-76-3, Trioctylamine 2052-49-5, Tetrabutylammonium hydroxide
RL: TEM (Technical or engineered material use); USES (Uses)
(lithog. evaluation of highly transparent photoresist platforms for 157-nm microlithog.)

IT 385422-30-0P, p-Hydroxytetrafluorostyrene
RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)
(monomer; design and evaluation of highly transparent resist platforms for 157-nm microlithog.)

IT 343305-41-9P, p-tert-Butoxytetrafluorostyrene
RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)
(monomer; design and evaluation of highly transparent resist platforms for 157-nm microlithog. based on tetrafluorostyrene polymers)

IT 646-97-9 695-12-5 19701-19-0 196314-61-1 242812-08-4
369375-16-6
479072-81-6 479072-89-4
RL: NUU (Other use, unclassified); USES (Uses)
(monomer; design and evaluation of monomers for highly transparent resist platforms for 157-nm microlithog.)

FOR 10559534 by Cynthia Hamilton

IT 188739-86-8P
RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)
(monomer; polymerization with hexafluoroisopropyl alc.-functionalized acrylate monomer)
IT 479072-78-1P
RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)
(monomer; polymerization with methyladamantyl(trifluoromethyl)acrylate)
IT 144317-44-2, Triphenylsulfonium perfluoro-1-butanesulfonate
RL: TEM (Technical or engineered material use); USES (Uses)
(photoacid generator; lithog. evaluation of highly transparent photoresist platforms for 157-nm microlithog.)
IT 479072-77-0P
RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)
(reaction with trifluoromethylacryloyl chloride)
IT 84540-57-8, Propylene glycol methyl ether acetate
RL: TEM (Technical or engineered material use); USES (Uses)
(solvent; lithog. evaluation of highly transparent photoresist platforms for 157-nm microlithog.)
RE.CNT 24 THERE ARE 24 CITED REFERENCES AVAILABLE FOR THIS RECORD
RE

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- (11) Ito, H; Poly Mater Sci Eng 1997, V77, P449 CA
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- (15) Nozaki, K; Chem Mater 1994, V6, P1492 CA
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- (17) Przybilla, K; Adv Mater 1992, V4, P239 CA
- (18) Przybilla, K; Proc SPIE-Int Soc Opt Eng 1992, V1672, P500 CA
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- (22) Schmaljohann, D; Proc SPIE-Int Soc Opt Eng 2000, V3999, P330 CA
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L8 ANSWER 12 OF 32 CA COPYRIGHT 2008 ACS on STN

AN 138:229130 CA

ED Entered STN: 03 Apr 2003

TI Photospeed considerations for extreme ultraviolet lithography resists
AU Dentinger, Paul M.; Hunter, Luke L.; O'Connell, Donna J.; Gunn, Scott;
Goods, Daniel; Fedynyshyn, Theodore H.; Goodman, Russell B.; Astolfi,

David K.
CS Sandia National Laboratories, Livermore, CA, 94550, USA
SO Journal of Vacuum Science & Technology, B: Microelectronics and Nanometer Structures (2002), 20(6), 2962-2967
CODEN: JVTBD9; ISSN: 0734-211X
PB American Institute of Physics
DT Journal
LA English
CC 74-5 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)
AB Photospeed is a prime consideration for wafer throughput of extreme-UV (EUV) lithog. Faster photoresists addnl. provide system advantages such as less thermal management of the mirrors and mask, and potentially increased component lifetimes. However, there are some predicted detrimental considerations when using fast photoresists such as shot noise. The authors report details of the formulation of photoresists exposed at 248 nm and identical formulations exposed at 13.4 nm. Compsns. typically contained co- or terpolymers of poly-4-hydroxystyrene, tert-Bu acrylate and as an option, styrene, a photoacid generator of bis-tert-butylphenyl iodonium camphorsulfonate or perfluorobenzensulfonate and tetrabutylammonium or triphenylsulfonium hydroxide base. With these formulations, the EUV photospeed was varied from 34 to 2.7 mJ/cm². Scanning electron microscope anal. was done for all wafers at Sandia using GORA software to determine the line-edge roughness
(LER). Identical formulations were exposed at photon dense deep-UV (DUV) wavelengths and comparatively photon-sparse EUV wavelengths. Therefore, ratioing the LER of identical formulations exposed at DUV minimizes the confounding effects of formulation changes that affect dissoln., resolution,
etc. A plot of the line-edge roughness ratio of LEREU/LERDUV as a function of the photospeed was used to gain insight into the effect of shot noise on the LER of dense lines. If shot noise effects were dominating the LER for EUV, then the ratio would increase above the nominal value as the dose to size was decreased. It is clear that the current generation of DUV resists, with the invention of no new material required, can be reformulated to satisfy the photospeed specification of
5
mJ/cm² for EUV lithog. and no shot noise effects have been observed
ST photospeed extreme UV lithog resist formulation; absorbed photon distribution photoresist extreme UV lithog
IT Photoresists
(extreme-UV; photospeed of resist composition for extreme-UV lithog.
and
effect of absorbed photon distribution of identical formulations under deep-UV and extreme-UV exposures)
IT 2052-49-5, Tetrabutylammonium hydroxide 17287-03-5, Trimethylsulfonium hydroxide 58621-56-0, Triphenylsulfonium hydroxide
RL: PRP (Properties); TEM (Technical or engineered material use); USES (Uses)
(base; photospeed of resist composition for extreme-UV lithog. and effect of
absorbed photon distribution of identical formulations under deep-UV and extreme-UV exposures)
IT 194861-06-8, Bis-tert-butylphenyliodonium camphorsulfonate 218151-20-3,

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Di-tert-butylphenyliodonium perfluorobutanesulfonate 220122-68-9,
Di-tert-butylphenyliodonium perfluorobenzenesulfonate 240435-11-4,
Di-tert-butylphenyliodonium perfluoroctanesulfonate
RL: PRP (Properties); TEM (Technical or engineered material use); USES
(Uses)
(photoacid generator; photospeed of resist composition for
extreme-UV lithog. and effect of absorbed photon distribution of
identical formulations under deep-UV and extreme-UV exposures)
IT 159296-87-4, Tert-Butyl acrylate-4-hydroxystyrene copolymer
200808-68-0, Tert-Butyl acrylate-4-hydroxystyrene-styrene
copolymer
RL: PRP (Properties); TEM (Technical or engineered material use); USES
(Uses)
(photospeed of resist composition for extreme-UV lithog. and effect of
absorbed photon distribution of identical formulations under deep-UV
and extreme-UV exposures)

RE.CNT 15 THERE ARE 15 CITED REFERENCES AVAILABLE FOR THIS RECORD

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- (6) Dentinger, P; J Vac Sci Technol B 2000, V18, P3364 CA
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- (8) Gallatin, G; Lithography for Semiconductor Manufacturing 2001, VII
- (9) Neureuther, A; J Vac Sci Technol B 1988, V6, P167 CA
- (10) O'Brien, S; Optical Microlithography 2001, VXIV
- (11) Rau, N; J Vac Sci Technol B 1998, V16, P3784 CAPLUS
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L8 ANSWER 13 OF 32 CA COPYRIGHT 2008 ACS on STN

AN 137:370814 CA

ED Entered STN: 12 Dec 2002

TI Synthesis of silyl-containing (meth)acrylate-based copolymers
and bilayer resist compositions therefrom

IN Lee, Haiwon; Kim, Sung Soo

PA Hanyang Hak Won Co., Ltd., S. Korea

SO U.S. Pat. Appl. Publ., 10 pp.

CODEN: USXXCO

DT Patent

LA English

IC ICM C08F124-00

INCL 526266000

CC 37-3 (Plastics Manufacture and Processing)
Section cross-reference(s): 35, 74

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	US 2002173609	A1	20021121	US 2002-144110	20020510
	US 6596830	B2	20030722		

KR 2002086177	A	20021118	KR 2001-25960	20010511
PRAI KR 2001-25960	A	20010511		

CLASS

PATENT NO.	CLASS	PATENT FAMILY CLASSIFICATION CODES
US 2002173609	ICM INCL IPCI IPCR	C08F124-00 526266000 C08F0124-00 [ICM,7] C08F0230-00 [I,C*]; C08F0230-08 [I,A]; G03F0007-004 [N,C*]; G03F0007-004 [N,A]; G03F0007-075 [I,C*]; G03F0007-075 [I,A] NCL ECLA
KR 2002086177	IPCI IPCR ECLA	526/266.000; 526/279.000; 526/319.000 C08F230/08; G03F007/075M2 G03F0007-075 [ICM,7] C08F0230-00 [I,C*]; C08F0230-08 [I,A]; G03F0007-004 [N,C*]; G03F0007-004 [N,A]; G03F0007-075 [I,C*]; G03F0007-075 [I,A] C08F230/08; G03F007/075M2

AB The title copolymers, for use in a chemical amplified resist compns., are generally obtained from copolyrn. of a silyl-containing (meth)acrylate monomer (A) and a lactone-containing (meth)acrylate monomer (B), wherein A is synthesized from a silyl-containing alc. with (meth)acryloyl chloride, and B is synthesized from a OH-containing lactone with (meth)acryloyl chloride. Thus, reacting [(CH₃)₃SiCH₂]₂CH₂OH, prepared

from

Grignard reaction of (CH₃)₃SiCH₂MgCl and HCOOEt, with methacryloyl chloride gave an A monomer, reacting pantolactone with methacryloyl chloride gave a B monomer, A and B were then polymerized at 65-70° in the presence of AIBN to give a title copolymer, 2 g of which was then dissolved in 16 g polypropylene glycol Me ether acetate with 0.02 g triarylsulfonium triflate, filtered and spin-coated on a silicon wafer treated with hexamethyldisilazane to form a resist layer.

ST silyl lactone contg methacrylate copolymer synthesis bilayer resist compn
IT Photoresists

(bilayer resist compns. using silyl- and lactone-containing (meth) acrylate copolymers)

IT Polymers, preparation

RL: IMF (Industrial manufacture); PEP (Physical, engineering or chemical process); PRP (Properties); PYP (Physical process); TEM (Technical or engineered material use); PREP (Preparation); PROC (Process); USES (Uses)
(co-; synthesis of silyl- and lactone-containing (meth) acrylate copolymers suitable for bilayer resist compns.)

IT Resistors

(film; bilayer resist compns. using silyl- and lactone-containing (meth) acrylate copolymers)

IT Sulfonic acids, uses

RL: MOA (Modifier or additive use); USES (Uses)
(salts, photoacid generator; bilayer resist compns. using silyl- and lactone-containing (meth)acrylate copolymers)

IT 79-50-5, dl-Pantoyl lactone

RL: RCT (Reactant); RACT (Reactant or reagent)
(in synthesis of lactone-containing methacrylate monomer for silyl- and lactone-containing (meth)acrylate copolymers)

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IT 109-94-4, Ethyl formate 920-46-7, Methacryloyl chloride 2344-80-1, (Chloromethyl)trimethylsilane
RL: RCT (Reactant); RACT (Reactant or reagent)
(in synthesis of silyl-containing methacrylate monomer for silyl- and lactone-containing (meth)acrylate copolymers)

IT 17887-33-1P, 1,3-Bis(trimethylsilyl)-2-propanol
RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation);
RACT
(Reactant or reagent)
(intermediate; in synthesis of silyl-containing methacrylate monomer for
silyl- and lactone-containing (meth)acrylate copolymers)

IT 156938-13-5P
RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation);
RACT
(Reactant or reagent)
(monomer; in synthesis of lactone-containing methacrylate monomer for
silyl- and lactone-containing (meth)acrylate copolymers)

IT 195044-28-1P, 1,3-Bis(trimethylsilyl)isopropyl methacrylate
RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation);
RACT
(Reactant or reagent)
(monomer; synthesis of silyl-containing methacrylate monomer for
silyl- and
lactone-containing (meth)acrylate copolymers)

IT 475599-61-2P 475599-62-3P
RL: IMF (Industrial manufacture); PEP (Physical, engineering or chemical
process); PRP (Properties); PYP (Physical process); TEM (Technical or
engineered material use); PREP (Preparation); PROC (Process); USES (Uses)
(synthesis of silyl- and lactone-containing (meth)acrylate
copolymers suitable for bilayer resist compns.)

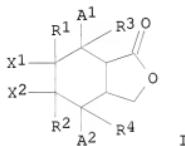
IT 7440-21-3, Silicon, miscellaneous
RL: MSC (Miscellaneous)
(wafer, substrate; bilayer resist compns. using silyl- and
lactone-containing (meth)acrylate copolymers)

L8 ANSWER 14 OF 32 CA COPYRIGHT 2008 ACS on STN
AN 137:39324 CA
ED Entered STN: 11 Jul 2002
TI (Meth)acrylate esters, starting alcohols for the preparation
thereof, processes for preparing both, polymers of the esters, chemically
amplifiable resist compositions, and method for forming patterns
IN Kamon, Yoshihiro; Fujiwara, Tadayuki; Kuwano, Hideaki; Momose, Hikaru;
Koizumi, Atsushi
PA Mitsubishi Rayon Co., Ltd., Japan
SO PCT Int. Appl., 109 pp.
CODEN: PIXXD2
DT Patent
LA Japanese
IC ICM C07D307-93
ICS C07D307-88; C07D493-18; C07D307-77; C07D493-18; C07D307-04;
C07D307-33
CC 74-5 (Radiation Chemistry, Photochemistry, and Photographic and Other
Reproductive Processes)
Section cross-reference(s): 27, 35, 38

FOR 10559534 by Cynthia Hamilton

FAN.CNT 1					
	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	WO 2002046179	A1	20020613	WO 2001-JP10628	20011205
	W: KR, US				
	RW: AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL,				
	PT, SE, TR				
	JP 2002234882	A	20020823	JP 2001-366958	20011130
	JP 2002275215	A	20020925	JP 2001-368904	20011203
	EP 1352904	A1	20031015	EP 2001-999568	20011205
	R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,				
	IE, FI, CY, TR				
	TW 583182	B	20040411	TW 2001-90130267	20011206
	US 2004063882	A1	20040401	US 2003-433570	20030605
	US 7041838	B2	20060509		
	US 2005113538	A1	20050526	US 2004-974876	20041028
	US 7339014	B2	20080304		
PRAI	JP 2000-371712	A	20001206		
	JP 2001-1728	A	20010109		
	JP 2001-366958	A	20011130		
	JP 2001-368904	A	20011203		
	WO 2001-JP10628	W	20011205		
	US 2003-433570	A3	20030605		
CLASS					
	PATENT NO.	CLASS	PATENT FAMILY	CLASSIFICATION CODES	
	WO 2002046179	ICM	C07D307-93		
		ICS	C07D307-88; C07D493-18; C07D307-77; C07D493-18;		
			C07D307-04; C07D307-33		
		IPCI	C07D0307-93 [ICM, 7]; C07D0307-88 [ICS, 7]; C07D0493-18		
			[ICS, 7]; C07D0493-00 [ICS, 7, C*]; C07D0307-77 [ICS, 7, C*]		
			C07D0307-04 [ICS, 7]; C07D0307-33 [ICS, 7]; C07D0307-00		
		IPCR	[ICS, 7, C*]; C07D0307-00 [I, C*]; C07D0307-88 [I, A]; C07D0307-93		
			[I, A]; C07D0493-00 [I, C*]; C07D0493-18 [I, A];		
			C08F0220-00 [I, C*]; C08F0220-18 [I, A]; C08F0220-28		
			[I, A]; G03F0007-039 [I, C*]; G03F0007-039 [I, A]		
		ECLA	C07D493/18+307C+307C+307B+3; C08F220/18; C08F220/28;		
			G03F0007/039C15; C07D307/88; C07D307/93		
JP	2002234882	IPCI	C07D0307-00 [ICM, 7]; C07D0307-88 [ICS, 7]; C07D0493-18		
		IPCR	[ICS, 7]; C07D0493-00 [ICS, 7, C*]		
			C07D0307-00 [I, C*]; C07D0307-00 [I, A]; C07D0307-88		
			[I, A]; C07D0493-00 [I, C*]; C07D0493-18 [I, A]		
JP	2002275215	IPCI	C08F0020-28 [ICM, 7]; C08F0020-00 [ICM, 7, C*];		
			G03F0007-039 [ICS, 7]; H01L0021-027 [ICS, 7];		
H01L0021-02					
		IPCR	[ICS, 7, C*]; G03F0007-039 [I, C*]; G03F0007-039 [I, A]; C08F0020-00		
			[I, C*]; C08F0020-28 [I, A]; H01L0021-02 [I, C*];		
			H01L0021-027 [I, A]		
EP	1352904	IPCI	C07D0307-93 [ICM, 7]; C07D0307-88 [ICS, 7]; C07D0493-18		
		IPCR	[ICS, 7]; C07D0493-00 [ICS, 7, C*]; C07D0307-77 [ICS, 7, C*]		
			C07D0307-04 [ICS, 7]; C07D0307-33 [ICS, 7]; C07D0307-00		
			[ICS, 7, C*]; C07D0307-00 [I, C*]; C07D0307-88 [I, A]; C07D0307-93		

		[I,A]; C07D0493-00 [I,C*]; C07D0493-18 [I,A]; C08F0220-00 [I,C*]; C08F0220-18 [I,A]; C08F0220-28 [I,A]; G03F0007-039 [I,C*]; G03F0007-039 [I,A]
	ECLA	C07D307/88; C07D307/93; C07D493/18+307C+307C+307B+3; C08F220/18; C08F220/28; G03F007/039C1S
TW 583182	IPCI	C07D0307-93 [ICM,7]; C07D0307-88 [ICs,7]; C07D0307-77 [ICs,7]; C07D0307-00 [ICs,7,C*]
	IPCR	C07D0307-00 [I,C*]; C07D0307-88 [I,A]; C07D0307-93 [I,A]; C07D0493-00 [I,C*]; C07D0493-18 [I,A]; C08F0220-00 [I,C*]; C08F0220-18 [I,A]; C08F0220-28 [I,A]; G03F0007-039 [I,C*]; G03F0007-039 [I,A]
US 2004063882	IPCI	C07D0407-00 [I,A]
	IPCR	C07D0307-00 [I,C*]; C07D0307-88 [I,A]; C07D0307-93 [I,A]; C07D0493-00 [I,C*]; C07D0493-18 [I,A]; C08F0220-00 [I,C*]; C08F0220-18 [I,A]; C08F0220-28 [I,A]; G03F0007-039 [I,C*]; G03F0007-039 [I,A]
	NCL	526/266.000; 549/305.000
	ECLA	C07D307/88; C07D307/93; C07D493/18+307C+307C+307B+3; C08F220/18; C08F220/28; G03F007/039C1S
US 2005113538	IPCI	C08F0024-00 [I,A]; C08F0034-02 [I,A]; C08F0034-00 [I,C*]
	IPCR	C07D0307-00 [I,C*]; C07D0307-88 [I,A]; C07D0307-93 [I,A]; C07D0493-00 [I,C*]; C07D0493-18 [I,A]; C08F0220-00 [I,C*]; C08F0220-18 [I,A]; C08F0220-28 [I,A]; G03F0007-039 [I,C*]; G03F0007-039 [I,A]
	NCL	526/266.000; 526/319.000
	ECLA	C07D307/88; C07D307/93; C07D493/18+307C+307C+307B+3; C08F220/18; C08F220/28; G03F007/039C1S
OS	MARPAT	137:39324
GI		



AB (Meth)acrylate esters are represented by the general formula I (R1-4 = H, Me, Et; one of X1 and X2 is (meth)acryloyloxy and the other is H; A1 and A2 are H or form O, CH₂, CH₂CH₂). These esters can be prepared by preparing a product of addition of a 1,3-diene with maleic anhydride by Diels-Alder reaction, reducing this product into a lactone, hydrating this lactone into an alc., and esterifying this alc. with (meth)acrylic acid. The (co)polymers produced by polymerizing monomer compns. containing the (meth)acrylate esters are excellent in transparency, dry-etching resistance, and solubility in organic solvents, and useful as resins for chemical

amplifiable resist compns.

ST methacrylate acrylate ester copolymer electron beam resist photoresist; chem amplification resist

IT Electron beam resists

Photoresists

 ((meth)acrylate-based chemical amplification-type resist)

IT Diels-Alder reaction

 (preparation of (meth)acrylate-based chemical amplification-type resist)

IT 66003-78-9, Triphenylsulfoniumtriflate

RL: CAT (Catalyst use); USES (Uses)

 (photoacid; (meth)acrylate-based chemical amplification-type resist)

IT 80-62-6, Methyl methacrylate 85-43-8 108-31-6, Maleic anhydride, reactions 760-93-0, Methacrylic anhydride 826-62-0, 5-Norbornene-2,3-dicarboxylic anhydride 920-46-7, Methacrylic acid chloride 6118-51-0, exo-3,6-Epoxy-1,2,3,6-tetrahydrophthalic anhydride 25134-21-8, Methyl-5-norbornene-2,3-dicarboxylic anhydride

RL: RCT (Reactant); RACT (Reactant or reagent)

 (preparation of (meth)acrylate-based chemical amplification-type resist)

IT 24327-08-0P, endo-Bicyclo[2.2.2]octo-5-ene-2,3-dicarboxylic anhydride 85718-44-1P, 4-Oxatricyclo[5.2.1.0_{2,6}]oct-8-decene-3-one 436852-32-3P 436852-33-4P 436852-34-5P 436852-35-6P 436852-36-7P 436852-37-8P 436852-38-9P 436852-40-3P 436852-41-4P 436852-42-5P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

 (preparation of (meth)acrylate-based chemical amplification-type resist)

IT 436852-43-6P 436852-44-7P 436852-45-8P 436852-46-9P 436852-47-0P 436852-48-1P 436852-49-2P 436852-50-5P 436852-51-6P 436852-52-7P 436852-54-9P 436852-57-2P 436852-59-4P

RL: SPN (Synthetic preparation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

 (preparation of (meth)acrylate-based chemical amplification-type resist)

IT 97-64-3, Ethyl lactate 84540-57-8, Propylene glycol monomethylether acetate

RL: TEM (Technical or engineered material use); USES (Uses)

 (preparation of (meth)acrylate-based chemical amplification-type resist)

IT 68-12-2, N,N-Dimethylformamide, uses 108-65-6, 2-Acetoxy-1-methoxypropane 109-99-9, Tetrahydrofuran, uses 123-91-1, 1,4-Dioxane, uses

RL: NNU (Other use, unclassified); USES (Uses)

 (solvent; preparation of (meth)acrylate-based chemical amplification-type resist)

RE.CNT 5 THERE ARE 5 CITED REFERENCES AVAILABLE FOR THIS RECORD

RE

(1) Mitsubishi Rayon Co Ltd; JP 20026502 A 2002

(2) Mori, K; Liebigs Ann Chem 1993, 6, P671 CA

(3) Numata, A; Yakugaku Zasshi 1968, V88(9), P1151 CA

(4) Squibb E R And Sons Inc; US 4143054 A 1980 CA

(5) Squibb E R And Sons Inc; US 4187236 A 1980 CA

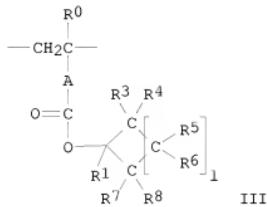
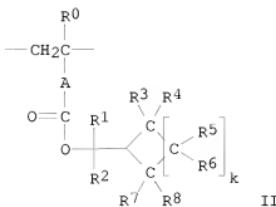
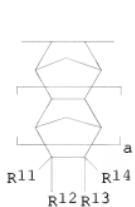
FOR 10559534 by Cynthia Hamilton

L8 ANSWER 15 OF 32 CA COPYRIGHT 2008 ACS on STN
AN 136:301775 CA
ED Entered STN: 02 May 2002
TI Positive-working photoresist compositions containing alkali-soluble resin
having norbornene structure
IN Sato, Kenichiro
PA Fuji Photo Film Co., Ltd., Japan
SO Jpn. Kokai Tokkyo Koho, 40 pp.
CODEN: JKXXAF
DT Patent
LA Japanese
IC ICM G03F007-039
ICS C08F220-18; C08F222-40; C08F232-00; C08K005-00; C08L045-00;
G03F007-004; H01L021-027
CC 74-5 (Radiation Chemistry, Photochemistry, and Photographic and Other
Reproductive Processes)
FAN.CNT 2
PATENT NO. KIND DATE APPLICATION NO. DATE

PI JP 2002099087 A 20020405 JP 2000-290654 20000925
US 2002064727 A1 20020530 US 2001-960343 20010924
US 6727039 B2 20040427
KR 765245 B1 20071009 KR 2001-58973 20010924
TW 231890 B 20050501 TW 2001-90123605 20010925
PRAI JP 2000-290654 A 20000925
JP 2000-296881 A 20000928
CLASS
PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES

JP 2002099087 ICM G03F007-039
ICS C08F220-18; C08F222-40; C08F232-00; C08K005-00;
C08L045-00; G03F007-004; H01L021-027
IPCI G03F0007-039 [ICM,7]; C08F0220-18 [ICS,7]; C08F0220-00
[ICS,7,C*]; C08F0222-40 [ICS,7]; C08F0222-00
[ICS,7,C*]; C08F0232-00 [ICS,7]; C08K0005-00 [ICS,7];
C08L0045-00 [ICS,7]; G03F0007-004 [ICS,7];
H01L0021-027
[ICS,7]; H01L0021-02 [ICS,7,C*]
IPCR G03F0007-039 [I,C*]; G03F0007-039 [I,A]; C08F0220-00
[I,C*]; C08F0220-18 [I,A]; C08F0222-00 [I,C*];
C08F0222-40 [I,A]; C08F0232-00 [I,C*]; C08F0232-00
[I,A]; C08K0005-00 [I,C*]; C08K0005-00 [I,A];
C08L0045-00 [I,C*]; C08L0045-00 [I,A]; G03F0007-004
[I,C*]; G03F0007-004 [I,A]; H01L0021-02 [I,C*];
H01L0021-027 [I,A]
US 2002064727 IPCI G03F0007-004 [ICM,7]
IPCR G03F0007-039 [I,C*]; G03F0007-039 [I,A]
NCL 430/270.100; 430/914.000; 430/921.000
ECLA G03F0007/039C15
KR 765245 IPCI G03F0007-039 [I,A]
TW 231890 IPCI G03F0007-039 [ICS,7]; C08F0220-10 [ICS,7]; C08F0220-00
[ICS,7,C*]; C08F0222-06 [ICS,7]; C08F0222-00
[ICS,7,C*]
IPCR G03F0007-039 [I,C*]; G03F0007-039 [I,A]
OS MARPAT 136:301775

GI



AB The compns., useful for semiconductor devices, contain (A) resins which contain a repeating unit I (R11-R14 = H, alkyl; a = 0, 1) and II (R0 = H, lower alkyl; A = direct bond, alkylene, cycloalkylene, O, S, CO, ester group; R1, R2 = lower alkyl; R3-R8 = H, lower alkyl, lower alkoxy, halo; CR3R4, CR5R6 may be CO; R3 and R5 may be bonded together to form alkylene; k = 2-5) or III (R0-R8, A = any group given for those in

II; 1 = 2-5) and show increased solubility to an alkaline developer by action of acids and (B) photoacid generators. The photoresist compns.

show good storage stability, adhesion, and low exposure margin.

ST pos photoresist norbornene cyclohexylmethyl acrylate copolymer

IT Positive photoresists

(pos.-working photoresist compns. containing alkali-soluble norbornene-

acrylate polymers)

IT 66003-78-9 81416-37-7 116808-67-4 138529-81-4 138529-84-7
 142096-70-6 144317-44-2 145612-66-4 153698-46-5 179419-32-0
 241806-75-7 307531-76-6 312386-77-9 324771-13-3 338445-29-7
 338445-31-1 341548-84-3 406722-76-7 406722-77-8
 RL: CAT (Catalyst use); TEM (Technical or engineered material use); USES
 (Uses)
 (photoacid generator; pos.-working photoresist compns. containing

alkali-soluble norbornene-acrylate polymers)
IT 398140-79-9P 406722-62-1P 406722-63-2P 406722-64-3P 406722-65-4P
406722-66-5P 406722-67-6P 406722-68-7P 406722-69-8P 406722-71-2P
406722-72-3P 406722-73-4P 406722-74-5P 406722-75-6P
RL: SPN (Synthetic preparation); TEM (Technical or engineered material
use); PREP (Preparation); USES (Uses)
(pos.-working photoresist compns. containing alkali-soluble
norbornene-
acrylate polymers)

L8 ANSWER 16 OF 32 CA COPYRIGHT 2008 ACS on STN
AN 136:301658 CA
ED Entered STN: 02 May 2002
TI Tailoring transparency of imageable fluoropolymers at 157 nm by
incorporation of hexafluoroisopropyl alcohol to photoresist backbones
AU Bae, Young C.; Douki, Katsushi; Yu, Tianyue; Dai, Junyan; Schmaljohann,
Dirk; Koerner, Hilmar; Ober, Christopher K.; Conley, Will
CS Department of Materials Science & Engineering, Cornell University,
Ithaca,
NY, 14853, USA
SO Chemistry of Materials (2002), 14(3), 1306-1313
CODEN: CMATEX; ISSN: 0897-4756
PB American Chemical Society
DT Journal
LA English
CC 74-5 (Radiation Chemistry, Photochemistry, and Photographic and Other
Reproductive Processes)
AB Hexafluoroisopropyl alc.-functionalized acrylic and styrenic monomers,
such as 2-[4-(2-hydroxyhexafluoro
isopropyl)cyclohexane]hexafluoroisopropyl
1 acrylate (2), 2-[4-(2,2,2-trifluoro-1-methoxy-methoxy-1-
trifluoromethyl)ethyl)cyclohexane]hexafluoroisopropyl acrylate
(3), and 2-[4-(2,2,2-trifluoro-1-ethoxymethoxy-1-
trifluoromethyl)ethyl]styrene (4), were synthesized, and their (co
)polymers were studied as photoresist platforms for 157 nm lithog. It
was found that these (co)polymers are unusually transparent at 157
nm, and absorbances of poly(2) and poly(2-co-4) were determined to be
1.93 and 2.38 μ m⁻¹, resp. Results indicated that both
electron-withdrawing effects and bulkiness of CF₃ groups play important
roles in tailoring the absorbance of chromophores. Lithog. studies were
carried out with poly(2)-based resists using 157 and 248 nm steppers, and
it was demonstrated that, after selective modification, it is possible to
use conventional resist backbones, such as acrylic or styrenic polymers,
in the design of single-layer resists for 157 nm lithog.
ST lithog photoresist acrylic styrene polymer hexafluoroisopropyl alc group;
photolithog vacuum UV resist polymer hexafluoroisopropyl alc group;
fluorocarbonol contg polymer chem amplified photoresist vacuum UV lithog
IT Photoresists
(chemical amplified; lithog. characterization of chemical amplified
157 nm photoresists based on polymers containing hexafluoroisopropyl
alc.-functionalized acrylic and styrenic monomers)
IT Absorptivity
Thermal stability

Transparency
UV and visible spectra
(lithog. characterization of chemical amplified 157 nm photoresists
based on polymers containing hexafluoroisopropyl alc.-functionalized
acrylic and
styrenic monomers)
IT Fluoropolymers, properties
RL: PRP (Properties); SPN (Synthetic preparation); TEM (Technical or
engineered material use); PREP (Preparation); USES (Uses)
(lithog. characterization of chemical amplified 157 nm photoresists
based on polymers containing hexafluoroisopropyl alc.-functionalized
acrylic and
styrenic monomers)
IT 1116-76-3, Trioctylamine 2052-49-5, Tetrabutylammonium hydroxide
RL: NUU (Other use, unclassified); USES (Uses)
(acid diffusion inhibitor; lithog. characterization of chemical
amplified
157 nm photoresists based on polymers containing hexafluoroisopropyl
alc.-functionalized acrylic and styrenic monomers)
IT 75-59-2, Tetramethylammonium hydroxide
RL: NUU (Other use, unclassified); USES (Uses)
(developer; lithog. characterization of chemical amplified 157 nm
photoresists based on polymers containing hexafluoroisopropyl
alc.-functionalized acrylic and styrenic monomers)
IT 406939-14-8P
RL: PRP (Properties); SPN (Synthetic preparation); TEM (Technical or
engineered material use); PREP (Preparation); USES (Uses)
(lithog. characterization of chemical amplified 157 nm photoresists
based on polymers containing hexafluoroisopropyl alc.-functionalized
acrylic and
styrenic monomers)
IT 368422-52-0P, 2-[4-(2-Hydroxyhexafluoro
isopropyl)cyclohexane]hexafluorois
opropyl acrylate homopolymer 406939-11-5P 406939-12-6P
RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
(lithog. characterization of polymers containing hexafluoroisopropyl
alc.-functionalized acrylic and styrenic monomers in chemical
amplified
photoresist formulations)
IT 144317-44-2, Triphenylsulfonium perfluoro-1-butanesulfonate
RL: TEM (Technical or engineered material use); USES (Uses)
(photoacid generator; lithog. characterization of chemical
amplified 157 nm photoresists based on polymers containing
hexafluoroisopropyl alc.-functionalized acrylic and styrenic monomers)
IT 84540-57-8, Propylene glycol methyl ether acetate
RL: NUU (Other use, unclassified); USES (Uses)
(resist solvent; lithog. characterization of chemical amplified 157 nm
photoresists based on polymers containing hexafluoroisopropyl
alc.-functionalized acrylic and styrenic monomers)
IT 406939-09-1P
RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
(Reactant or reagent)

FOR 10559534 by Cynthia Hamilton

(synthesis of hexafluoroisopropyl alc.-containing acrylic monomers for use in chemical amplified photoresist for vacuum-UV lithog. lithog)
IT 367522-45-0P, 2-[4-(2-Hydroxyhexafluoroisopropyl)cyclohexane]hexafluoroisopropyl acrylate 367522-47-2P, 2-[4-(2,2,2-Trifluoro-1-methoxy-1-trifluoromethylhexyl)cyclohexane]hexafluoroisopropyl acrylate 367522-47-2P, 2-[4-(2,2,2-Trifluoro-1-ethoxymethoxy-1-trifluoromethylhexyl)styrene
RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)
(synthesis of hexafluoroisopropyl alc.-containing polymers for use in chemical amplified photoresist for vacuum-UV lithog. lithog)
RE.CNT 30 THERE ARE 30 CITED REFERENCES AVAILABLE FOR THIS RECORD
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L8 ANSWER 17 OF 32 CA COPYRIGHT 2008 ACS on STN

AN 136:270596 CA

ED Entered STN: 18 Apr 2002

TI Blends of hydroxystyrene polymers for use in chemically amplified positive

resist formulations

IN Chen, Kuang-Jung; DellaGuardia, Ronald Anthony; Ito, Hiroshi; Jordhamo, George Michael; Katnani, Ahmad Dauod

FOR 10559534 by Cynthia Hamilton

PA International Business Machines Corporation, USA
SO U.S., 12 pp.
CODEN: USXXAM
DT Patent
LA English
IC ICM G03F007-004
INCL 430270100
CC 74-5 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)
Section cross-reference(s): 35, 38, 76

FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI US 6365321	B1	20020402	US 1999-291389	19990413
PRAI US 1999-291389		19990413		

CLASS

PATENT NO.	CLASS	PATENT FAMILY CLASSIFICATION CODES
US 6365321	ICM	G03F007-004
	INCL	430270100
	IPCI	G03F0007-004 [ICM,7]
	IPCR	G03F0007-004 [I,C*]; G03F0007-004 [I,A]; G03F0007-039 [I,C*]; G03F0007-039 [I,A]
	NCL	430/270.100; 430/905.000; 430/910.000
	ECLA	G03F007/004D; G03F007/039C1S

AB A photoresist binder composition comprises a homogeneous blend of (A) a hydroxystyrene copolymer comprising a first monomer that is optionally substituted hydroxystyrene and a second monomer containing an acid labile group, preferably pendant to the polymer backbone, and (B) and a phenolic polymer, that is optionally partially or wholly protected, such as polyhydroxystyrene, poly(hydroxystyrene-co-styrene), poly(hydroxystyrene-co-styrene-co-t-Bu acrylate), novolac, and the like. Also provided is a lithog. resist composition comprising the homogeneous blend of the photoresist binder composition, and a radiation-sensitive acid generator which generates an acid upon exposure to radiation, and a process for using the resist composition to generate resist images on a substrate, such as in the manufacture of integrated circuits or the like.

ST chem amplified photoresist binder hydroxystyrene blend photolithog integrated circuit

IT Phenolic resins, uses

RL: POF (Polymer in formulation); USES (Uses)
(blends of hydroxystyrene polymers for use in chemical amplified pos. resist formulations)

IT Polymer blends

RL: POF (Polymer in formulation); TEM (Technical or engineered material use); USES (Uses)
(blends of hydroxystyrene polymers for use in chemical amplified pos. resist formulations)

IT Photolithography
(blends of hydroxystyrene polymers for use in chemical amplified pos.

resist formulations for)
IT Integrated circuits
(blends of hydroxystyrene polymers for use in chemical amplified pos.
resist formulations in relation to)
IT Positive photoresists
(chemical amplified; blends of hydroxystyrene polymers for use in
chemical
amplified pos. resist formulations)
IT 24979-74-6, p-Hydroxystyrene-styrene copolymer
RL: POF (Polymer in formulation); TEM (Technical or engineered material
use); USES (Uses)
(blend with hydrolyzed acetoxystyrene-tert-Bu acrylate
copolymer; blends of hydroxystyrene polymers for use in chemical
amplified
pos. resist formulations in relation to)
IT 174476-25-6DP, p-Acetoxystyrene-tert-butyl acrylate copolymer,
hydrolyzed
RL: POF (Polymer in formulation); SPN (Synthetic preparation); TEM
(Technical or engineered material use); PREP (Preparation); USES (Uses)
(blend with hydroxystyrene-styrene copolymer; blends of hydroxystyrene
polymers for use in chemical amplified pos. resist formulations in
relation to)
IT 174476-25-6P, p-Acetoxystyrene-tert-butyl acrylate copolymer
RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
(Reactant or reagent)
(blends of hydroxystyrene polymers for use in chemical amplified pos.
resist formulations in relation to)
IT 45187-15-3, Perfluororobutanesulfonate 194861-06-8, Di-(tert-
butylphenyl)iodonium camphorsulfonate
RL: TEM (Technical or engineered material use); USES (Uses)
(photoacid generator; blends of hydroxystyrene polymers for
use in chemical amplified pos. resist formulations)
IT 97-64-3, Ethyl lactate 763-69-9, Ethyl 3-ethoxypropionate
RL: TEM (Technical or engineered material use); USES (Uses)
(solvent; blends of hydroxystyrene polymers for use in chemical
amplified
pos. resist formulations)
RE.CNT 14 THERE ARE 14 CITED REFERENCES AVAILABLE FOR THIS RECORD
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L8 ANSWER 18 OF 32 CA COPYRIGHT 2008 ACS on STN
AN 136:254474 CA
ED Entered STN: 11 Apr 2002
TI Free-radical synthesis of narrow polydispersed 2-hydroxyethyl methacrylate-based tetrapolymers for dilute aqueous base developable negative photoresists
AU Diakoumakos, Constantinos D.; Raptis, Ioannis; Tserepi, Angeliki; Argitis, Panagiotis
CS Institute of Microelectronics, NCSR Demokritos, Athens, 15343, Greece
SO Polymer (2001), Volume Date 2002, 43(4), 1103-1113
CODEN: POLMAG; ISSN: 0032-3861
PB Elsevier Science Ltd.
DT Journal
LA English
CC 74-5 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)
Section cross-reference(s): 37, 38
AB Novel (meth)acrylate tetrapolymers based on 2-hydroxyethyl methacrylate (HEMA) were synthesized via free-radical polymerization in refluxing xylene under monomer-starved conditions for use in neg. photoresist formulations. 2,2'-Azobis(2-methylbutyronitrile) was used as initiator and 2-mercaptoethanol as chain transfer agent. Optimized resist formulations were obtained with a relatively narrow polydispersed (D=1.86) low mol. weight copolymer (Mn=1677) of 2-hydroxyethyl methacrylate (HEMA), isobornyl methacrylate (IBMA), cyclohexyl methacrylate (CHMA) and acrylic acid (AA), in a 40/30/23/7 weight ratio. A novel high-resolution single layer neg. tone photoresist suitable for 193 nm and e-beam lithog. that meets basic performance requirements (aqueous-base development, enhanced etch resistance, sub-0.2 μ m resolution) was developed from the aforementioned (meth)acrylate tetrapolymer, the poly(2-hydroxyethyl methacrylate-co-cyclohexyl methacrylate-co-isobornyl methacrylate-co-acrylic acid) (PHECIMA) and a sulfonium salt photoacid generator. The key-components for the neg. image formation (photoacid induced crosslinking) are the hydroxyl groups of the HEMA moieties. The swelling-free neg. resist material was developed in diluted aqueous base [tetramethyl ammonium hydroxide, (TMAH) 0.26+10-2N] and presented enhanced etch resistance without the use of etch resistance promoters. 0.20-0.14 μ M lines were obtained upon 193 nm and/or e-beam lithog.
ST methacrylate deriv tetrapolymer synthesis swelling free neg photoresist lithog; aq base developable neg photoresist methacrylate deriv tetrapolymer synthesis; free radical polymn methacrylate deriv tetrapolymer synthesis neg photoresist
IT Photolithography
 (UV; free-radical synthesis of narrow polydispersed 2-hydroxyethyl methacrylate-based tetrapolymers for swelling-free dilute aqueous base developable neg. photoresists)
IT UV and visible spectra
 (absorption; free-radical synthesis of narrow polydispersed 2-hydroxyethyl methacrylate-based tetrapolymers for swelling-free dilute

aqueous base developable neg. photoresists)
IT Electron beam lithography
Negative photoresists
(free-radical synthesis of narrow polydispersed 2-hydroxyethyl methacrylate-based tetrapolymers for swelling-free dilute aqueous base developable neg. photoresists)
IT Swelling, physical
(free-radical synthesis of narrow polydispersed 2-hydroxyethyl methacrylate-based tetrapolymers for swelling-free dilute aqueous base developable neg. photoresists in relation to)
IT Etching
(plasma; swelling-free dilute aqueous base developable neg. photoresists in relation to)
IT Polymerization
(radical; free-radical synthesis of narrow polydispersed 2-hydroxyethyl methacrylate-based tetrapolymers for swelling-free dilute aqueous base developable neg. photoresists)
IT 301532-99-0, AR 19
RL: CPS (Chemical process); PEP (Physical, engineering or chemical process); PROC (Process)
(anti-reflective coating; free-radical synthesis of narrow polydispersed 2-hydroxyethyl methacrylate-based tetrapolymers for swelling-free dilute aqueous base developable neg. photoresists)
IT 60-24-2, 2-Mercaptoethanol
RL: CPS (Chemical process); PEP (Physical, engineering or chemical process); PROC (Process)
(chain transfer agent; free-radical synthesis of narrow polydispersed 2-hydroxyethyl methacrylate-based tetrapolymers for swelling-free dilute aqueous base developable neg. photoresists)
IT 75-59-2, Tetramethyl ammonium hydroxide
RL: CPS (Chemical process); PEP (Physical, engineering or chemical process); PROC (Process)
(developer; free-radical synthesis of narrow polydispersed 2-hydroxyethyl methacrylate-based tetrapolymers for swelling-free dilute aqueous base developable neg. photoresists)
IT 374938-51-9P, Acrylic acid-cyclohexyl methacrylate-2-hydroxyethyl methacrylate-isobornyl methacrylate copolymer
RL: CPS (Chemical process); PEP (Physical, engineering or chemical process); PRP (Properties); SPN (Synthetic preparation); PREP (Preparation); PROC (Process)
(free-radical synthesis of narrow polydispersed 2-hydroxyethyl methacrylate-based tetrapolymers for swelling-free dilute aqueous base developable neg. photoresists)
IT 79-10-7, Acrylic acid, reactions 101-43-9, Cyclohexyl methacrylate 868-77-9, 2-Hydroxyethyl methacrylate 7534-94-3, Isobornyl methacrylate
RL: CPS (Chemical process); PEP (Physical, engineering or chemical process); RCT (Reactant); PROC (Process); RACT (Reactant or reagent)
(free-radical synthesis of narrow polydispersed 2-hydroxyethyl methacrylate-based tetrapolymers for swelling-free dilute aqueous base developable neg. photoresists)
IT 25249-16-5, Poly(2-hydroxyethyl methacrylate)

RL: PRP (Properties)
(free-radical synthesis of narrow polydispersed 2-hydroxyethyl methacrylate-based tetrapolymers for swelling-free dilute aqueous base developable neg. photoresists in relation to)

IT 13472-08-7, 2,2'-Azobis(2-methylbutyronitrile)
RL: CAT (Catalyst use); USES (Uses)
(initiator; free-radical synthesis of narrow polydispersed 2-hydroxyethyl methacrylate-based tetrapolymers for swelling-free dilute aqueous base developable neg. photoresists)

IT 57840-38-7, Triphenylsulfonium hexafluoroantimonate
RL: CPS (Chemical process); PEP (Physical, engineering or chemical process); PROC (Process)
(photoacid generator; free-radical synthesis of narrow polydispersed 2-hydroxyethyl methacrylate-based tetrapolymers for swelling-free dilute aqueous base developable neg. photoresists)

IT 104137-08-8, AZ 5214 180513-74-0, UV III
RL: CPS (Chemical process); PEP (Physical, engineering or chemical process); PROC (Process)
(photoresist; free-radical synthesis of narrow polydispersed 2-hydroxyethyl methacrylate-based tetrapolymers for swelling-free dilute aqueous base developable neg. photoresists)

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L8 ANSWER 19 OF 32 CA COPYRIGHT 2008 ACS on STN
AN 136:191685 CA
ED Entered STN: 14 Mar 2002
TI Positively working photoresist composition for far-ultraviolet exposure
IN Nakao, Hajime; Sato, Kenichiro
PA Fuji Photo Film Co., Ltd., Japan
SO Jpn. Kokai Tokkyo Koho, 55 pp.
CODEN: JKXXAF
DT Patent
LA Japanese
IC ICM G03F007-039
ICS C08F232-00; C08K005-00; C08K005-10; C08K005-17; C08K005-372;
C08K005-541; C08L045-00; G03F007-004; H01L021-027
CC 74-5 (Radiation Chemistry, Photochemistry, and Photographic and Other
Reproductive Processes)
Section cross-reference(s): 76
FAN.CNT 1
PATENT NO. KIND DATE APPLICATION NO. DATE

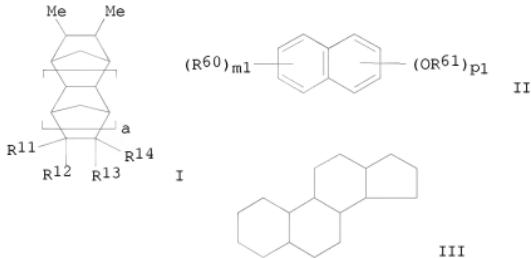
PI JP 2002049154 A 20020215 JP 2000-233146 20000801
PRAI JP 2000-233146 20000801
CLASS
PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES

JP 2002049154 ICM G03F007-039
ICS C08F232-00; C08K005-00; C08K005-10; C08K005-17;
C08K005-372; C08K005-541; C08L045-00; G03F007-004;
H01L021-027
IPCI G03F0007-039 [ICM,7]; C08F0232-00 [ICS,7]; C08K0005-00
[ICS,7]; C08K0005-10 [ICS,7]; C08K0005-17 [ICS,7];

C08L0045-00 C08K0005-372 [ICS, 7]; C08K0005-541 [ICS, 7];
 [ICS, 7]; G03F0007-004 [ICS, 7]; H01L0021-027 [ICS, 7];
 H01L0021-02 [ICS, 7, C*]; G03F0007-039 [I, A]; C08F0232-00
 [I, C*]; C08F0232-00 [I, A]; C08K0005-00 [I, C*];
 C08K0005-00 [I, A]; C08K0005-10 [I, A]; C08K0005-17
 [I, A]; C08K0005-372 [I, A]; C08K0005-54 [I, A];
 C08K0005-541 [I, A]; C08L0045-00 [I, C*]; C08L0045-00
 [I, A]; G03F0007-004 [I, C*]; G03F0007-004 [I, A];
 H01L0021-02 [I, C*]; H01L0021-027 [I, A]

OS MARPAT 136:191685

GI



AB The composition, useful for ultramicroolithog. process in fabrication of ultra-large-scale integrated circuits (ULSI), contains (A) polymers having alicyclic repeating unit I [R11-R14 = H, (substituted) alkyl; a = 0, 1] and [CH₂CR(ACO₂W)] unit [R1 = H, Me; A = none, alkylene, cycloalkylene, O, S, CO, and/or ester; W = CRaRbRc, CHRdORe; Ra-Rc, Re = (halo-, alkoxy-, alkoxycarbonyl, acyl-, or acyloxy-substituted) C1-20 linear or branched alkyl, C3-20 cycloalkyl; Ra and Rb may form an alicyclic ring; Rd = H, alkyl] to increase alkali developability by acids, (B) photoacid generators, and (C) R[X(CR51R52)qCO₂R']n (II; X = O, S, NR53, none; R51-R53 = H, alkyl; R' = acid-degradable group as CO₂R'; R = bridged hydrocarbon, saturated alicyclic compound, naphthalene-containing n-valent residue; n = 1-4; q = 0-10), naphthalene derivs. III (R60 = alkyl, halo-, R61 = acid-degradable group as OR61; m = 0-4; p = 1-4), or a cholic acid derivative having structure IV substituted with ≥2 groups having ≥1 substituent containing carboxyl group protected with acid-unstable group. The compds. II-IV work as dissoln. inhibitors and the composition gives high-resolution contact hole and trench patterns in fabrication of

FOR 10559534 by Cynthia Hamilton

semiconductor devices.

ST pos photoresist far UV dissoln inhibitor; contact hole trench pattern photoresist pos

IT Polysiloxanes, uses

RL: TEM (Technical or engineered material use); USES (Uses)
(KP 341, surfactant; pos.-working photoresist composition for far-UV exposure)

IT Positive photoresists
(UV; pos.-working photoresist composition for far-UV exposure)

IT 24556-20-5 156301-83-6 169228-97-1 184633-80-5 198548-99-1
202654-70-4 265119-61-7 332136-74-0 399041-03-3 399041-04-4
399041-05-5 399041-06-6

RL: TEM (Technical or engineered material use); USES (Uses)
(dissoln. inhibitor; pos.-working photoresist composition for far-UV exposure)

IT 251949-14-1P, tert-Butyl cholate-glutaryl dichloride copolymer
321994-64-3P

RL: PNU (Preparation, unclassified); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)
(oligomeric, dissoln. inhibitor; pos.-working photoresist composition for far-UV exposure)

IT 14159-45-6 66003-78-9 135539-92-3 138529-81-4 144089-15-6
144317-44-2 153698-46-5 197447-16-8 197447-17-9 241806-75-7
258341-99-0 307531-76-6 307976-40-5 312386-77-9 391232-40-9

RL: CAT (Catalyst use); TEM (Technical or engineered material use); USES (Uses)
(photoacid generator; pos.-working photoresist composition for far-UV exposure)

IT 260448-02-0P, tert-Butyl acrylate-maleic anhydride-norbornene copolymer 369371-67-5P 383196-78-9P 383196-80-3P 383196-82-5P
383196-83-6P 383196-85-8P 383196-87-0P 383196-88-1P 383196-89-2P
383196-91-6P 383196-93-8P 383196-95-0P 391232-36-3P 391232-38-5P

RL: PNU (Preparation, unclassified); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)
(pos.-working photoresist composition for far-UV exposure)

IT 484-47-9, 2,4,5-Triphenylimidazole 1122-58-3 6674-22-2, DBU

RL: TEM (Technical or engineered material use); USES (Uses)
(pos.-working photoresist composition for far-UV exposure)

IT 9016-45-9, Polyoxyethylene nonylphenyl ether 137462-24-9, Megafac F 176
216679-67-3, Megafac R 08

RL: TEM (Technical or engineered material use); USES (Uses)
(surfactant; pos.-working photoresist composition for far-UV exposure)

L8 ANSWER 20 OF 32 CA COPYRIGHT 2008 ACS on STN

AN 136:110037 CA

ED Entered STN: 07 Feb 2002

TI New polymer for 157-nm single-layer resist based on fluorine-containing acryl copolymer

AU Ogata, Toshiyuki; Endo, Koutaro; Komano, Hiroshi; Nakayama, Toshimasa

CS Advanced Materials Development Division 1, Tokyo Ohka Kogyo Co., Ltd.,
Samukawa-cho, Koza-gun, Kawasaki-shi, 253-0114, Japan

SO Proceedings of SPIE-The International Society for Optical Engineering

(2001), 4345(Pt. 2, Advances in Resist Technology and Processing XVIII), 1048-1055
CODEN: PSISDG; ISSN: 0277-786X
PB SPIE-The International Society for Optical Engineering
DT Journal
LA English
CC 74-5 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)
AB Development is reported of acrylic polymer based on methacrylate and acrylate monomers containing various trifluoromethyl groups for the application to 157 nm chemical amplified pos.-tone resists. The authors developed a novel monomer, trifluoromethyl-iso-adamantyl methacrylate (TFIAmdMA) and a new co-polymer being a combination of fluorinated methacrylate derivs. and substituted p-hydroxystyrene. The absorption coefficient of poly(p-tert-butoxystyrene-hexafluoro-tert-Bu methacrylate-co-methacrylic acid) was <3 μm^{-1} at 157 nm. Patterning was done with 157 nm contact exposure system of VUVES-4500 by LTJ. One of the exptl. resists, based on a particular polymer ratio and photoacid generator, has clearly achieved 180 nm line and space pattern resolution. At 140 nm resist film thickness, the sensitivity was 31 mJ/cm² when using 0.26 N tetramethylammonium hydroxide surfactant developer.
ST vacuum UV chem amplification photoresist fluorine substituted acrylic copolymer
IT Fluoropolymers, properties
RL: PRP (Properties); TEM (Technical or engineered material use); USES (Uses)
 (acrylic; lithog. characterization of fluorine-containing acrylic polymers
 for 157-nm single-layer chemical amplification photoresist)
IT Photoresists
 (chemical amplification; polymer for 157-nm single-layer resist based on
 fluorine-containing acrylic copolymer)
IT Absorptivity
 (lithog. characterization of fluorine-containing acrylic polymers for
 157-nm single-layer chemical amplification photoresist)
IT 75-59-2, Tetramethylammonium hydroxide
RL: NUU (Other use, unclassified); USES (Uses)
 (developer; etch rate of fluorine-containing acrylic polymers for
157-nm
 single-layer chemical amplification photoresist)
IT 388613-62-5 388613-63-6
RL: PRP (Properties); TEM (Technical or engineered material use); USES (Uses)
 (lithog. characterization of fluorine-containing acrylic polymers for
 157-nm single-layer chemical amplification photoresist)
IT 75-46-7, Trifluoromethane 75-73-0, Carbon tetrafluoride
RL: NUU (Other use, unclassified); USES (Uses)
 (plasma; etch rate of fluorine-containing acrylic polymers for 157-nm
 single-layer chemical amplification photoresist)
IT 28825-23-2 354818-15-8 388613-61-4
RL: PRP (Properties)
 (polymers for 157-nm single-layer chemical amplification photoresist
based

on fluorine-containing acrylic copolymer)
IT 388613-59-0P
RL: PNU (Preparation, unclassified); RCT (Reactant); PREP (Preparation);
RACT (Reactant or reagent)
(polymers photoresist application in for 157 nm lithog. containing
copolymers of novel monomer of trifluoromethyl-iso-adamantyl
methacrylate)
RE.CNT 9 THERE ARE 9 CITED REFERENCES AVAILABLE FOR THIS RECORD
RE
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L8 ANSWER 21 OF 32 CA COPYRIGHT 2008 ACS on STN
AN 136:12719 CA
ED Entered STN: 27 Dec 2001
TI Negative (meth)acrylate resist materials based on novel
crosslinking chemistry
AU Diakoumakos, C. D.; Raptis, I.; Tserepi, A.; Argitis, P.
CS NCSR "Demokritos", Institute of Microelectronics, Athens, Aghia
Paraskevi,
15310, Greece
SO Microelectronic Engineering (2001), 57-58, 539-545
CODEN: MIENEF; ISSN: 0167-9317
PB Elsevier Science B.V.
DT Journal
LA English
CC 74-5 (Radiation Chemistry, Photochemistry, and Photographic and Other
Reproductive Processes)
AB Novel neg.-tone resist materials suitable for 193 nm and e-beam lithog.,
based on newly-synthesized (meth)acrylate copolymers that
contain (2-hydroxyethyl methacrylate) as the necessary component for
imaging and aqueous base development, are presented. A representative
resist
of this class is formulated using poly(2-hydroxyethyl methacrylate-
co-cyclohexyl methacrylate-co-isobornyl methacrylate-
co-acrylic acid) (PHECIMA) and a sulfonium salt photo acid
generator. The neg. image formation is based on acid induced
crosslinking
of the hydroxyl groups of the 2-hydroxyethyl methacrylate (HEMA)
moieties.
The neg. resist presented no swelling phenomena in the aqueous base
developer
and enhanced etch resistance. Dense 0.18 μ m and isolated 0.13 μ m
lines have been obtained upon 193 nm and e-beam lithog. establishing
PHECIMA's resist formulations as promising candidates for high-resolution
lithog. upon a further material and process optimization.
ST electron beam lithog deep UV resist hydroxyethyl methacrylate copolymer;
acrylate copolymer neg resist electron beam lithog; photolithog UV

FOR 10559534 by Cynthia Hamilton

hydroxyethyl methacrylate copolymer neg resist; photoresist neg deep UV
hydroxyethyl methacrylate copolymer
IT Photolithography
(UV; neg. methacrylate resist materials based on novel crosslinking
chemical)
IT Electron beam lithography
Negative photoresists
(neg. methacrylate resist materials based on novel crosslinking
chemical)
IT 75-59-2, Tetramethyl ammonium hydroxide
RL: PEP (Physical, engineering or chemical process); PROC (Process)
(developer; neg. methacrylate resist materials based on novel
crosslinking chemical)
IT 374938-51-9P, Acrylic acid-cyclohexyl methacrylate-2-hydroxyethyl
methacrylate-isobornyl methacrylate copolymer
RL: PEP (Physical, engineering or chemical process); PRP (Properties);
SPN
(Synthetic preparation); PREP (Preparation); PROC (Process)
(neg. methacrylate resist materials based on novel crosslinking
chemical)
IT 57840-38-7, Triphenylsulfonium hexafluoroantimonate
RL: PEP (Physical, engineering or chemical process); PROC (Process)
(photoacid generator; neg. methacrylate resist materials
based on novel crosslinking chemical)
RE.CNT 8 THERE ARE 8 CITED REFERENCES AVAILABLE FOR THIS RECORD
RE
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(7) Naito, T; Proc SPIE 1998, V3333, P503 CA
(8) Raptis, I; Microelectron Eng 2000, V53, P589

L8 ANSWER 22 OF 32 CA COPYRIGHT 2008 ACS on STN
AN 135:325178 CA
ED Entered STN: 15 Nov 2001
TI Effect of comonomer structure on dissolution characteristics: ArF
negative
resist system using androsterone derivative with δ -hydroxy acid
AU Yokoyama, Yoshiyuki; Hattori, Takashi; Kimura, Kaori; Tanaka, Toshihiko;
Shiraishi, Hiroshi
CS Central Research Laboratory, Hitachi Ltd., Tokyo, 185-8601, Japan
SO Journal of Photopolymer Science and Technology (2001), 14(3), 393-400
CODEN: JSTEW; ISSN: 0914-9244
PB Technical Association of Photopolymers, Japan
DT Journal
LA English
CC 74-5 (Radiation Chemistry, Photochemistry, and Photographic and Other
Reproductive Processes)
AB A neg. resist system utilizing acid-catalyzed intramol. esterification of
 δ -hydroxy acid has been developed for ArF phase-shifting lithog.
The system is made up of an acrylate polymer with pendant
structure of androsterone derivative with δ -hydroxy acid and a

photoacid generator. The authors investigated the effect of the comonomer and found that it changes the affinity of the resist polymer to the aqueous base developer. The change of the polarity of the comonomer was found to drastically affect the dissoln. properties and the resolution capability. Optimization of the δ -hydroxy acid content and the developer concentration prevented pattern deformation such as "winding lines" and scum between the lines. The improved resist formulation combined with an ArF excimer-laser stepper with a phase-shifting mask produced a clearly resolved 100-nm line-and-space patterns.

ST monomer structure effect dissoln photoresist androsterone deriv hydroxy acid

IT Dissolution rate
Molecular structure-property relationship
Negative photoresists
(co-monomer structure effect on dissoln. characteristics of neg. photoresist system using acrylate polymer with pendant of androsterone derivative with δ -hydroxy acid group)

IT 367947-20-4DP, hydrolyzed 367947-21-5DP, hydrolyzed
RL: PRP (Properties); SPN (Synthetic preparation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)
(co-monomer structure effect on dissoln. characteristics of neg. photoresist system using acrylate polymer with pendant of androsterone derivative with δ -hydroxy acid group)

IT 75-59-2, Tetramethyl ammonium hydroxide
RL: NUU (Other use, unclassified); USES (Uses)
(developer; co-monomer structure effect on dissoln. characteristics of neg. photoresist system using acrylate polymer with pendant of androsterone derivative with δ -hydroxy acid group)

IT 367947-19-1DP, hydrolyzed
RL: PRP (Properties); SPN (Synthetic preparation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)
(dissoln. characteristics of neg. photoresist system using acrylate polymer with pendant of androsterone derivative with δ -hydroxy acid group)

IT 367947-17-9P
RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)
(oxidation reaction in synthesis of acrylate polymer with pendant structure of androsterone derivative with δ -hydroxy acid group)

IT 66003-78-9, Triphenylsulfonium triflate
RL: NUU (Other use, unclassified); USES (Uses)
(photoacid generator; co-monomer structure effect on dissoln. characteristics of neg. photoresist system using acrylate polymer with pendant of androsterone derivative with δ -hydroxy acid group)

IT 367947-18-0P
RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)
(synthesis of acrylate polymer with pendant structure of androsterone derivative with δ -hydroxy acid group)

RE.CNT 33 THERE ARE 33 CITED REFERENCES AVAILABLE FOR THIS RECORD

RE

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- (30) Tsuchiya, Y; J Photopolym Sci Technol 1997, V10, P579 CA
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L8 ANSWER 23 OF 32 CA COPYRIGHT 2008 ACS on STN

AN 135:325173 CA

ED Entered STN: 15 Nov 2001

TI Rejuvenation of 248 nm resist backbones for 157 nm lithography

AU Bae, Young C.; Douki, Katsuji; Yu, Tianyue; Dai, Junyan; Schmaljohann, Dirk; Kang, Seok Ho; Kim, Keon Hyeong; Koerner, Hilmar; Conley, Will; Miller, Daniel; Balasubramanian, Raghu; Holl, Susan; Ober, Christopher K.

CS Department of Materials Science & Engineering, Cornell University, Ithaca, NY, 14853, USA

SO Journal of Photopolymer Science and Technology (2001), 14(4), 613-620
CODEN: JSTEEW; ISSN: 0914-9244

PB Technical Association of Photopolymers, Japan

DT Journal

LA English

CC 74-5 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)

AB Fluorocarbonol-based acrylic and styrenic monomers, such as 2-[4-(2-hydroxyhexafluoroisopropyl)cyclohexane]hexafluoroisopropyl acrylate (1) 2-[4-(2,2,2-trifluoro-1-methoxymethoxy-1-

trifluoromethyl ethyl) cyclohexane]h exafluoroisopropyl acrylate (2), and 2-[4-(2,2,2-trifluoro-1-ethoxy methoxy-1-trifluoromethyl ethyl) styrene (3), were synthesized and their (co)polymers were studied as photoresist platforms for 157 nm lithog. It was

found that these (co)polymers are unusually transparent at 157 nm, and absorbances of poly(1) and poly(1-co-3) were determined to be 1.93 and 2.38 μm^{-1} , resp. It was also found that further improvement in transparency is possible by adding transparency enhancers to the resist

platform. Lithog. studies were carried out with poly(1)- and poly(3)-based resists using 157 and 248 nm steppers, and it was proven that, after selective modification, it is possible to use conventional resist backbones, such as acrylic or styrenic, in the design of single-layer resists for 157 nm lithog.

ST vacuum UV lithog photoresist fluorocarbonol based polymer; fluorocarbonol based acrylic styrenic monomer copolymer photolithog photoresist

IT Photoresists

Thermal stability

UV and visible spectra

(chemical amplified photoresist for 157 nm lithog. based on homo- and copolymers of fluorocarbonol-based acrylic and styrenic monomers)

IT 367522-48-3P

RL: PRP (Properties); SPN (Synthetic preparation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(chemical amplified photoresist for 157 nm lithog. based on homo- and copolymers of fluorocarbonol-based acrylic and styrenic monomers)

IT 367522-49-4 367522-50-7 367522-52-9

RL: PRP (Properties); TEM (Technical or engineered material use); USES (Uses)

(chemical amplified photoresist for 157 nm lithog. based on homo- and copolymers of fluorocarbonol-based acrylic and styrenic monomers)

IT 75-59-2, Tetramethylammonium hydroxide

RL: NUU (Other use, unclassified); USES (Uses)

(developer; chemical amplified photoresist for 157 nm lithog. based on homo- and copolymers of fluorocarbonol-based acrylic and styrenic monomers)

IT 144317-44-2

RL: TEM (Technical or engineered material use); USES (Uses)
(photoacid generator; chemical amplified photoresist for 157 nm lithog. based on homo- and copolymers of fluorocarbonol-based acrylic and styrenic monomers)

IT 367522-45-0P,

2-[4-(2-Hydroxyhexafluoroisopropyl)cyclohexane]hexafluoroisopropyl acrylate 367522-46-1P, 2-[4-(2,2,2-Trifluoro-1-methoxy methoxy-1-trifluoromethyl ethyl)cyclohexane]h exafluoroisopropyl acrylate 367522-47-2P, 2-[4-(2,2,2-Trifluoro-1-ethoxy methoxy-1-trifluoromethyl ethyl)styrene

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)
(synthesis of fluorocarbonol-based acrylic and styrenic monomers for preparation of photoresist copolymers for 157 nm lithog.)

RE.CNT 11 THERE ARE 11 CITED REFERENCES AVAILABLE FOR THIS RECORD

RE

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FOR 10559534 by Cynthia Hamilton

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L8 ANSWER 24 OF 32 CA COPYRIGHT 2008 ACS on STN

AN 135:218725 CA

ED Entered STN: 27 Sep 2001

TI Positive-working far-UV photoresist composition containing sulfonium salts

as photoacid generators

IN Sato, Kenichiro; Mizutani, Kazuyoshi

PA Fuji Photo Film Co., Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 35 PP.

CODEN: JKXXAF

DT Patent

LA Japanese

IC ICM G03F007-039

ICS C08G061-08; C08K005-42; C08L065-00; G03F007-004; H01L021-027

CC 74-5 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)

Section cross-reference(s): 76

FAN.CNT 1

PATENT NO.

KIND

DATE

APPLICATION NO.

DATE

PI JP 2001235869

A

20010831

JP 2000-47971

PRAI JP 2000-47971

20000224

CLASS

PATENT NO.

CLASS

PATENT FAMILY CLASSIFICATION CODES

JP 2001235869

ICM G03F007-039

ICS C08G061-08; C08K005-42; C08L065-00; G03F007-004;

H01L021-027

IPCI G03F0007-039 [ICM,7]; C08G0061-08 [ICS,7]; C08G0061-00

[ICS,7,C*]; C08K005-42 [ICS,7]; C08K005-00

[ICS,7,C*]; C08L0065-00 [ICS,7]; G03F0007-004 [ICS,7];

H01L0021-027 [ICS,7]; H01L0021-02 [ICS,7,C*]

IPCR G03F0007-039 [I,C*]; G03F0007-039 [I,A]; C08G0061-00

[I,C*]; C08G0061-08 [I,A]; C08K0005-00 [I,C*];

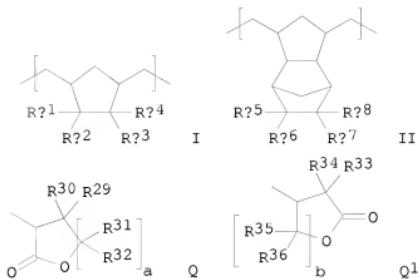
C08K0005-42 [I,A]; C08L0065-00 [I,C*]; C08L0065-00

[I,A]; G03F0007-004 [I,C*]; G03F0007-004 [I,A];

H01L0021-02 [I,C*]; H01L0021-027 [I,A]

OS MARPAT 135:218725

GI



AB The compns., which are storage stable and show high sensitivity and resolution in formation of contact hole pattern in semiconductor device fabrication, contain (A) $S+R1R2R3\text{Z-}$ [$R1-R3$ = (un)substituted alkyl, (un)substituted aryl; 2 of $R1-R3$ may bonded together to via direct bond or substituent; $Z-$ = counter anion] which generate acids by irradiation with actinic ray or radiation and (B) a resin which is decomposed with acids to become alkali-soluble and contains ≥ 1 repeating unit selected from I [$R1-R4$ = H, (un)substituted alkyl, (un)substituted cyclic hydrocarbyl, halo, cyano, CO_2H , COYARc_9 , $\text{COYACO}_2(\text{CH}_2)2\text{SiR}'\text{R}''\text{R}'''$, CO_2Rc_1 , $\text{CO}_2(\text{CH}_2)2\text{SiR}'\text{R}''\text{R}'''$; R' , R'' , R''' = alkyl, trialkylsilyl, trialkylsilyloxy; Y = O, S, NH, NHSO_2 , NHSO_2NH ; Rc_9 = CO_2H , $\text{CO}_2\text{Rc}_{10}$ (Rc_{10} = any group given for Rc_{11} , Q, Q1), cyano, OH, (un)substituted alkoxy, CONHRc_{11} , $\text{CONHSO}_2\text{Rc}_{11}$, Q, Q1; Rc_{11} = (un)substituted alkyl, (un)substituted cycloalkyl; A = direct bond, (un)substituted alkylene, ether, thioether, CO, ester, amido, etc.; $R29-R36$ = H, (un)substituted alkyl; a, b = 1, 2; ≥ 1 od $Rc1-Rc4$ = ≥ 1 of $Rc1-Rc4$ = $\text{COYACO}_2(\text{CH}_2)2\text{SiR}'\text{R}''\text{R}'''$ or $\text{CO}_2(\text{CH}_2)2\text{SiR}'\text{R}''\text{R}'''$] and II ($Rc5-Rc8$ = any group given for $Rc1-Rc4$).

ST far UV pos photoresist sulfonium photoacid generator; semiconductor device far UV pos photoresist; silylethoxycarbonyl contg resin far UV pos photoresist

IT Positive photoresists
(UV; pos.-working far-UV photoresist composition containing sulfonium salts as photoacid generators and silylethoxycarbonyl-containing resins)

IT Semiconductor device fabrication
(pos.-working far-UV photoresist composition containing sulfonium salts as photoacid generators and silylethoxycarbonyl-containing resins)

IT Sulfonium compounds
RL: CAT (Catalyst use); USES (Uses)
(pos.-working far-UV photoresist composition containing sulfonium salts as photoacid generators and silylethoxycarbonyl-containing resins)

FOR 10559534 by Cynthia Hamilton

IT 66003-78-9 144089-15-6 153698-46-5 206861-54-3 241806-75-7
258341-99-0 260061-58-3 279218-75-6 301525-08-6 350251-56-8
350251-57-9 357444-19-0 357444-20-3
RL: CAT (Catalyst use); USES (Uses)
(pos.-working far-UV photoresist composition containing sulfonium salts as
photoacid generators and silylethoxycarbonyl-containing resins)
IT 46276-02-2P 327023-46-1P
RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation);
RACT
(Reactant or reagent)
(pos.-working far-UV photoresist composition containing sulfonium salts as
photoacid generators and silylethoxycarbonyl-containing resins)
IT 351195-80-7DP, ring-opening polymerization
RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)
(pos.-working far-UV photoresist composition containing sulfonium salts as
photoacid generators and silylethoxycarbonyl-containing resins)
IT 542-92-7, Cyclopentadiene, reactions 3121-61-7, 2-Methoxyethyl acrylate 131494-24-1, 2-(Trimethylsilyl)ethyl acrylate
RL: RCT (Reactant); RACT (Reactant or reagent)
(pos.-working far-UV photoresist composition containing sulfonium salts as
photoacid generators and silylethoxycarbonyl-containing resins)
IT 351195-81-8D, ring-opening polymerization 351195-82-9D, ring-opening polymerization
351195-84-1D, ring-opening polymerization 357444-12-3D, ring-opening polymerization
357444-15-6 357444-17-8D, ring-opening polymerization
RL: TEM (Technical or engineered material use); USES (Uses)
(pos.-working far-UV photoresist composition containing sulfonium salts as
photoacid generators and silylethoxycarbonyl-containing resins)

L8 ANSWER 25 OF 32 CA COPYRIGHT 2008 ACS on STN
AN 135:129573 CA
ED Entered STN: 16 Aug 2001
TI Deep UV positive photoresist compositions containing norbornene- or dicyclopentadiene-based polymers
IN Mizutani, Kazuyoshi
PA Fuji Photo Film Co., Ltd., Japan
SO Jpn. Kokai Tokkyo Koho, 30 pp.
CODEN: JKXXAF
DT Patent
LA Japanese
IC ICM G03F007-039
ICS G03F007-004; G03F007-095; G03F007-26; H01L021-027
CC 74-5 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)
FAN.CNT 1
PATENT NO. KIND DATE APPLICATION NO. DATE

PI JP 2001201855 A 20010727 JP 2000-8239 20000117

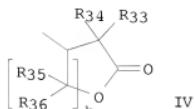
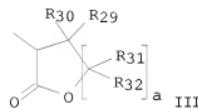
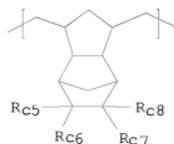
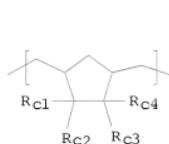
PRAI JP 2000-8239

20000117

CLASS

PATENT NO.	CLASS	PATENT FAMILY CLASSIFICATION CODES
JP 2001201855	ICM	G03F007-039
	ICS	G03F007-004; G03F007-095; G03F007-26; H01L021-027
	IPCI	G03F0007-039 [ICM,7]; G03F0007-004 [ICS,7]; G03F0007-095 [ICS,7]; G03F0007-26 [ICS,7];
H01L0021-027		[ICS,7]; H01L0021-02 [ICS,7,C*]
	IPCR	H01L0021-02 [I,C*]; H01L0021-027 [I,A]; G03F0007-004 [I,C*]; G03F0007-004 [I,A]; G03F0007-039 [I,C*]; G03F0007-039 [I,A]; G03F0007-095 [I,C*]; G03F0007-095 [I,A]; G03F0007-26 [I,C*]; G03F0007-26 [I,A]

GI



AB The photoresist compns. contain (A) active light- or radiation-sensitive acid generators and (B) resins whose solubilities into alkaline solns. are

increased by acidolysis and which involve repeating units norbornene derivs. I and/or dicyclopentadiene derivs. II [Rc1-Rc8 = H, (substituted) alkyl, (substituted) cyclohydrocarbyl, halo, cyano, CO2H, C(O)YARc9, C(O)YACO2(CH2)2SiR1R2R3, CO2Rc11, CO2(CH2)2SiR1R2R3; ≥ 1 of Rc1-Rc4 = C(O)YACO2(CH2)2SiR1R2R3 or CO2(CH2)2SiR1R2R3; ≥ 1 of Rc5-Rc8 = C(O)YACO2(CH2)2SiR1R2R3 or CO2(CH2)2SiR1R2R3; R1-R3 = alkyl, trialkylsilyl, trialkylsilyloxy; Y = O, S, NH, NHSO2, NHSO2NH; Rc9 = CO2H, CO2Rc10 (Rc10 = same as Rc11 or lactones III or IV), CN, OH, (substituted) alkoxy, CONHRc11, CONHSO2Rc11, or lactones III or IV; Rc11 = (substituted) alkyl, (substituted) cycloalkyl; A = single bond; alkylene, substituted alkylene, O, S, CO, CO2, amide, sulfonamide, urethane, urea; R29-R36 = H, alkyl; a, b = 1, 2]. The compns. may further

contain (C) organic bases, (D) silicone-based, F-containing, or nonionic surfactants and (E) organic solvents. In the bilayer resist process, pattern shift on pattern transfer to underlayers while O plasma etching is minimized. Its pattern formation on i-ray resist coated on a Si wafer by exposing to ArF excimer laser was exemplified.

ST deep UV pos. photoresist norbornene polymer; cyclopentadiene trimethylsilyl ethyl acrylate reaction polymer photoresist; methoxyethyl acrylate cyclopentadiene reaction polymer photoresist; dicyclopentadiene polymer deep UV pos. photoresist; argon fluorine excimer laser photoresist

IT Polysiloxanes, uses
RL: MOA (Modifier or additive use); USES (Uses)
(KP 341, surfactants; deep UV pos. photoresist compns. containing norbornene- or dicyclopentadiene-based polymers)

IT Positive photoresists
(UV; deep UV pos. photoresist compns. containing norbornene- or dicyclopentadiene-based polymers)

IT Cycloalkenes
RL: PNU (Preparation, unclassified); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)
(polymers; deep UV pos. photoresist compns. containing norbornene- or dicyclopentadiene-based polymers)

IT 351195-80-7DP, hydrogenated
RL: PNU (Preparation, unclassified); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)
(deep UV pos. photoresist compns. containing norbornene- or dicyclopentadiene-based polymers)

IT 351195-81-8D, hydrogenated 351195-82-9D, hydrogenated 351195-84-1D, hydrogenated
RL: TEM (Technical or engineered material use); USES (Uses)
(deep UV pos. photoresist compns. containing norbornene- or dicyclopentadiene-based polymers)

IT 57840-38-7 66003-76-7 66003-78-9 144089-15-6 153698-46-5
335385-79-0 335385-81-4 335385-82-5
RL: CAT (Catalyst use); USES (Uses)
(photoacid generator; deep UV pos. photoresist compns. containing norbornene- or dicyclopentadiene-based polymers)

IT 484-47-9, 2,4,5-Triphenylimidazole 1122-58-3, 4-Dimethylaminopyridine 6674-22-2, 1,8-Diazabicyclo[5.4.0]undec-7-ene
RL: MOA (Modifier or additive use); USES (Uses)
(polymer dissoln. promoters; deep UV pos. photoresist compns. containing norbornene- or dicyclopentadiene-based polymers)

IT 121-46-0, Norbornadiene 3121-61-7, 2-Methoxyethyl acrylate 131494-24-1, 2-(Trimethylsilyl)ethyl acrylate
RL: RCT (Reactant); RACT (Reactant or reagent)
(starting materials for monomer preparation; deep UV pos. photoresist compns. containing norbornene- or dicyclopentadiene-based polymers)

IT 9016-45-9, Poly(oxyethylene) nonylphenyl ether 137462-24-9, Megafac F 176 216679-67-3, Megafac R 08
RL: MOA (Modifier or additive use); USES (Uses)
(surfactants; deep UV pos. photoresist compns. containing norbornene- or

FOR 10559534 by Cynthia Hamilton

dicyclopentadiene-based polymers)

L8 ANSWER 26 OF 32 CA COPYRIGHT 2008 ACS on STN
AN 135:114439 CA
ED Entered STN: 09 Aug 2001

TI Positive-working far-UV sensitive photoresist composition containing specific acid-sensitive resin and specific additives
IN Adegawa, Yutaka; Sato, Kenichiro; Kodama, Kunihiko; Aogo, Toshiaki
PA Fuji Photo Film Co., Ltd., Japan
SO Jpn. Kokai Tokkyo Koho, 93 pp.
CODEN: JKXXAF

DT Patent

LA Japanese

IC ICM G03F007-039
ICS C08F222-00; C08F232-00; G03F007-004; H01L021-027
CC 74-5 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)
Section cross-reference(s): 35

FAN.CNT 1

PATENT NO. KIND DATE APPLICATION NO. DATE

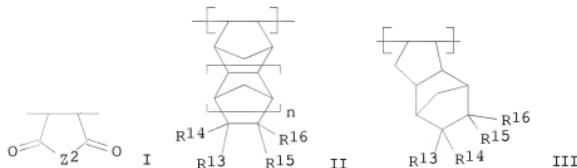
PI JP 2001194786 A 20010719 JP 1999-317407 19991108
PRAI JP 1999-197911 A 19990712
JP 1999-302917 A 19991025

CLASS

PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES

JP 2001194786 ICM G03F007-039
ICS C08F222-00; C08F232-00; G03F007-004; H01L021-027
IPCI G03F0007-039 [ICM, 7]; C08F0222-00 [ICS, 7]; C08F0232-00 [ICS, 7]; G03F0007-004 [ICS, 7]; H01L0021-027 [ICS, 7]
IPCR H01L0021-02 [I,C*]; H01L0021-027 [I,A]; C08F0222-00 [I,C*]; C08F0232-00 [I,A]; C08F0222-00 [I,A]; G03F0007-004 [I,A]; G03F0007-039 [I,C*]; G03F0007-039 [I,A]

GI



AB The title composition contains a photoacid generator, copolymer having repeating unit chosen from $[-CH(CO-X-A-R1)-CH(CO-X-A-R2)-]$ (R1-2 = H, cyano, OH, etc.; X = O, S, -NH-, etc.; A = single bond, 2-valent connecting group) and I (Z2 = -O-, amino), and a repeating

unit chosen from II and III (R13-16 = H, halo, cyano, one of R13-16 must be adamantyl acrylate). The composition, which contains the aforementioned acid-sensitive resin and the additives, generates little faulty development of the resist layer without forming scum.

ST pos working photoresist compn light sensitive resin additive

IT Cycloalkenes

RL: SPN (Synthetic preparation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)
(polymers; pos.-working photoresist composition containing specific acid-sensitive resin and specific additives)

IT Photoresists

(pos.-working photoresist composition containing specific acid-sensitive resin and specific additives)

IT 331747-08-1P 331747-10-5P 331747-12-7P 331747-14-9P 331747-16-1P
331747-18-3P 331747-20-7P 331747-21-8P 350498-51-0P 350498-52-1P
350498-55-4P

RL: PNU (Preparation, unclassified); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)
(acid-sensitive resin in pos.-working photoresist composition)

IT 66003-78-9, Triphenylsulfonium triflate 288303-62-8 288303-65-1
288303-68-4 288303-71-9 288303-73-1 288303-75-3 288303-77-5
301525-08-6

RL: TEM (Technical or engineered material use); USES (Uses)
(photoacid generator in pos.-working photoresist composition)

IT 50-00-0, Formaldehyde, reactions 74-88-4, Methyl iodide, reactions 93-11-8, 2-Naphthalenesulfonyl chloride 107-21-1, Ethylene glycol, reactions 108-67-8, Mesitylene, reactions 126-81-8, Dimedone 832-53-1, Pentafluorobenzenesulfonyl chloride 945-51-7, Diphenyl sulfoxide 1694-31-1, tert-Butyl acetoacetate 2033-24-1 21286-54-4, (+)-Camphorsulfonyl chloride 29420-49-3, Potassium nonafluorobutanesulfonate 31017-40-0, Phenylcyclohexene

RL: RCT (Reactant); RACT (Reactant or reagent)

(pos.-working photoresist composition containing specific acid-sensitive resin and specific additives)

IT 331747-03-6P

RL: SPN (Synthetic preparation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)
(pos.-working photoresist composition containing specific acid-sensitive resin and specific additives)

L8 ANSWER 27 OF 32 CA COPYRIGHT 2008 ACS on STN

AN 135:68557 CA

ED Entered STN: 19 Jul 2001

TI Photolithography and its chemically-amplified photoresists containing specific sulfonyldiazomethane compounds

IN Seki, Akihiro; Takemura, Katsuya; Osawa, Yoichi; Watanabe, Atsushi; Nagura, Shigehiro

PA Shin-Etsu Chemical Industry Co., Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 49 pp.

CODEN: JKXXAF

DT Patent

FOR 10559534 by Cynthia Hamilton

LA Japanese
 IC ICM G03F007-004
 ICS C07C381-14; C08K005-09; C08K005-13; C08K005-16; C08K005-41;
 C08K005-43; C08L025-02; C08L025-18; C08L033-02; C08L033-04;
 C08L035-00; G03F007-039; G03F007-26
 CC 74-5 (Radiation Chemistry, Photochemistry, and Photographic and Other
 Reprographic Processes)
 Section cross-reference(s): 25, 37

FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI JP 2001174984	A	20010629	JP 2000-294695	20000927
JP 3750725	B2	20060301		
US 6395446	B1	20020528	US 2000-680481	20001005
PRAI JP 1999-285450	A	19991006		

CLASS

PATENT NO.	CLASS	PATENT FAMILY CLASSIFICATION CODES
JP 2001174984	ICM	G03F007-004
	ICS	C07C381-14; C08K005-09; C08K005-13; C08K005-16; C08K005-41; C08K005-43; C08L025-02; C08L025-18; C08L033-02; C08L033-04; C08L035-00; G03F007-039; G03F007-26
	IPCI	G03F007-004 [I,A]; C07C0381-14 [I,A]; C07C0381-00 [I,C*]; C08K005-09 [I,A]; C08K005-13 [I,A]; C08K005-16 [I,A]; C08K005-41 [I,A]; C08K005-43 [I,A]; C08K005-00 [I,C*]; C08L025-02 [I,A]; C08L025-18 [I,A]; C08L025-00 [I,C*]; C08L033-02 [I,A]; C08L033-04 [I,A]; C08L033-00 [I,C*]; C08L0035-00 [I,A]; G03F007-039 [I,A]
	IPCR	G03F007-004 [I,C*]; G03F007-004 [I,A]; C07C0381-00 [I,C*]; C07C0381-14 [I,A]; C08K005-00 [I,C*]; C08K005-09 [I,A]; C08K005-13 [I,A]; C08K005-16 [I,A]; C08K005-41 [I,A]; C08K005-43 [I,A]; C08L0025-00 [I,C*]; C08L0025-02 [I,A]; C08L0025-18 [I,A]; C08L0033-00 [I,C*]; C08L0033-02 [I,A]; C08L0033-04 [I,A]; C08L0035-00 [I,C*]; C08L0035-00 [I,A]; G03F0007-039 [I,C*]; G03F0007-039 [I,A]; G03F0007-26 [I,C*]; G03F0007-26 [I,A]
US 6395446	IPCI	G03F0007-004 [ICM,7]; G03F0007-30 [ICS,7]
	IPCR	G03F0007-004 [I,C*]; G03F0007-004 [I,A]; G03F0007-038 [I,C*]; G03F0007-038 [I,A]; G03F0007-039 [I,C*]; G03F0007-039 [I,A]; G03F0007-16 [I,C*]; G03F0007-16 [I,A]
	NCL	430/170.000; 430/270.100; 430/326.000; 430/330.000; 430/905.000
	ECLA	G03F007/004D; G03F007/038C; G03F007/039C; G03F007/16Z

OS MARPAT 135:68557

AB The photoresists contain (i) [C6H5-p-(R1CO2)qR2pSO2]nC:N2(GR3)m (R1, R3

=

C1-10 alkyl, C6-14 aryl; R2 = C1-6 alkyl; G = SO₂, CO; p = 0-4
 integer; q = 1-5 integer; 1 ≤ p + q ≤ 5; n = 1, 2; m = 0, 1;
 m + n = 2) or (ii) R1CO2-p-C6H4SO2C1:N2SO2-p-C6H4OCOR1 (R1 = the same
 definition as above) as photoacid generators. The photoresists
 may comprise (α-methyl-p-hydroxystyrene-(meth) acrylate

ester copolymers with M_w 3,000-100,000 containing ≤ 80 ($\neq 0$)-mol% acid-labile substituents. Markush structures for preferable acid-labile substituents are given. Photolithog. employing the photoresists and ≤ 300 -nm high-energy beam or electron beam is also claimed. The photoresists show excellent post-development profiles.

ST sulfonylazomethane photoacid generator chem amplified photoresist; hydroxystyrene methacrylate copolymer chem amplified photoresist; development profile improved far UV photoresist

IT Positive photoresists (chemical amplified; chemical-amplified pos. photoresists containing alkali-solubility-improved sp. sulfonylazomethanes for far-UV photolithog.)

IT Photolithography (chemical-amplified pos. photoresists containing alkali-solubility-improved sp. sulfonylazomethanes for far-UV photolithog.)

IT 2628-17-3D, p-Hydroxystyrene, ethoxyethyl ether, 1,2-propanediol divinyl ether copolymer 2628-17-3D, p-Hydroxystyrene, ethoxyethyl ether,tert-butoxycarbonic ester, 1,2-propanediol divinyl ether copolymer 59269-51-1D, Polyhydroxystyrene, ethoxyethyl ether 155214-68-9D, ethoxyethyl ether 189257-17-8, Poly(hydroxystyrene) acetate 326925-68-2 326925-73-9 345580-95-2 346428-50-0 346428-52-2
RL: PEP (Physical, engineering or chemical process); TEM (Technical or engineered material use); PROC (Process); USES (Uses)
(chemical-amplified pos. photoresists containing alkali-solubility-improved sp. sulfonylazomethanes for far-UV photolithog.)

IT 104884-57-3P 327614-10-8P 334700-88-8P 334700-90-2P 334700-93-5P 334700-97-9P 346428-58-8P 346428-65-7P
RL: PNU (Preparation, unclassified); RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent)
(in preparation of sulfonyldiazomethane derivs. as photoacid generators of chemical-amplified photoresists)

IT 334700-94-6P 334700-95-7P 334700-99-1P 334701-00-7P
RL: PNU (Preparation, unclassified); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)
(in preparation of sulfonyldiazomethane derivs. as photoacid generators of chemical-amplified photoresists)

IT 70-11-1, α -Bromoacetophenone 75-09-2, Dichloromethane, reactions 75-36-5, Acetyl chloride 79-03-8, Propionyl chloride 98-88-4, Benzoyl chloride 637-89-8, 4-Hydroxythiophenol 941-55-9, p-Toluenesulfonylazide 3282-30-2, Pivaloyl chloride 68483-71-6, Chloromethylcyclohexyl sulfide
RL: RCT (Reactant); RACT (Reactant or reagent)
(in preparation of sulfonyldiazomethane derivs. as photoacid generators of chemical-amplified photoresists)

IT 39153-56-5, Bis(2,4-dimethylphenylsulfonyl)diazomethane 161453-44-7 161453-47-0 334701-01-8
RL: CAT (Catalyst use); USES (Uses)
(photoacid generators; chemical-amplified pos. photoresists containing alkali-solubility-improved sp. sulfonylazomethanes for far-UV photolithog.)

IT 334700-91-3P 334700-96-8P

RL: CAT (Catalyst use); PNU (Preparation, unclassified); PREP (Preparation); USES (Uses)
(photoacid generators; chemical-amplified pos. photoresists
containing alkali-solubility-improved sp. sulfonylazomethanes for
far-UV
photolithog.)

L8 ANSWER 28 OF 32 CA COPYRIGHT 2008 ACS on STN
AN 134:346380 CA

ED Entered STN: 31 May 2001

TI Preparation and characterization of photoreactive copolymers containing
curable pendants for positive photoresist

AU Liu, Jui-Hsiang; Lin, Seng-Hei; Shih, Jen-Chieh

CS Department of Chemical Engineering, National Cheng Kung University,
Tainan, 70101, Taiwan

SO Journal of Applied Polymer Science (2001), 80(3), 328-333
CODEN: JAPNAB; ISSN: 0021-8995

PB John Wiley & Sons, Inc.

DT Journal

LA English

CC 74-5 (Radiation Chemistry, Photochemistry, and Photographic and Other
Reproductive Processes)

Section cross-reference(s): 36

AB Tert-Bu methacrylate (TBMA) was polymerized with various monomers like Me
methacrylate (MMA), Bu acrylate (NBA), acrylic acid (AA), and
2-hydroxyethyl methacrylate (HEMA). From film phys. properties,
poly(TBMA-co-HEMA) and poly(TBMA-co-AA-co
-NBA), were selected as resin binders. To introduce double bonds into

the
polymer side chain, the polymers were further functionalized with
acryloyl

chloride and glycidyl methacrylate. Polymers synthesized were identified
using FTIR and NMR. The thermal decomposition temperature of
functionalized

poly(TBMA-co-HEMA) showed obvious difference before and after
crosslinking. Adding a small amount of EGDMA as the crosslinking agent
increased the degree of crosslinking and improved the phys. properties.
Functionalized poly(TBMA-co-HEMA) was used as a binder resin and
combined with a photoacid generator for pos. photoresists. From
exposure characteristics, the optimal lithog. condition was achieved when
exposed for 90 s, PEB at 100° for 2.5 min, and developed in 10%
Na2CO3 developer for 30 s. After completing the lithog. process, the
residual pattern of pos. photoresist was further treated at 140°
for 30 min to cure the pendant unsatd. groups. The resolution of the
pos.

photoresist was analyzed by an optical microscope and SEM technique.
ST photoreactive polymer curable pendant pos photoresist

IT Crosslinking

Crosslinking agents

(effect of, on thermal properties of photoreactive copolymers
containing
curable pendants for deep-UV pos.-tone photoresist)

IT IR spectra

NMR (nuclear magnetic resonance)

Thermal decomposition

FOR 10559534 by Cynthia Hamilton

UV and visible spectra
(of photoreactive copolymers containing curable pendants for deep-UV
pos.-tone photoresist)
IT Positive photoresists
Thermal properties
(thermal properties of photoreactive copolymers containing curable
pendants
for deep-UV pos.-tone photoresist)
IT 97-90-5, Ethylene glycol dimethacrylate
RL: RCT (Reactant); RACT (Reactant or reagent)
(effect of, on thermal properties of photoreactive copolymers
containing
curable pendants for deep-UV pos.-tone photoresist)
IT 88410-78-0P 129698-93-7P, tert-Butyl methacrylate-2-hydroxyethyl
methacrylate copolymer 338445-54-8P 338445-58-2P, tert-Butyl
methacrylate-2-hydroxyethyl methacrylate copolymer acrylate
RL: DEV (Device component use); PNU (Preparation, unclassified); PRP
(Properties); PREP (Preparation); USES (Uses)
(preparation and characterization of photoreactive copolymers
containing curable
pendants for deep-UV pos.-tone photoresist)
RE.CNT 11 THERE ARE 11 CITED REFERENCES AVAILABLE FOR THIS RECORD
RE
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(6) Hiraoka, H; J Electrochem Soc 1981, V128, P2645 CA
(7) Kosbar, L; ACS Symp Series 1993, V527, P245 CA
(8) Liu, J; J Appl Polym Sci 1998, V70, P2401 CA
(9) Liu, J; Polym Adv Technol 2000, V11, P228 CA
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L8 ANSWER 29 OF 32 CA COPYRIGHT 2008 ACS on STN

AN 134:139209 CA

ED Entered STN: 22 Feb 2001

TI Positive-working ultraviolet ray-sensitive resin composition and resist
pattern formation using same

IN Imai, Kenji; Hasegawa, Takeya

PA Kansai Paint Co., Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 18 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

IC ICM G03F007-039

ICS G03F007-004

CC 74-5 (Radiation Chemistry, Photochemistry, and Photographic and Other
Reproductive Processes)
Section cross-reference(s): 38

FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI JP 2001022067	A	20010126	JP 1999-190434	19990705

PATENT NO.	CLASS	PATENT FAMILY CLASSIFICATION CODES
JP 2001022067	ICM G03F007-039 ICS G03F007-004 IPC1 G03F0007-039 [ICM,7]; G03F0007-004 [ICS,7] IPCR G03F0007-039 [I,C*]; G03F0007-039 [I,A]; G03F0007-004 [I,C*]; G03F0007-004 [I,A]	19990705

AB The title resin composition contains (a) a (co)polymer of a α -hydroxy- α -methylstyrene compound and optional other copolymerizable unsatd. monomers, (b) a CO₂H-containing resin, (c) an ether bond-containing olefinic unsatd. compd, and (d) a photoacid generator. The title process comprises the steps of applying the composition on a substrate to form a UV-sensitive coating, exposing the coating to a UV ray laser beam directly or a UV ray through a mask film, and developing the coating.

The process may comprise the steps of: (i) coating the composition on a support to obtain a pos.-working UV ray-sensitive dry film having a UV ray-sensitive layer made of the composition; (ii) laminating the dry film on a substrate so that the layer is contacted with the substrate; (iii) peeling the support off, if necessary; (iv) exposing the layer to a UV ray laser beam directly or a UV ray through a mask film; and (v) developing the layer to form a resist pattern. In the latter process, when the support is not peeled off in step (iii), the layer may be developed after peeling the support off to form a pattern. The composition useful as a resist provides high resolution resist patterns.

ST UV resist hydroxymethyl styrene copolymer; carboxy resin pos photoresist; ether olefin photoresist

IT Photoresists
(UV, pos.; photoresist composition containing hydroxymethylstyrene copolymer, carboxy-containing resin, olefin with ether bond, and photoacid generator)

IT 85342-62-7
RL: TEM (Technical or engineered material use); USES (Uses)
(acid generator; photoresist composition containing hydroxymethylstyrene copolymer, carboxy-containing resin, olefin with ether bond, and photoacid generator)

IT 80-05-7DP, Bisphenol A, reaction products with chloroethyl vinyl ether 110-75-8DP, 2-Chloroethyl vinyl ether, reaction products with bisphenol A 764-48-7DP, 2-Hydroxyethyl vinyl ether, reaction products with isocyanate compound 25067-83-8P, Acrylic acid-butyl acrylate-2-hydroxyethyl acrylate-styrene copolymer 28805-80-3DP, Toluene diisocyanate trimethylolpropane adduct (3:1), reaction products wit hydroxyethyl vinyl ether 51032-74-7P, Poly(p-hydroxy- α -methylstyrene) 62385-58-4P

FOR 10559534 by Cynthia Hamilton

321336-83-8P
RL: PNU (Preparation, unclassified); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)
(photoresist composition containing hydroxymethylstyrene copolymer, carboxy-containing resin, olefin with ether bond, and photoacid generator)
IT 216573-41-0P
RL: PNU (Preparation, unclassified); RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent)
(preparation and hydrolysis of)
IT 216573-39-6P
RL: PNU (Preparation, unclassified); RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent)
(preparation and polymerization of)
IT 4286-23-1, p-Hydroxy- α -methylstyrene
RL: RCT (Reactant); RACT (Reactant or reagent)
(reaction with Et vinyl ether)
IT 109-92-2, Ethyl vinyl ether
RL: RCT (Reactant); RACT (Reactant or reagent)
(reaction with hydroxymethylstyrene)

L8 ANSWER 30 OF 32 CA COPYRIGHT 2008 ACS on STN
AN 134:123579 CA

ED Entered STN: 15 Feb 2001

TI Positive-working visible ray-sensitive resin composition and resist pattern formation using same

IN Imai, Kenji; Kogure, Hideo

PA Kansai Paint Co., Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 18 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

IC ICM G03F007-039

ICS G03F007-004; H05K003-06

CC 74-5 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)
Section cross-reference(s): 38

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	JP 2001022068	A	20010126	JP 1999-190435	19990705
PRAI	JP 1999-190435		19990705		

CLASS

	PATENT NO.	CLASS	PATENT FAMILY CLASSIFICATION CODES
	JP 2001022068	ICM	G03F007-039
		ICS	G03F007-004; H05K003-06
		IPCI	G03F0007-039 [ICM,7]; G03F0007-004 [ICS,7]; H05K0003-06 [ICS,7]
		IPCR	H05K0003-06 [I,C*]; H05K0003-06 [I,A]; G03F0007-004 [I,C*]; G03F0007-004 [I,A]; G03F0007-039 [I,C*]; G03F0007-039 [I,A]

AB The title resin composition contains (a) a (co)polymer of a

p-hydroxy- α -methylstyrene compound and optional other copolymerizable unsatd. monomers, (b) a CO₂H-containing resin, (c) an ether bond-containing olefinic unsatd. compd, (d) a photoacid generator, and (e) a photosensitizer. The title process comprises the steps of applying the composition on a substrate to form a visible ray-sensitive coating, exposing the coating to a visible ray laser beam directly or a visible ray through a mask film, and developing the coating to form a resist pattern. The process may comprise the steps of: (i) coating the composition on a support to obtain a pos.-working visible ray-sensitive dry film having a visible ray-sensitive layer made of the composition; (ii) laminating the dry film on a substrate so that the layer is contacted with the substrate; (iii) peeling the support off, if necessary; (iv) exposing the layer to a visible ray laser beam directly or a visible ray through a mask film; and (v) developing the layer to form a resist pattern. In the latter process, when the support is not peeled off in step (iii), the layer may be developed after peeling the support off to form a pattern. The composition useful as a resist provides high resolution resist patterns.

ST visible ray resist hydroxymethylstyrene copolymer; ether olefin photoresist; carboxy resin photoresist; photoacid generator sensitizer photoresist

IT Resist

(pos.-working; photoresist composition containing hydroxymethylstyrene copolymer, carboxy-containing resin, olefin with ether bond, and photoacid generator)

IT 85342-62-7

RL: TEM (Technical or engineered material use); USES (Uses) (acid generator; photoresist composition containing hydroxymethylstyrene copolymer, carboxy-containing resin, olefin with ether bond, and photoacid generator)

IT 80-05-7DP, Bisphenol A, reaction products with chloroethyl vinyl ether 110-75-8DP, 2-Chloroethyl vinyl ether, reaction products with bisphenol A 764-48-7DP, 2-Hydroxyethyl vinyl ether, reaction products with isocyanate compound 25067-83-8P, Acrylic acid-butyl acrylate-2-hydroxyethyl acrylate-styrene copolymer 28805-80-3DP, Tolylene diisocyanate trimethylolpropane adduct (3:1), reaction products wit hydroxethyl vinyl ether 51032-74-7P, Poly(p-hydroxy- α -methylstyrene) 62385-58-4P 321336-83-8P

RL: PNU (Preparation, unclassified); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (photoresist composition containing hydroxymethylstyrene copolymer, carboxy-containing resin, olefin with ether bond, and photoacid generator)

IT 216573-41-0P

RL: PNU (Preparation, unclassified); RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent) (preparation and hydrolysis of)

IT 216573-39-6P

FOR 10559534 by Cynthia Hamilton

RL: PNU (Preparation, unclassified); RCT (Reactant); PREP (Preparation);
RACT (Reactant or reagent)
(preparation and polymerization of)
IT 4286-23-1, p-Hydroxy- α -methylstyrene
RL: RCT (Reactant); RACT (Reactant or reagent)
(reaction with Et vinyl ether)
IT 109-92-2, Ethyl vinyl ether
RL: RCT (Reactant); RACT (Reactant or reagent)
(reaction with hydroxymethylstyrene)
IT 63226-13-1
RL: TEM (Technical or engineered material use); USES (Uses)
(sensitizer; photoresist composition containing hydroxymethylstyrene
copolymer,
carboxy-containing resin, olefin with ether bond, and photoacid
generator)

L8 ANSWER 31 OF 32 CA COPYRIGHT 2008 ACS on STN
AN 133:105930 CA

ED Entered STN: 11 Aug 2000

TI Preparations and compositions of lithographic resists containing
photosensitive polymers with cyclic ether backbone

IN Choi, Sang Joon; Chung, Dong Hang; Lee, Si Hyung

PA Samsung Electronics Co., Ltd., S. Korea

SO Jpn. Kokai Tokkyo Koho, 9 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

IC ICM C08F220-18

ICS C08F236-20; C08K005-36; C08L033-06; G03F007-039; H01L021-027

CC 38-3 (Plastics Fabrication and Uses)

Section cross-reference(s): 74, 76

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	JP 2000191732	A	20000711	JP 1999-364811	19991222
	JP 3660842	B2	20050615		
	KR 2000042004	A	20000715	KR 1998-58045	19981224
	TW 476022	B	20020211	TW 1999-88107907	19990515
	US 6287747	B1	20010911	US 1999-465926	19991217
PRAI	KR 1998-58045	A	19981224		

CLASS

	PATENT NO.	CLASS	PATENT FAMILY CLASSIFICATION CODES
JP	2000191732	ICM	C08F220-18
		ICS	C08F236-20; C08K005-36; C08L033-06; G03F007-039;
			H01L021-027
		IPCI	C08F0220-18 [ICM,7]; C08F0220-00 [ICM,7,C*]; C08F0236-20 [ICS,7]; C08F0236-00 [ICS,7,C*]; C08K0005-36 [ICS,7]; C08K0005-00 [ICS,7,C*]; C08L0033-06 [ICS,7]; C08L0033-00 [ICS,7,C*]; G03F0007-039 [ICS,7]; H01L0021-027 [ICS,7];
			H01L0021-02
			[ICS,7,C*]
		IPCR	H01L0021-02 [I,C*]; H01L0021-027 [I,A]; C08F0220-00 [I,C*]; C08F0220-18 [I,A]; C08F0222-00 [I,C*];

C08F0222-20 [I,A]; C08F0236-00 [I,C*]; C08F0236-20
 [I,A]; C08K0005-00 [I,C*]; C08K0005-36 [I,A];
 C08L0033-00 [I,C*]; C08L0033-06 [I,A]; G03F0007-004
 [N,C*]; G03F0007-004 [N,A]; G03F0007-039 [I,C*];
 G03F0007-039 [I,A]
 ECLA C08F0222/20; G03F007/039
 KR 2000042004 IPCI G03F0007-031 [ICM,7]
 IPCR H01L0021-02 [I,C*]; H01L0021-027 [I,A]; C08F0220-00
 [I,C*]; C08F0220-18 [I,A]; C08F0222-00 [I,C*];
 C08F0222-20 [I,A]; C08F0236-00 [I,C*]; C08F0236-20
 [I,A]; C08K0005-00 [I,C*]; C08K0005-36 [I,A];
 C08L0033-00 [I,C*]; C08L0033-06 [I,A]; G03F0007-004
 [N,C*]; G03F0007-004 [N,A]; G03F0007-039 [I,C*];
 G03F0007-039 [I,A]
 TW 476022 IPCI G03F0007-039 [ICM,7]
 IPCR H01L0021-02 [I,C*]; H01L0021-027 [I,A]; C08F0220-00
 [I,C*]; C08F0220-18 [I,A]; C08F0222-00 [I,C*];
 C08F0222-20 [I,A]; C08F0236-00 [I,C*]; C08F0236-20
 [I,A]; C08K0005-00 [I,C*]; C08K0005-36 [I,A];
 C08L0033-00 [I,C*]; C08L0033-06 [I,A]; G03F0007-004
 [N,C*]; G03F0007-004 [N,A]; G03F0007-039 [I,C*];
 G03F0007-039 [I,A]
 ECLA C08F0222/20; G03F007/039
 US 6287747 IPCI G03F0007-039 [ICM,7]; C08F0024-00 [ICS,7]; C08F0136-20
 [ICS,7]; C08F0136-00 [ICS,7,C*]; C08F0232-08 [ICS,7];
 C08F0232-00 [ICS,7,C*]
 IPCR H01L0021-02 [I,C*]; H01L0021-027 [I,A]; C08F0220-00
 [I,C*]; C08F0220-18 [I,A]; C08F0222-00 [I,C*];
 C08F0222-20 [I,A]; C08F0236-00 [I,C*]; C08F0236-20
 [I,A]; C08K0005-00 [I,C*]; C08K0005-36 [I,A];
 C08L0033-00 [I,C*]; C08L0033-06 [I,A]; G03F0007-004
 [N,C*]; G03F0007-004 [N,A]; G03F0007-039 [I,C*];
 G03F0007-039 [I,A]
 NCL 430/270.100; 526/266.000; 526/282.000; 526/309.000
 ECLA C08F0222/20; G03F007/039

AB The chemical amplifiable photoresists suitable for micro-patterning by dry etching with ArF excimer laser beams in the semiconductor device fabrication, comprise a (meth)acrylic acid ester-based copolymer having cyclic ether units of CH₂Z (Z = tetrahydropyran-3,5-diyl group bearing carboxylic acid esters on the 3- and 5-position, resp., provided that at least 1 of the esters is C₇-20 alicyclic hydrocarbyl type) in the backbone and photoacid generator (PAG). Thus, heating diadamantyl 2,2'-(oxydimethylene)diacrylate 18.2 with diethoxyethyl 2,2'-(oxydimethylene)diacrylate 10.0 and methacrylic acid 2.6 g in THF in the presence of AIBN at reflux for .apprx.24 h gave a copolymer having cyclic ether units, weight-average mol. weight of 15,400 and polydispersity of 2.4. Dissolving the copolymer 1.0, triphenylsulfonium triflate (PAG) 0.02 and triisobutylamine 0.002 in propylene glycol monomethyl ether acetate 7 g, and filtering gave a photoresist which was coated on a silicon wafer to 0.45 μ m thickness, pre-baked at 110° for 90 s, exposed with ArF excimer laser, post-exposure baked at 120° for 90 s and developed with a 2.38% tetramethylammonium hydroxide solution to give line-and-space

ST pattern of 0.30 μm under an exposure dose of .apprx.17 mJ/cm^2 .
lithog resist photosensitive polymer cyclic ether unit; semiconductor device manuf dry etching resist chem amplification; photoresist dry etching ArF excimer laser photocurable methacrylate copolymer; adamantly methacrylate ether dimer copolymer photoresist

IT Excimer lasers
(ArF; preps. and compns. of lithog. resists containing photosensitive polymers with cyclic ether backbone)

IT Ethers, uses
RL: IMF (Industrial manufacture); PEP (Physical, engineering or chemical process); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); PROC (Process); USES (Uses)
(cyclic, polymers; preps. and compns. of lithog. resists containing photosensitive polymers with cyclic ether backbone)

IT Sulfonium compounds
RL: CAT (Catalyst use); USES (Uses)
(photoacid generator; preps. and compns. of lithog. resists containing photosensitive polymers with cyclic ether backbone)

IT Etching
Photoresists
Resists
Semiconductor device fabrication
(preps. and compns. of lithog. resists containing photosensitive polymers with cyclic ether backbone)

IT Acids, uses
RL: CAT (Catalyst use); USES (Uses)
(strong; preps. and compns. of lithog. resists containing photosensitive polymers with cyclic ether backbone)

IT Amines, uses
RL: CAT (Catalyst use); USES (Uses)
(tertiary, crosslinking co-catalyst; preps. and compns. of lithog. resists containing photosensitive polymers with cyclic ether backbone)

IT 102-71-6, uses 111-42-2, uses 121-44-8, uses 1116-40-1,
Triisobutylamine 25549-16-0, Triisooctylamine
RL: CAT (Catalyst use); USES (Uses)
(crosslinking co-catalyst; preps. and compns. of lithog. resists containing photosensitive polymers with cyclic ether backbone)

IT 34684-40-7, N-Hydroxysuccinimide triflate 66003-76-7, Diphenyliodonium triflate 66003-78-9, Triphenylsulfonium triflate 144317-44-2,
Triphenylsulfonium nonaflate 157959-61-0 162845-55-8,
Triphenylsulfonium antimonate 168706-59-0 259229-69-1 259229-70-4D,
salts
RL: CAT (Catalyst use); USES (Uses)
(photoacid generator; preps. and compns. of lithog. resists containing photosensitive polymers with cyclic ether backbone)

IT 142-68-7DP, Tetrahydropyran, derivs., polymers
RL: IMF (Industrial manufacture); PEP (Physical, engineering or chemical process); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); PROC (Process); USES (Uses)
(preps. and compns. of lithog. resists containing photosensitive polymers with cyclic ether backbone)

IT 254109-23-4P, Diadamantyl 2,2'-(oxydimethylene)diacrylate-di-tert-butyl 2,2'-(oxydimethylene)diacrylate copolymer 282118-22-3P 282118-23-4P 282118-24-5P 282118-25-6P 282118-26-7P 282118-27-8P 282118-28-9P
RL: IMF (Industrial manufacture); POF (Polymer in formulation); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)
(prepns. and compns. of lithog. resists containing photosensitive polymers
with cyclic ether backbone)
IT 1663-39-4 5888-33-5, Isobornyl acrylate 30525-89-4,
Paraformaldehyde 52351-91-4, 1-Ethoxyethyl acrylate
121601-93-2, 1-Adamantyl acrylate
RL: RCT (Reactant); RACT (Reactant or reagent)
(reactant; prepns. and compns. of lithog. resists containing photosensitive polymers with cyclic ether backbone)

L8 ANSWER 32 OF 32 CA COPYRIGHT 2008 ACS on STN
AN 130:160483 CA
ED Entered STN: 13 Mar 1999
TI Dissolution characteristics of chemically amplified 193 nm resists
AU Itani, Toshiro; Yoshino, Hiroshi; Hashimoto, Shuichi; Yamana, Mitsuharu;
Miyasaka, Mami; Tanabe, Hiroyoshi
CS NEC Corporation, Sagamihara, Kanagawa, 229-1198, Japan
SO Journal of Vacuum Science & Technology, B: Microelectronics and Nanometer
Structures (1998), 16(6), 3726-3729
CODEN: JVTBD9; ISSN: 0734-211X
PB American Institute of Physics
DT Journal
LA English
CC 74-5 (Radiation Chemistry, Photochemistry, and Photographic and Other
Reproductive Processes)
AB The dissoln. kinetics of two types of chemical amplified pos. 193 nm
resists
were investigated; a terpolymer resist consisting of
poly(tricyclodecylacrylate-co-tetrahydrodipyranyl-methacrylate-
co-methacrylic acid) and triphenylsulfonium triflate as a
photoacid generator, and a copolymer resist consisting of
poly(carboxyteretracyclododecylmethacrylate-co-tetrahydro-
pyranyloxy-carbonyl-tetracyclododecylmethacrylate) and triphenylsulfonium
triflate as a photoacid generator. The dissoln. rate contrast
was higher and the slope of dissoln. rate curve was steeper for the
terpolymer resist than those for the copolymer resist. However, the
Arrhenius plots of the dissoln. rates were straight lines for both
resists
irresp. of the exposure doses. This indicates that only one mechanism
dets. the dissoln. of both resists, and it is believed that the dominant
rate-determining step in both resists is the tetramethylammoniumhydroxide
penetration into the resist films. The resolution capability of the
terpolymer resist was very high, 0.14 μ m lines and spaces pattern. The
resolution capability of the copolymer resist was moderate, 0.16 μ m
lines
and spaces in spite of its lower dissoln. contrast and smaller slope
value. In addition, the dry-etch resistance of both resists was close to
that of a conventional polyhydroxystyrene base KrF resist. These results

FOR 10559534 by Cynthia Hamilton

indicate that both the terpolymer and the copolymer resists are candidates for practical use.

ST chem amplified pos photoresist dissoln characteristic

IT Photoresists (chemical amplified, pos., deep-UV; dissoln. characteristics of)

IT 66003-78-9, Triphenylsulfonium triflate 170969-47-8, (Tricyclodecyl acrylate-tetrahydrodpyranyl methacrylate-methacrylic acid copolymer 195458-42-5

RL: TEM (Technical or engineered material use); USES (Uses) (dissoln. characteristics of chemical amplified, pos., deep-UV photoresists containing)

RE.CNT 14 THERE ARE 14 CITED REFERENCES AVAILABLE FOR THIS RECORD

RE

(1) Allen, R; J Photopolym Sci Technol 1995, V8, P623 CA
(2) Allen, R; Proc SPIE 1995, V2438, P474 CA
(3) Houlihan, F; Proc SPIE 1997, V3049, P84 CA
(4) Itani, T; J Photopolym Sci Technol 1997, V10, P409 CA
(5) Itani, T; J Vac Sci Technol B 1997, V15, P2541 CA
(6) Itani, T; Jpn J Appl Phys Part 1 1994, V33, P7005 CA
(7) Iwasa, S; J Photopolym Sci Technol 1996, V9, P447 CA
(8) Maeda, K; Proc SPIE 1996, V2724, P377 CA
(9) Maeda, K; Proc SPIE 1997, V3049, P55 CA
(10) Nakano, K; Proc SPIE 1995, V2438, P433 CA
(11) Nakano, K; Proc SPIE 1995, V2438, P433 CA
(12) Shida, N; J Photopolym Sci Technol 1996, V9, P457 CA
(13) Takechi, S; J Photopolym Sci Technol 1996, V9, P475 CA
(14) Wallow, T; Proc SPIE 1996, V2724, P355

=> s carbon dioxide and acrylate and photoacid

1307200 CARBON
507921 DIOXIDE
241516 CARBON DIOXIDE
(CARBON(W)DIOXIDE)

194728 ACRYLATE
4104 PHOTOACID

L9 1 CARBON DIOXIDE AND ACRYLATE AND PHOTOACID

=> d all

L9 ANSWER 1 OF 1 CA COPYRIGHT 2008 ACS on STN

AN 140:10624 CA

ED Entered STN: 25 Dec 2003

TI Method for forming chemically amplified resist patterns with excellent cross-sectional shapes

IN Endo, Masataka; Sasago, Masaru

PA Matsushita Electric Industrial Co., Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 9 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

IC ICM H01L021-027

ICS G03F007-039; G03F007-26; G03F007-38

CC 74-5 (Radiation Chemistry, Photochemistry, and Photographic and Other

Reproductive Processes)					
FAN.CNT 1	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	JP 2003338452	A	20031128	JP 2002-147330	20020522
	CN 1461040	A	20031210	CN 2003-131009	20030514
	US 2003224589	A1	20031204	US 2003-438859	20030516
	US 6902999	B2	20050607		
PRAI	JP 2002-147330	A	20020522		
CLASS					
PATENT NO.	CLASS	PATENT FAMILY	CLASSIFICATION CODES		
JP 2003338452	ICM	H01L021-027			
	ICS	G03F007-039; G03F007-26; G03F007-38			
	IPCI	H01L021-027 [ICM,7]; H01L021-02 [ICM,7,C*]; G03F007-039 [ICS,7]; G03F007-26 [ICS,7]; G03F007-38 [ICS,7]			
	IPCR	G03F0007-039 [I,C*]; G03F0007-039 [I,A]; G03F0007-004 [N,C*]; G03F0007-004 [N,A]; G03F0007-16 [I,C*]; G03F0007-16 [I,A]; G03F0007-26 [I,C*]; G03F0007-26 [I,A]; G03F0007-38 [I,C*]; G03F0007-38 [I,A]; H01L0021-02 [I,C*]; H01L0021-027 [I,A]; H01L0021-3105 [I,A]; H01L0021-311 [I,A]			
CN 1461040	IPCI	H01L0021-027 [ICM,7]; H01L0021-02 [ICM,7,C*]; G03F0007-00 [ICS,7]			
	IPCR	G03F0007-039 [I,C*]; G03F0007-039 [I,A]; G03F0007-004 [N,C*]; G03F0007-004 [N,A]; G03F0007-16 [I,C*]; G03F0007-16 [I,A]; G03F0007-26 [I,C*]; G03F0007-26 [I,A]; G03F0007-38 [I,C*]; G03F0007-38 [I,A]; H01L0021-02 [I,C*]; H01L0021-027 [I,A]; H01L0021-3105 [I,A]; H01L0021-311 [I,A]			
	ECLA	H01L021/027B; H01L021/3105B2B; H01L021/3105P; H01L021/311D			
US 2003224589	IPCI	H01L0021-22 [ICM,7]; H01L0021-02 [ICM,7,C*]			
	IPCR	G03F0007-039 [I,C*]; G03F0007-039 [I,A]; G03F0007-004 [N,C*]; G03F0007-004 [N,A]; G03F0007-16 [I,C*]; G03F0007-16 [I,A]; G03F0007-26 [I,C*]; G03F0007-26 [I,A]; G03F0007-38 [I,C*]; G03F0007-38 [I,A]; H01L0021-02 [I,C*]; H01L0021-027 [I,A]; H01L0021-3105 [I,A]; H01L0021-311 [I,A]			
	NCL	438/552.000; 257/E21.024; 257/E21.242; 257/E21.245; 257/E21.257			
	ECLA	H01L021/027B; H01L021/3105B2B; H01L021/3105P; H01L021/311D; G03F007/16			

AB The method contains treating porous films or organic material-containing films in

supercrit. fluids (CO₂, preferably), forming chemical amplified resist layers

on the smoothed films, patterning the layers, and developing them, wherein

the resists may contain base polymers bearing tert-Bu, tert-butyloxycarbonyl, or adamantly groups and imide-type acid photogenerators.

ST chem amplification resist pattern sharp edge; photoresist patterning surface smoothing porous film; supercrit carbon dioxide

FOR 10559534 by Cynthia Hamilton

IT treatment film photoresist
IT Photoresists
with (chemical amplified; formation of chemical amplified resist patterns
good cross-sectional shapes on films surface-treated with supercrit.
fluids)
IT Supercritical fluids
cross-sectional (formation of chemical amplified resist patterns with good
shapes on films surface-treated with supercrit. fluids)
IT 2420-27-1, 2,4,6-Tris(methoxymethyl)aminotriazine
RL: TEM (Technical or engineered material use); USES (Uses)
(crosslinking agent for photoresist; formation of chemical amplified
resist patterns with good cross-sectional shapes on films
surface-treated with supercrit. fluids)
IT 203945-07-7, Silk
RL: PEP (Physical, engineering or chemical process); PYP (Physical
process); TEM (Technical or engineered material use); PROC (Process);
USES
(Uses)
patterns with (organic polymer film; formation of chemical amplified resist
good cross-sectional shapes on films surface-treated with supercrit.
fluids)
IT 41580-58-9, Phthalimido triflate 66003-78-9, Triphenylsulfonium
triflate
RL: TEM (Technical or engineered material use); USES (Uses)
(photoacid generator; formation of chemical amplified resist
patterns with good cross-sectional shapes on films surface-treated
with supercrit. fluids)
IT 59269-51-1, Poly(vinylphenol) 328249-39-4 627077-03-6,
 γ -Butyrolactone methacrylate-methoxymethyl acrylate
copolymer
RL: TEM (Technical or engineered material use); USES (Uses)
(photoresist; formation of chemical amplified resist patterns with
good cross-sectional shapes on films surface-treated with supercrit.
fluids)
IT 124-38-9, Carbon dioxide, uses
RL: NUU (Other use, unclassified); USES (Uses)
(supercrit.; formation of chemical amplified resist patterns with good
cross-sectional shapes on films surface-treated with supercrit.
fluids)

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(FILE 'HOME' ENTERED AT 20:31:49 ON 14 MAR 2008)

FILE 'REGISTRY' ENTERED AT 20:32:00 ON 14 MAR 2008

L1 0 S DIMETHYLHEXANE AND DIMETHACRYLATE
L2 2 S DIMETHYLHEXANE AND DIACRYLATE
L3 1 S 188837-15-2
L4 122 S DIMETHYL AND HEXANEDIOL AND DIACRYLATE

FOR 10559534 by Cynthia Hamilton

L5 122 S DIMETHYL AND HEXANEDIOL AND DIACRYLATE
L6 2 S C14H22O4/MF AND DIACRYLATE

FILE 'CA' ENTERED AT 20:35:52 ON 14 MAR 2008

L7 14 S L3
L8 32 S CO AND ACRYLATE AND PHOTOACID
L9 1 S CARBON DIOXIDE AND ACRYLATE AND PHOTOACID

=> log y		SINCE FILE	TOTAL
COST IN U.S. DOLLARS		ENTRY	SESSION
FULL ESTIMATED COST		163.70	240.47
DISCOUNT AMOUNTS (FOR QUALIFYING ACCOUNTS)		SINCE FILE	TOTAL
CA SUBSCRIBER PRICE		ENTRY	SESSION
		-35.25	-35.25

STN INTERNATIONAL LOGOFF AT 20:38:36 ON 14 MAR 2008